



Hydrogen from
Next-generation
Electrolyzers of Water

U.S. DEPARTMENT OF ENERGY

H2NEW: Hydrogen (H2) from Next-generation Electrolyzers of Water Overview

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Deputy Director: Richard Boardman, Idaho National Laboratory (INL)

Date: 6/7/2022

DOE Hydrogen Program

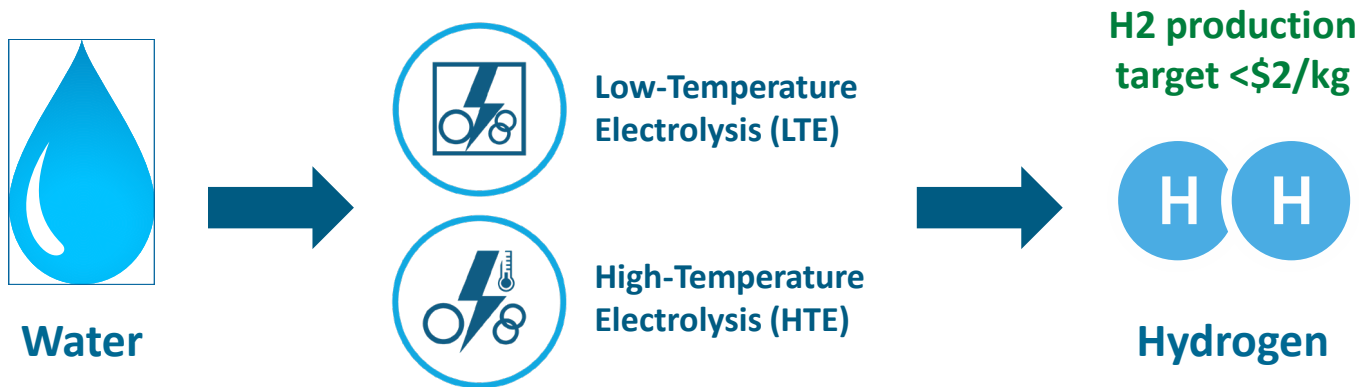
2022 Annual Merit Review and Peer Evaluation Meeting

Project ID # P196

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Goal: H2NEW will address components, materials integration, and manufacturing R&D to enable manufacturable electrolyzers that meet required cost, durability, and performance targets, simultaneously, in order to enable \$2/kg hydrogen (by 2025 on way to H2 Shot target, \$1/kg by 2031).



H2NEW has a clear target of establishing and utilizing experimental, analytical, and modeling tools needed to provide the scientific understanding of electrolysis cell performance, cost, and durability tradeoffs of electrolysis systems under predicted future operating modes

Timeline and Budget

- Start date (launch): **October 1, 2020**
- FY21 DOE funding: **\$10M**
- Anticipate minimum \$50M over 5 years
- 75% low temperature electrolysis due to higher TRL and intermittency

Barriers

- **Durability**
- **Cost**
- **Efficiency**

Consortium Team*



- NIST
- UC-Irvine
- Carnegie Mellon Univ
- Colorado School of Mines

* Expansion to include additional academic and industrial partners in anticipated in the near future.

