CODENEM

Formic Acid-based <u>Hydrogen Energy Production and</u> <u>D</u>istribution <u>System</u>

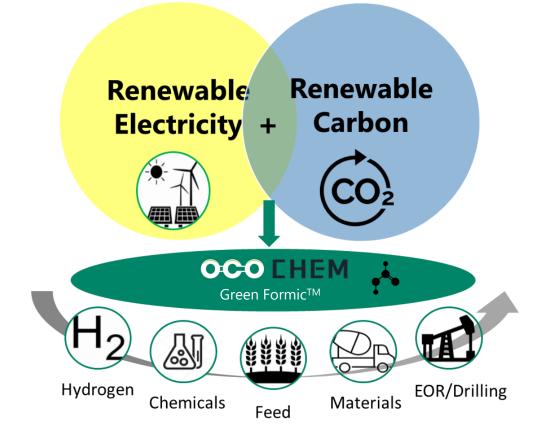
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AMR Project ID # ST245





OVERVIEW: End-to-End Demonstration of Formic Acid as Liquid Hydrogen Carrier

- **Primary goal** develop and demonstrate at an industrially relevant size an end-to-end cost-effective high performance novel clean liquid hydrogen carrier (LHC) production, distribution and dispensing supply chain using formic acid as the liquid hydrogen carrier.
- OCOchem is developing an industrial-scale height (1.15m) CO₂ electrolyzer device that converts captured CO₂, water and clean electricity into a liquid hydrogen carrier, formic acid. It is the world's largest CO₂-Formate electrolyzer using cutting-edge falling film gas diffusion electrode technology operating at 80% Faradaic Efficiency and 95% formic selectivity with a reactor output concentration of 40 wt.%. PNNL has developed a formic reforming process that de-hydrogenates formic acid and separates H₂ from CO₂ to liberate fuel-cell grade hydrogen. Together the technologies provide a safer and lower cost to make, store, move and use clean hydrogen using an energy-dense carrier.

Key Milestones & Deliverables							
Year 1• Scale-up Electrolyzer cell 10x to industrial-scale height and width (1.15m x 1.30m) • Scale-up Formic reformer 100x to 0.01 kg_H2/hr. Achieve 99.7% H2 PurityYear 2• Scale-out Electrolyzer 4x more to a 4-cell stack producing 150 kg/day. 40x scale-up+out total							
	Scale-up Formic reformer 100x to 0.01 kg_H ₂ /hr. Achieve 99.7% H ₂ Purity						
Year 2	 Year 2 • Scale-out Electrolyzer 4x more to a 4-cell stack producing 150 kg/day. 40x scale-up+out total 						
	 Scale-up Formic reformer 100x more to 1.0 kg_H₂/hr. Achieve 99.997% H₂ Purity 						



Benchmarks

- BP1/Year 1 Achieve two of the performance targets precisely at stated target values (Hydrogen production rate of 0.01 kg/hr, formate electrolyzer faradaic efficiency of 80%) and the rest of the performance targets within a +/- 20% variance
- BP2/Year 2 Achieve the performance targets within a +/- 10% variance



	Performance Targets by Budget Period								
	Budget Period:	Baseline	BP1-I	Pilot	BP2-D	Demo			
				20%		10%			
Crite	eria	Baseline	Target	Variance	Target	Variance			
Evolution	Au/C Catalyst Loading (g:g ratio)	0.250	0.025	0.031	0.025	0.028			
olu	CO by-product generated, ppm	< 1	<1	<1	< 1	<1			
	Separated Hydrogen Purity, % wt.	N/A	98.7%	79.0%	99.997%	90.0%			
H2	Hydrogen Production Rate, kg/hr.	0.0001	0.01	0.01	1.00	1.00			
	Operation Duration, hrs	1	24	19	100	90			
c	Elecrolyzer Scale Up	1 Tall Cell	1 Ful	l Cell	4 Full	Cells			
Production	Electrolyzer Scale, Active surface Area, m ²	0.162	1.62		6.48				
npc	Avg. Current Density, mA/cm ²	125	125	100	125	112.5			
Prc	Avg. Cell Voltage, V	4.5	4.5	5.4	4.5	5.0			
nic	Avg. Faradaic Efficiency	80%	80%	80%	80%	80%			
Formic	Formic Acid Production Rate, kg/day	3.75	37.5	30	150	135			
ш	Operation Duration, hrs	8	100	80	240	216			
	Note: Bolded performance targets are prec	ise targets, w	/ith no varian(e allowance i	in BP1 or BP2				