DOE Managers: David Peterson, Ned Stetson, Will Gibbons, Katie Randolph, Eric Miller, James Vickers, Sunita Satyapal
NREL: Bryan Pivovar, Shaun Alia, Mike Ulsh, Guido Bender, Nick Wunder, James Young, Jason Zack, Scott Mauger, Carlos Baez-Cotto, Jason Pfeilsticker, Sunil Khandavalli, Allen Kang, Elliot Padgett, Dave Ginley, Sarah Shulda, Alex Badgett, Rachel Rubin, Mark Ruth, Chai Engrakul, Chang Liu
LBNL: Nemanja Danilovic, Ethan Crumlin, Ahmet Kusoglu, Michael Tucker, Adam Weber, Julie Fornaciari, Claire Arthurs, Arthur Dizon, Jiangjin Liu, Grace Anderson, Rebecca Hamlyn, Jason Lee, Grace Lau
ANL: Debbie Myers, Rajesh Ahluwalia, C. Firat Cetinbas, Andrew Star, Jui-Kun Peng, Dennis Papadias, Xiaohua Wang, Nancy Kariuki, Jaehyung Park
LANL: Rangachary Mukundan, Siddharth Komini Babu, Xiaoxiao Qiao
ORNL: Alexey Serov, Erin Creel, Dave Cullen, Haoran Yu, Jefferey Baxter, Shawn Reeves, Michael Zachman, Harry Meyer, Michael Kirka, Christopher Ledford
INL: Richard Boardman, Dong Ding, Micah Casteel, Lucun Wang, Jeremy Hartwigsen, Josh Gomez, Daniel Wendt, Fred Steward
PNNL: Olga Marina, Jamie Holladay, Chris Coyle, Kerry Meinhardt, Dan Edwards, Matt Olszta, Nathan Canfield, Lorraine Seymour, Nathanael Royer, Jie Bao, Brian Koeppel
LLNL: Brandon Wood, Joel Berry, Penghao Xiao, Tim Hsu, Namhoon Kim
NETL : Harry Abernathy, Greg Hackett
NIST: Daniel Hussey, David Jacobson
University of California – Irvine: Iryna Zenyuk, Devashish Kulkarni, Andrea Perego, Yu Morimoto
Colorado School of Mines: Svitlana Pylypenko, Genevieve Stelmacovich , Matthew Coats, Jayson Foster
Carnegie Mellon University: Shawn Litster, Kara Ferner, Fausto Pasmay

- **Covers \$9.5B** for clean hydrogen:
 - \$8B for at least four regional clean hydrogen hubs
 - \$1B for electrolysis research, development and demonstration
 - \$500M for clean hydrogen technology manufacturing and recycling R&D



President Biden Signs the Bipartisan Infrastructure Bill on November 15, 2021.

Photo Credit: Kenny Holston/Getty Images

- **Aligns with Hydrogen Shot priorities by directing work to reduce the cost of clean hydrogen to \$2 per kilogram by 2026**
- **Requires developing a National Hydrogen Strategy and Roadmap**

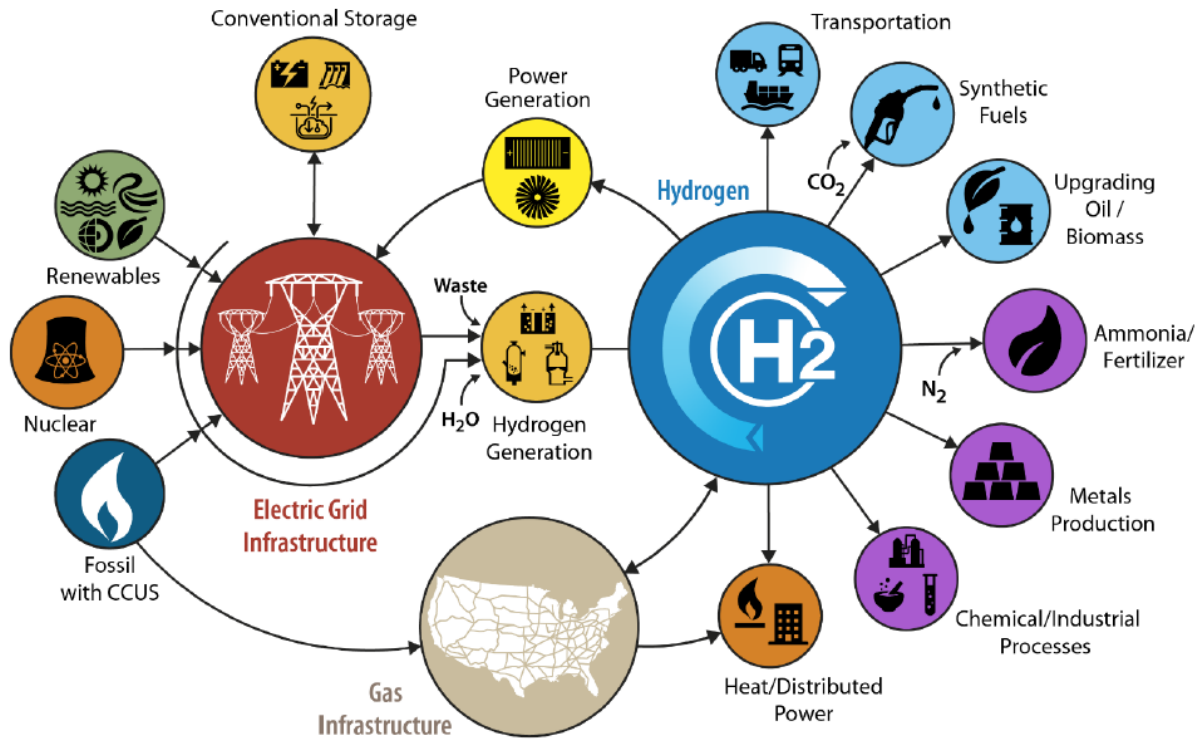
<https://www.energy.gov/sites/default/files/2021-12/h2iq-12082021.pdf>

- Workshops supporting the Bipartisan Infrastructure Law (BIL)
 - H2- PACE (Power Electronics) 12/2-12/3, 2021
 - **H2 – LAWE (Alkaline Electrolysis) Jan 26-27, 2022**
 - Bulk storage of gaseous hydrogen (Feb 10-11, 2022)
 - Liquid hydrogen (Feb 22-23, 2022)
 - **High Temperature Electrolysis Manufacturing (March 8-9, 2022)**
 - **Advanced Materials for PEM Electrolysis (March 30-31)**
 - **Manufacturing and Recycling (H2-MACH, May 24-26)**

Bold denotes - H2NEW played leading/co-organizing role in Workshop

<https://www.energy.gov/eere/fuelcells/workshop-and-meeting-proceedings>

Relevance – H2NEW connection to H2@Scale



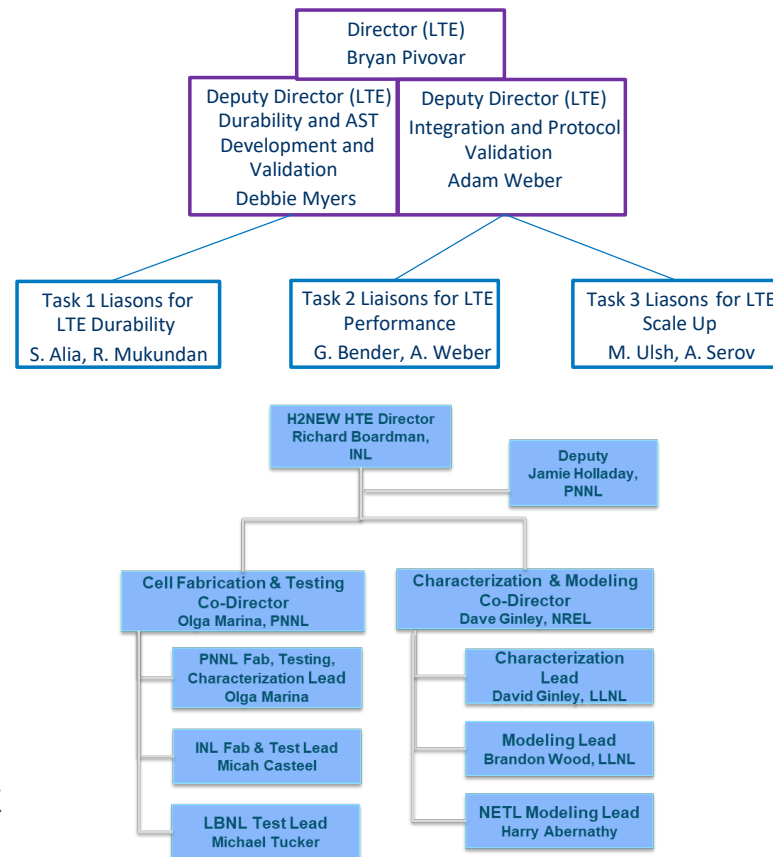
Illustrative example, not comprehensive
<https://www.energy.gov/eere/fuelcells/h2-scale>

- Making, storing, moving and using H2 more efficiently are the main H2@Scale pillars and all are needed.
- Making H2 is the inherently obvious, first step to spur the wide-ranging benefits of the H2@Scale vision.
- Electrolysis has most competitive economics and balances increasing renewable generation challenges.
- Timeframe is short, competition intense, coordinated effort critical for domestic competitiveness.