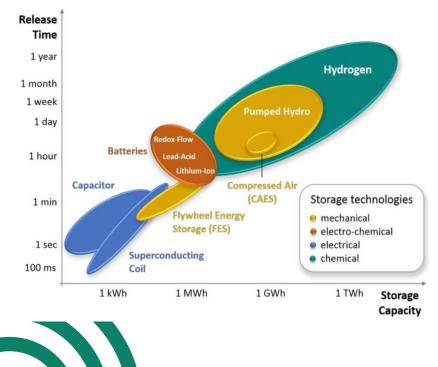


Potential Impact: The Hydrogen Economy is Arriving

Renewables are expected to provide >50% of global energy production by 2050. Batteries will only provide 2% of energy storage and is not adequate for most mobile, off-grid, high-duty, large volume and longduration energy storage applications. Therefore, the use of hydrogen and hydrogen carriers is inevitable.

Energy Storage Technologies

Only Chemically Bonded Hydrogen and Hydrogen Carriers Can Store TWh-scale Energy World Needs



Hydrogen Market Size

80% of the \$10-12 Trillion in Hydrogen market growth over the next 25 years will be delivered Hydrogen. Today, only 2% of Hydrogen is delivered.

\$10-12 Trillion (2050)Source: Bank of America. **Goldman Sachs** \$0.2 Trillion (2026)

On-Site Delivered

Market Trends

Hydrogen is being adopted as a fuel for many systems.

Hydrogen Vehicles



\$3/kg













Potential Impact: For H2 to be delivered affordably, Distribution Cost is the Key

Production Cost

- Green Hydrogen (\$6/kg)
- Gray Hydrogen (\$1.50/kg)
- This cost is falling due to larger more efficient electrolyzers, lower electricity cost and production tax credits

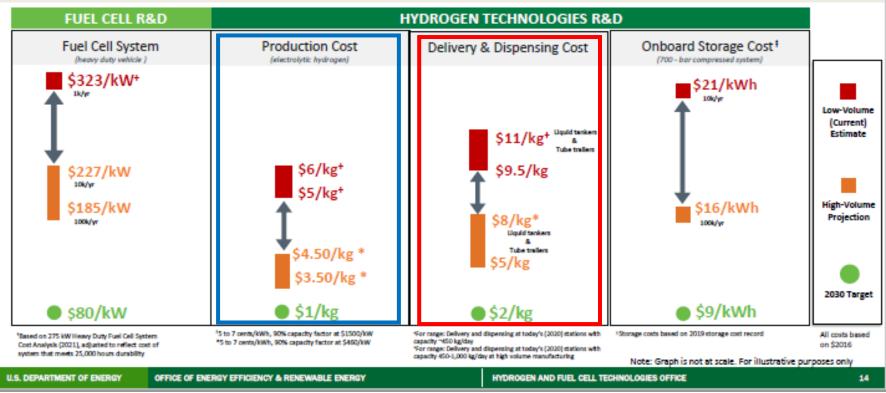
Distribution Cost

2015

- Gray Hydrogen (\$11/kg)
- Green Hydrogen (\$11/kg)
- This cost has not fallen since

Technology Targets Guide HFTO R&D Activities

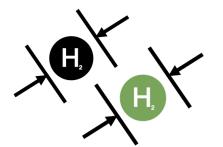
Key Goals: Reduce the cost of fuel cells and hydrogen production, delivery, storage, and meet performance and durability requirements – guided by applications specific targets





Potential Impact: Molecular H₂ has Several Challenges

The adoption of the **hydrogen economy** is inhibited by current hydrogen storage, distribution and compression problems. These **problems can be overcome** via the use of the liquid hydrogen carrier, formic acid, produced by **OCOchem's patented electrolyzer technology.**



Compressed

Hydrogen

(250 Bar)

Conventional (Green or Gray)