Approach – Consortium Structure and Support Posters

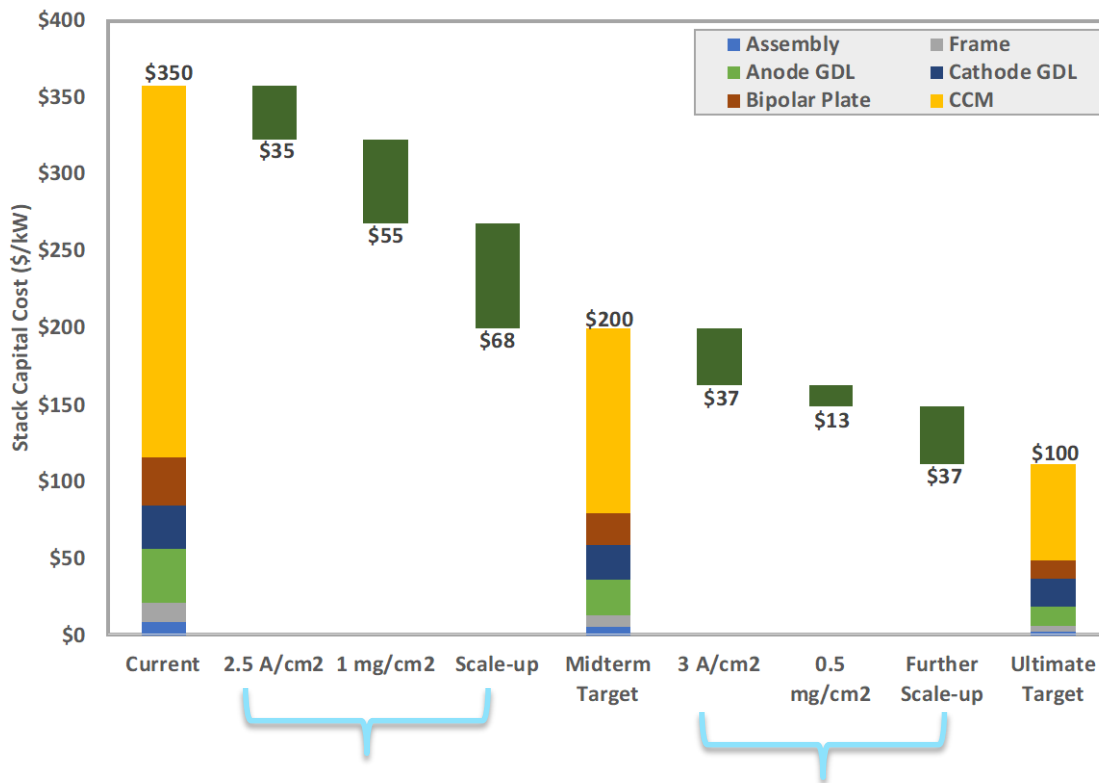
- Task Based Structure Split between:
- Low Temperature Electrolysis (LTE), 75%:
 - Task 1 Durability (P196a)
 - Task 2 Performance (P196b)
 - Task 3 Scale-up (P196c)
 - Task 3c Analysis (P196d)
- High Temperature Electrolysis (HTE), 25%:
 - Task 5 Durability and AST Development (P196e)
 - Task 6 Characterization (P196f)
 - Task 8 Modeling (P196g)

See associated [AMR posters](#) for additional details on Approach, Accomplishments, and Future Work



H2NEW Activities: Low Temperature Electrolysis (LTE)

Relevance: Stack Costs (PEM Centric)



| Stack Targets | Status | 2023 | 2025 |
|-----------------------------------|--------|------|------|
| Cell (A/cm ² @1.9V) | 2.0 | 2.5 | 3.0 |
| Efficiency (%) | 66 | 68 | 70 |
| Lifetime (khr) | 60 | 70 | 80 |
| Degradation (mV/khr) | 3.2 | 2.75 | 2.25 |
| Capital Cost (\$/kW) | 350 | 200 | 100 |
| PGM loading (mg/cm ²) | 3 | 1 | 0.5 |

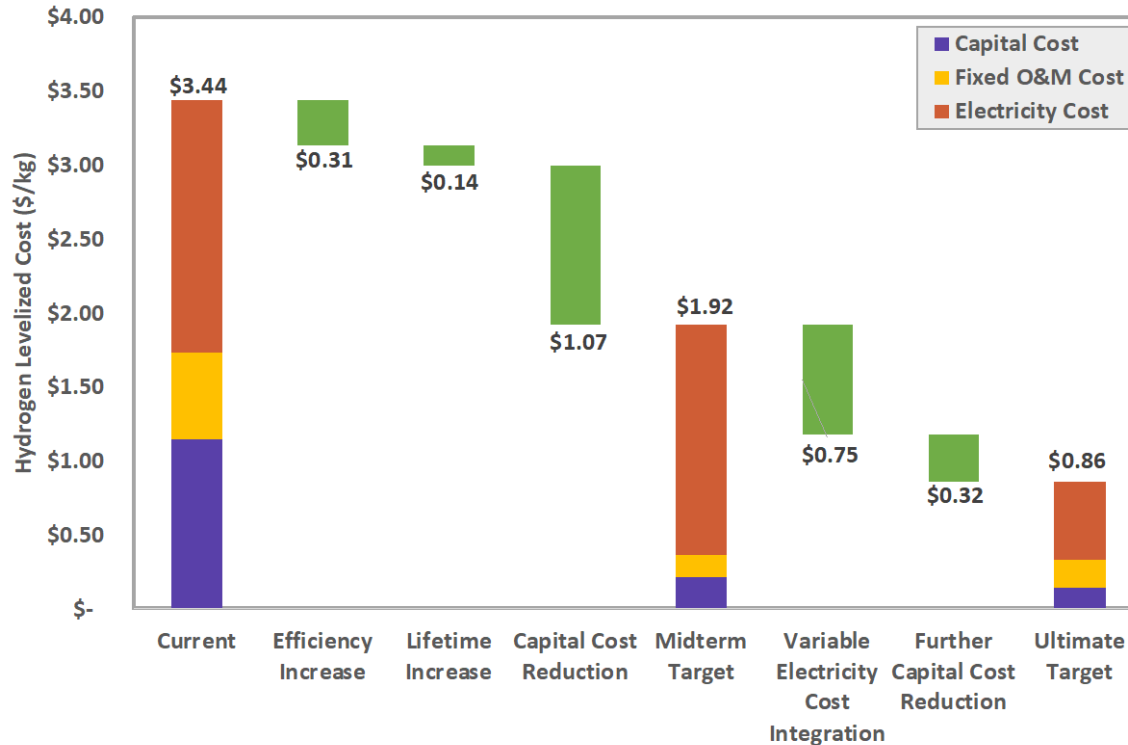
These 3 areas

1. Increased efficiency/current density
2. Decreased PGM loading
3. Scale-up

Are the strongest levers for addressing stack costs.

https://www.hydrogen.energy.gov/pdfs/review21/p196_pivovar_boardman_2021_o.pdf

Relevance: Hydrogen Levelized Cost (PEM Centric)



Select pathway to \$2/kg and \$1/kg identified.

Much of HLC gains possible through greatly decreasing capital costs and enabling lower cost electricity through variable operation.

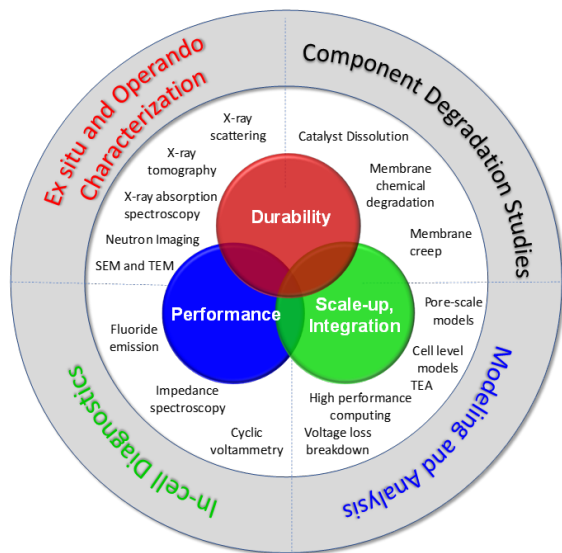
These advances can't come with compromised durability or efficiency, so all three areas are linked.