- × Gas
- × Less Energy Dense $(17.6 \text{ g}_H_2/\text{L})$
- × High Pressure (250 bar)
- × Explosive
- × Difficult to See, Smell or Detect
- × Specialized New H₂ Infrastructure
- × Expensive to Distribute (>\$10/kg)



OCO EHEM

Formic Acid

- Liquid
- Energy Dense (53.4 g_H₂/L)
- Standard Pressure and Temperature
- ✓ Low Flammability
- Easy to See, Smell and Detect
- ✓ Generic Existing Liquid Infrastructure
- ✓ Affordable to Distribute (<\$0.50/kg)</p>

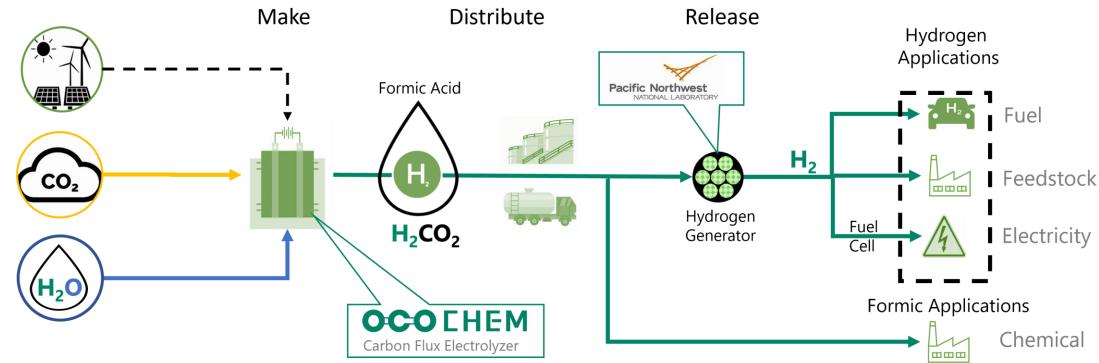
By making and storing green hydrogen <u>directly</u> into a liquid hydrogen carrier form, one can avoid the many cost and safety issues of molecular hydrogen





Potential Impact: Using CO₂ to Carry H₂ in a Liquid Form

OCOchem's process directly converts CO₂, water and clean electricity into a liquid hydrogen carrier, which can be stored and moved like a conventional liquid, and then reformed to generate H₂ on-demand



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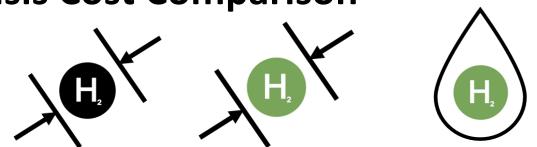
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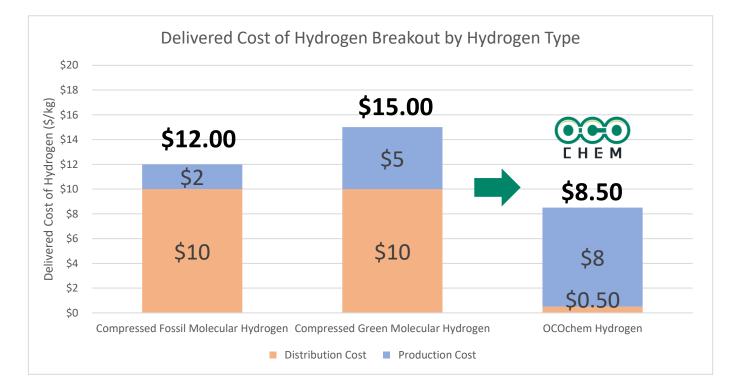
Potential Impact: Hydrogen Basis Cost Comparison

OCOchem plans to deliver Hydrogen at a **30-45% lower cost** than green or gray Molecular Hydrogen (H_2) because liquids are 20x less expensive to store and move than compressed gases.

Assumptions:

- 1. Assumes \$40/MWh green electricity cost, \$30/ton CO2 cost (Source: DOE, OCOchem extrapolated performance results)
- Distribution cost includes compression, storage, transport and dispensing costs typical of one-way transport distance of 100km (Source: DOE)







Potential Impact: Hydrogen Transport Comparison

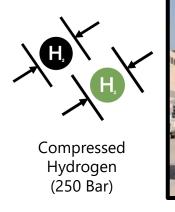
5.8x as much Hydrogen can be moved per truckload using formic as liquid hydrogen carrier than 250bar compressed molecular Hydrogen gas





Conventional Liquid Tanker Truck

Pressure:	1 Bar (unpressurized)
Storage Capacity:	60,000L
Hydrogen Stored:	2,200 kg of H2
Explosion Hazard:	No
Cost of Tank Trailer (New):	\$19,900-34,500





Steel Tube Trailer Truck	
Pressure:	250 Bar*
Storage Capacity:	18,600L
Hydrogen Stored:	380 kg of H2
Explosion Hazard:	Yes
Cost of Tube Trailer (New):	\$75,000-\$90,000

* Tube trailers are currently limited to pressures of 250 bar by U.S. Department of Transportation (DOT) regulations, but exemptions have been granted to enable operation at higher pressures (e.g., 500 bar or higher). Source: Department of Energy



Hydrogen Storage Comparison

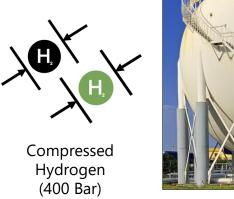
The cost of storing 10,000kg of Hydrogen is 50x more expensive with compressed molecular H2 than using formic acid







Above-ground Liquid Fuel Ta	nk
Pressure:	
Storage Capacity:	
Dimensions:	
Hydrogen Stored:	
Explosion Hazard:	
Cost of Tank Installed	
Safety Perimeter needed	`
Total Land footprint Needed	1600 SQFT (1



1 Bar* 1200 Barrels (185m³) 10m diameter x 2.5m height **10,000 kg of H2** No \$150,000-\$200,000 0' (12m x12m)



ASME Compressed Hydrogen Storage Spheres

Pressure:	400 Bar*
Storage Capacity:	1900 Barrels
Diameter:	18m
Hydrogen Stored:	10,000 kg of H2
Explosion Hazard:	Yes
Cost of Tank Installed:	\$5,000,000 (\$500/kg)
Safety Perimeter needed	` 165′ (50m)
Total Land footprint needed	40,000 SQFT (60x60m)

*400 Bar enables refueling of high duty hydrogen FCEVs at 350 bar (est.



Energy Density Comparison

Equivalent Energy Storage Content (1.8 MWh) Total System Weight





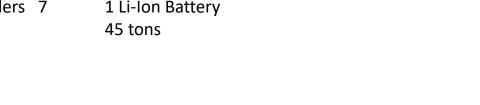
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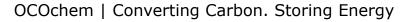
1000L Formic Acid 1.2 tons



128 Compressed (200bar) H2 Cylinders 7 tons

1 Li-Ion Battery







Approach: Safety Planning.

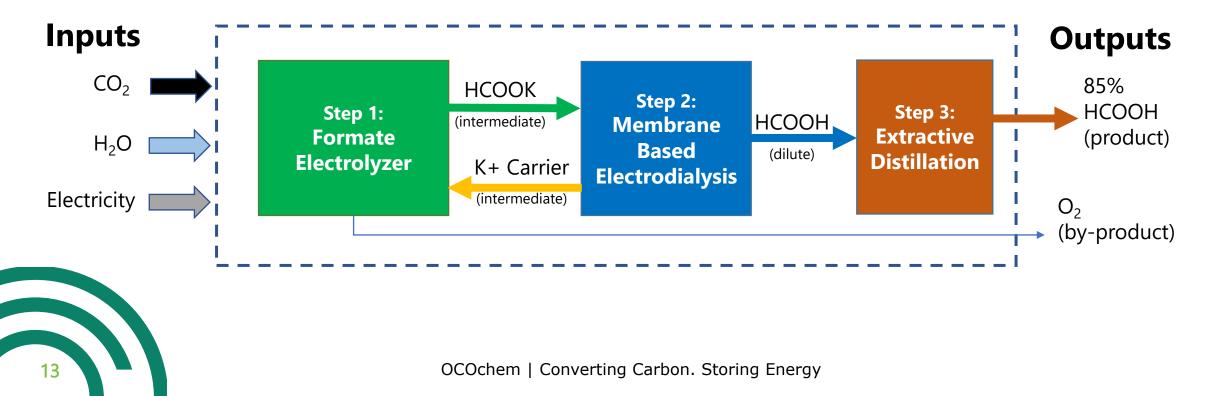
- As part of the Q4 2023 report deliverables to the DOE, PNNL has provided the DOE with its hydrogen safety plan and is currently working with the DOE on making modifications as requested by the DOE.
- OCOchem and PNNL conform with lab safety measures for all operations of this project. The personnel involved are adequately trained to handle the experimental requirements.