Approach: Select LTE Milestones*

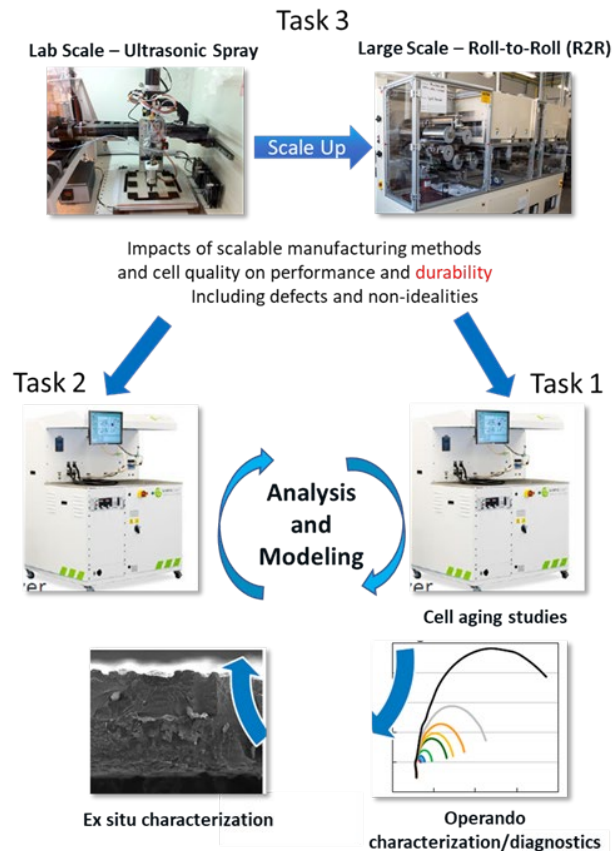
| Milestone Name/Description | Due Date | Type | Status |
|---|-----------|-----------|----------|
| Assess the impact of the stress tests on electrode, membrane, and PTL through voltage loss breakdown and use ex-situ characterization to identify degradation mechanisms and support AST development. Determine initial stressors to be investigated in detail for potential to accelerate specific degradation mechanisms. Labs – NREL, LBNL, ANL, LANL. | 6/30/2021 | QPM | Complete |
| Establish and validate single cell performance testing protocols on SOA MEAs with PGM loading of <math><0.8 \text{ mg/cm}^2</math> and membranes <math><100 \text{ }\mu\text{m}</math> demonstrating a minimum performance of 2 A/cm ² at 1.8 V and demonstrate ability to perform voltage loss breakdown modeling as verified by agreement within 10% across at least 3 labs. | 9/30/2021 | QPM | Complete |
| Targeting increased efficiency through membrane thickness reductions: Perform H2 crossover experiments between atmospheric and 30 bar backpressure, and current densities up to at least 3 A/cm ² to work towards an improved understanding of the mechanism and nature of hydrogen crossover, and its dependency with current density. | 12/31/21 | QPM | Complete |
| Targeting cost gains from scaling up processes: Using an automated, scalable coating manufacturing process for MEAs, match or exceed performance of baseline MEA (within 20 mV at current densities up to 2 A/cm ²) and project potential cost reductions through scaleup. (NREL, LBNL, ORNL) | 3/30/2022 | QPM | Complete |
| Quantify membrane chemical degradation rates under three different operating conditions including simulated (without current) and electrolyzer operating conditions using fluoride emission rate at the anode and cathode. | 6/30/2022 | QPM | On Track |
| Development and validation of AST protocols for PEM LTE/SOEC HTE. Propose at least 3 component-level LTE and HTE ASTs (each) and compare cell and AST performance loss. | 9/30/2022 | Milestone | On Track |

*For complete list, see associated [AMR posters](#) for additional details

Approach: LTE Tasks



- Durability (Task 1, P196a)
 - Establish fundamental degradation mechanisms
 - Develop accelerated stress tests
 - Determine cost, performance, durability tradeoffs
 - Develop mitigation
- Performance (Task 2, P196b)
 - Benchmark performance
 - Novel diagnostic development and application
 - Cell level models and loss characterization
- Scale-up (Task 3, P196a)
 - Transition to mass manufacturing
 - Correlate processing with performance and durability
 - Guide efforts with systems and techno-economic analysis (Task 3c, P196d)



Accomplishments – Select LTE Highlights Presented 2022 AMR

- Summary of Highest Impact Efforts in Last Year
 - Task 3c – Systems/Technoeconomic Analysis
 - System efficiency
 - GPRA
 - Wind/Solar/Battery
 - Task 1 - Durability
 - Start/Stop
 - Ir dissolution/modeling
 - Membrane
 - Task 2 - Performance
 - H₂ Xover
 - Voltage loss breakdown
 - Task 3 – Scale-up
 - Low loadings
 - Ink aging
 - PTL (cross task effort)