Approach: OCOchem Formic Synthesis Process

- Step 1: CO₂, water, K+ and electricity is converted into potassium formate (HCOOK)
- Step 2: HCOOK is acidified via electrodialysis to dilute formic acid (HCOOH), and K+ (with proprietary anion) is recycled to Step 1 to "carry" formate
- Step 3: dilute formic acid is concentrated via extractive distillation to product purity (85%)

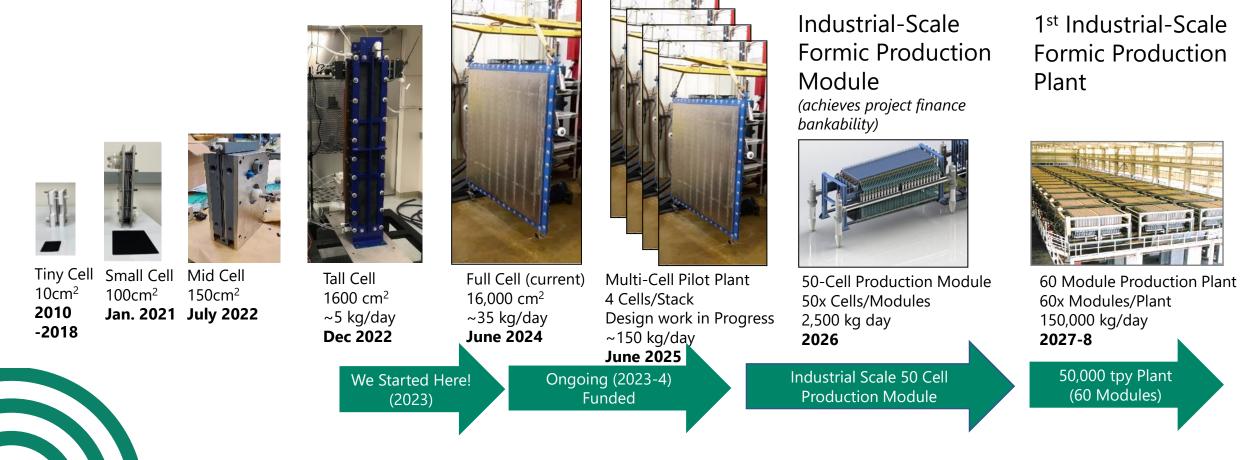




Approach: Scale-Up Process

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OCOchem is iteratively scaling from small to full, industry sized, multi-cell systems





Approach: OCOchem Scale-up Strategy

- 1) Adapting to existing large scale electrochemical industry for leveraging hardware & expertise
 - a) Chlor Alkali (C/A) industry uses ~ 3% of the worlds electricity for Cl2 & Caustic Soda, bulk chemicals with widespread uses
 - b) C/A has progressed over the last 50-70 years, has multiple (>5) reactor cell manufacturers worldwide
 - c) Rather than devising new cells, which will lead to significant supply chain, tooling/design issues, OCOchem focuses on adapting Chlor Alkali (C/A) cells to its technology, requiring only 15-20% modification of cell parts for its 1st generation cells
 - d) OCOchem is modifying and using cells built for C/A industry, adapting it for the key differences in process chemistry (more benign), operating conditions (lower T/P), reactor configuration and products (O2 and formate vs. Cl2 and caustic).
 - e) C/A cells, including ion exchange membranes typically demonstrate for 5-7 years of lifetime
 - f) 100s 1000s of cells can be sourced from one of the large-scale producers
- 2) OCOchem is currently modifying a chlor alkali full industry sized cell (15,000cm2 size), testing to commence soon
- 3) Optimization of the full cell, based on testing and learning from tall cell to be carried out through 2024
- A 4-cell pilot plant, with 4 modified C/A full industry sized cells, along with electrodialysis and distillation systems for 100-200 kg/day in-house formic acid production will be completed at OCOchem facility by Q1 2025, and optimized for target performance through Q3 2025.



Accomplishments: Scale-Up Progress





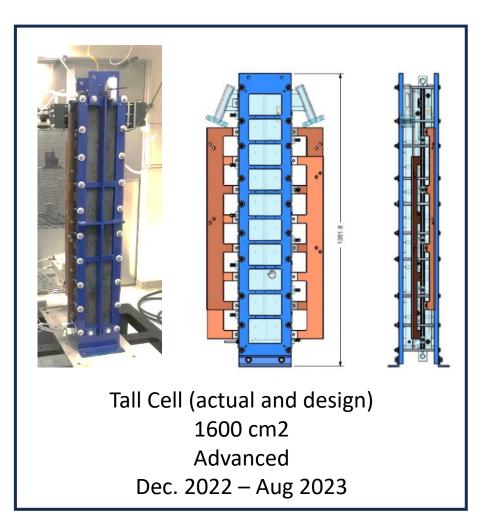
10cm2 2020

Small Cell 100cm2 Jan, 2021



July, 2022

NS-01 (Mid-Cell) 150cm2 Advanced



- OCOchem has methodically scaled the formate technology 3 times by a total of 160x in 2.5 years including
 - Catalyst and electrode (GDE)
 - Ion exchange membranes
 - Process chemistry •
 - Hydrodynamics impact
- Optimization to meet target performance achieved within 10%
- Currently scaling up to industrial sized full cells



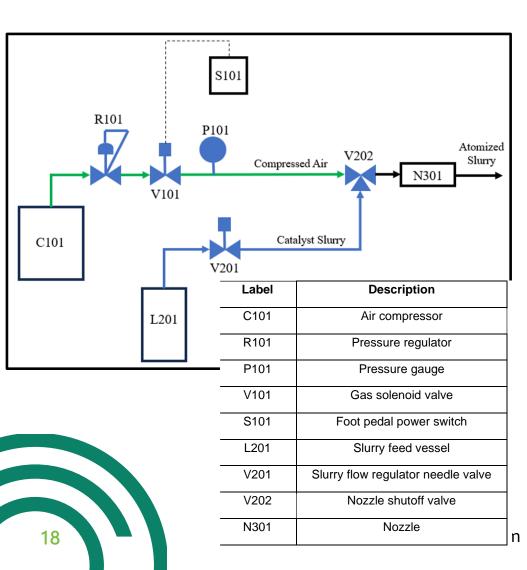
Accomplishments & Progress

• As OCOchem started working on this project in October 2023, we have no improvements based on last year's comments. We look forward to your feedback this year.



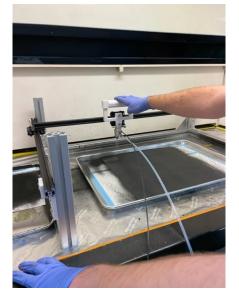


Accomplishments: In-house GDE semi-automated spray system



Multiple 40cm x 40cm panels spray capacity









Accomplishments: Single Full size cell Pilot Plant Progress

1.15m x 1.30m Electrolyzer Cell Housing Plates



Cell Mounting Carriage

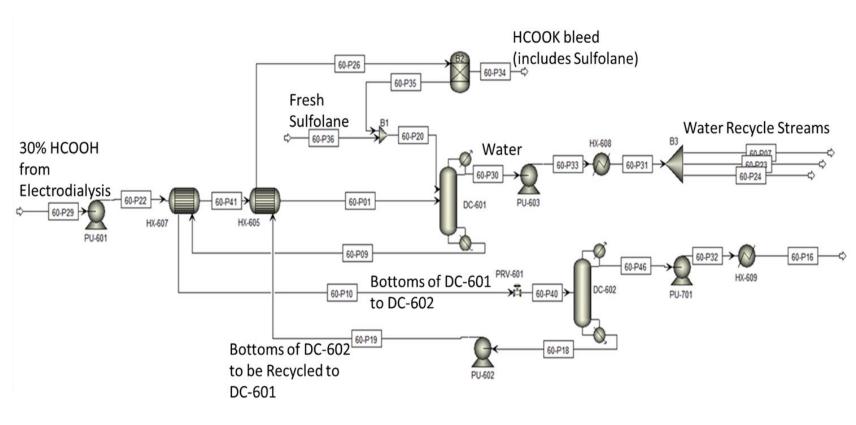


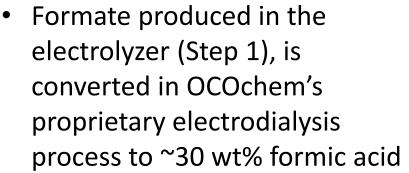
- OCOchem has iteratively designed and constructed proprietary custom parts (GDE cathode, GDE holder, center compartment, mesh structure) to be assembled into the full cell.
- The remainder cell components, including the carriage have been procured.
- Currently modifying high-bay facilities at OCOchem's Richland WA labs for housing and operating the single full cell pilot system

Energy

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Accomplishments: Electrodialysis & Distillation Process Development





- Sulfolane based extractive distillation, selected and optimized for product grade formic acid (85wt%) as the distillate of the recovery column with negligible sulfolane impurity.
- Water in the distillate of the extraction column is pure and reused/recycled



Proposed Future Work

🔆 We are here

- Before this project finishes, we anticipate:
 - Completing the design and construction of and operating a single full-cell scale formic acid demonstration plant
 - Completing the design and construction of and operating a 4 fullcell scale formic acid plant.

									E	Budg	et P	eric	od 1		Budget Period 2												
lask #	Task Name																										
		Length:	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	2	
0	Project Management	24																									
1	Development of a more frugal Au/C catalyst.	9						1.1																			
2	Separation of H2 from the H2/CO2 gas stream	12			2.1																						
3	Design, construction, and operation of a 0.01 kg H ₂ /h formic acid reformer	12									3.1																
4	Design and support fabrication/operation of demonstration-scale FA reformer	4												4.1			4.2										
5	Design and construction of a full cell formate electrolyzer	9									5.1																
6	Design and construction of formic acidification and balance of plant system	10				6.1																					
7	Integration and operation of a full cell scale formic production	3												*, ** (BP1 GNG, 7.1)													
8	Engineer, fabricate and operate of a 1.0 kg H2/h formic acid reformer	10																					8.1				
9	Design, construct and operate a 4 full cell electrollyzer stack and balance of plant	10															9.1			9.2							
10	Demonstration of LHC-based Hydrogen production, distribution and dispensing.	2																								10	
Key:																											
	OCOchem Tasks		·Go/	No-0	So De	cisio	on Po	int, S	MAR	TM	lest	one															
	PNNL Tasks		**. 10	dud	es Di	EI SM	ART	goal t	toen	nploy	ted	hnic	/No-Go Decision Point, SMART Milestones ncludes DEI SMART goal to employ technican from disdvantaged youth group per DEI Plan document.														



Collaboration & Coordination

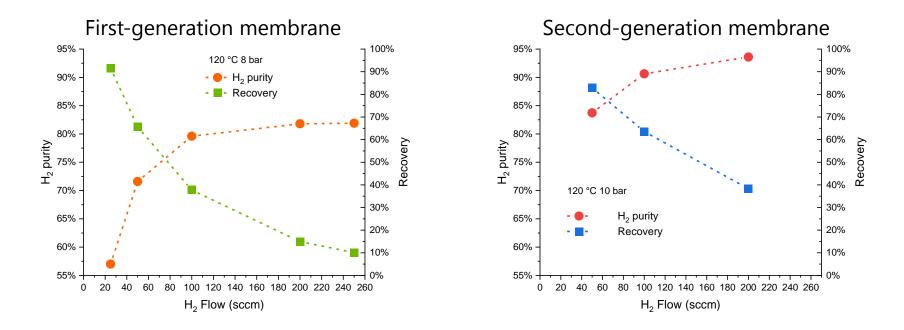
- OCOchem and PNNL are collaborating on this project
- Regular Monthly meetings to review and communicate progress and discuss next steps
- Quarterly reports are written in collaboration with team members providing appropriate input based on their involvement in the tasks.





Accomplishment: Purification to > 90% H₂

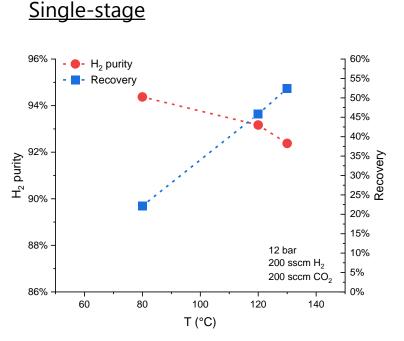
- Purification of H_2 is accomplished with a commercial membrane that permeates H_2 faster than CO_2
- Performance is characterized by <u>purity</u> (the H₂ composition exiting the membrane) and <u>recovery</u> (how much of the H₂ fed to the separator ends up in the higher purity exit)

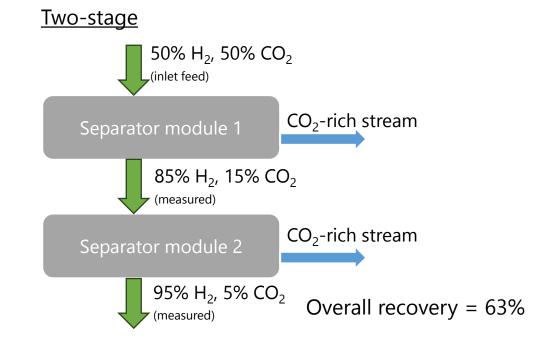


Improvements to the membrane have provided a significant boost in recovery and purity. Increased recovery at high purity is a remaining challenge.

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Accomplishment: 2-stage separator to increase recovery





Recovery improves at higher temperatures, without a significant compromise in purity

Enrichment to 95% purity has been demonstrated. Improved temperature management will increase this to >99% H_2 .

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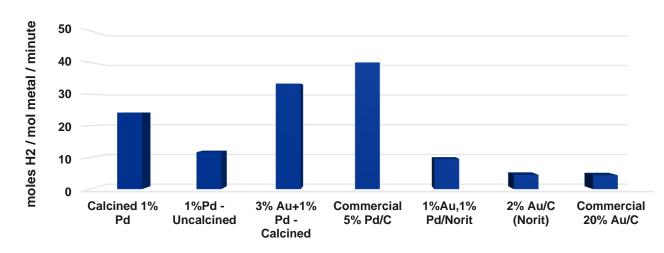
95% H₂ purity has been achieved by measurement of sequential separation stages Pre-heating the gas will increase recovery since it has a strong temperature dependence.



Accomplishment: Hydrogen release from formic acid

Formic acid dehydrogenation has been studied using carbon-supported metal catalysts at 160 °C

- Pd on carbon and Au+Pd on carbon show highest turnover frequency
 - Exceeds project target of 10 mol H₂ (mol metal)⁻¹ ¹min⁻¹
 - Pd/C commercially available
 - 9 times higher than baseline study of 4.4 mol H₂ (mol metal)⁻¹min⁻¹
- Pd/C produces ~1% CO side product
- No CO observed with Au+Pd/C catalyst



Comparison of Averaged Catalyst Activity

Catalyst

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A more cost-effective use of Pd/C catalysts for dehydrogenation of formic acid has been demonstrated compared to baseline performance. Small quantities of Au are effective at suppressing CO from the dehydration reaction.



Summary

- OCOchem is generating formic acid as a liquid hydrogen carrier (LHC) for an end-to-end, safe and cost-effective means for production, distribution and dispensing of hydrogen
- As part of this DOE project, OCOchem will be developing and demonstrating electrochemical formic acid production via CO2 and water at pilot scales (with a 1.5 m2 sized cell in year 1 and with a 4-cell sized pilot plant targeting 150 kg/day production in year 2)
- PNNL in collaboration with OCOchem, is developing the reformation technology to demonstrate effective conversion of formic acid to a H2, CO2 gas mixture, and then separation of pure H2 from the mixture, thus completing the value chain