Low Temperature Electrolysis (LTE)

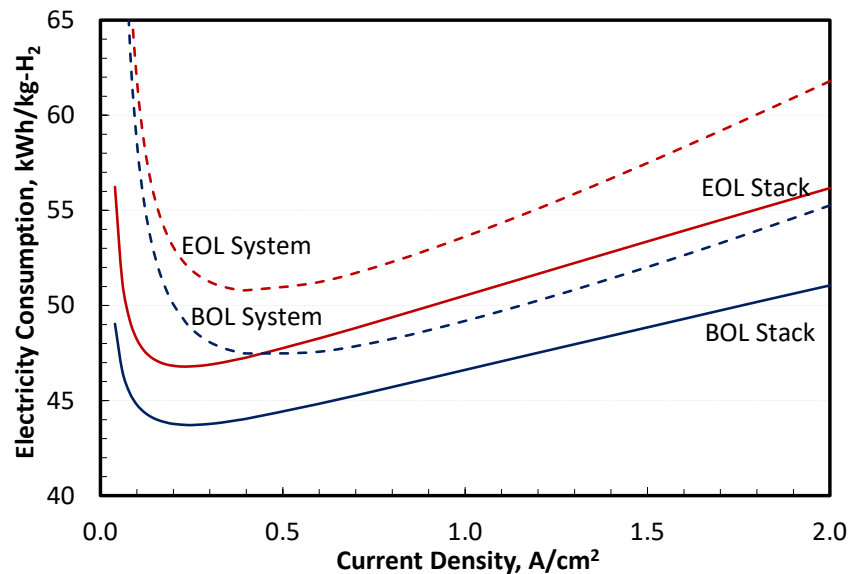
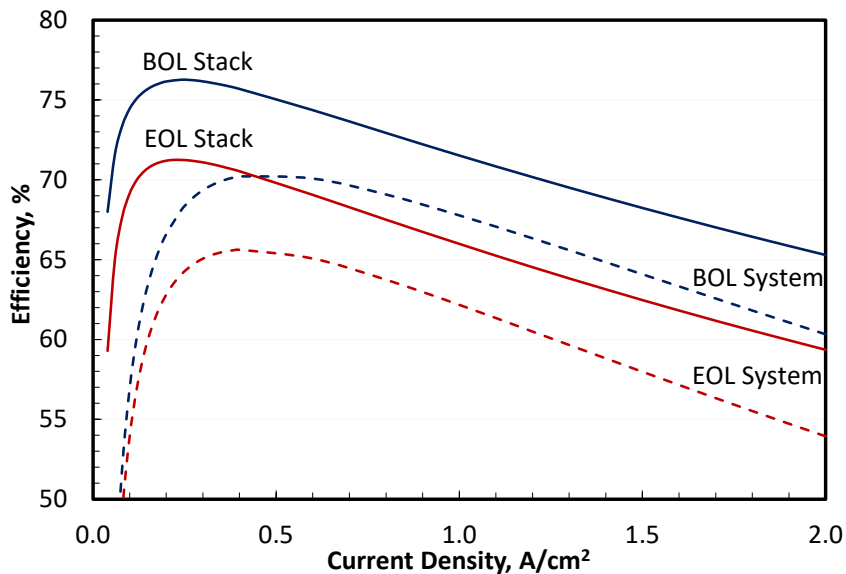
Task 3c: Analysis (more content in AMR Poster [P196d](#))

Accomplishment: Models Determine System Performance and Degradation

Efficiency Definitions

- Stack efficiency: LHV of net H₂ produced in the cathode divided by stack input power
- System efficiency: LHV of H₂ supplied to the pipeline divided by the total electrical power supplied to the transformer/rectifier and other BOP components

| | Efficiency (%) | | Efficiency (kWh/kg-H ₂) | |
|------------------------|----------------|------|-------------------------------------|------|
| | BOL | EOL | BOL | EOL |
| Stack | | | | |
| Rated Power Efficiency | 65.3 | 59.3 | 51.1 | 56.2 |
| Peak Efficiency | 76.3 | 71.2 | 43.7 | 46.8 |
| System | | | | |
| Rated Power Efficiency | 60.3 | 53.9 | 55.3 | 61.8 |
| Peak Efficiency | 70.2 | 65.6 | 47.5 | 50.8 |



Approach: Hydrogen Cost Analysis

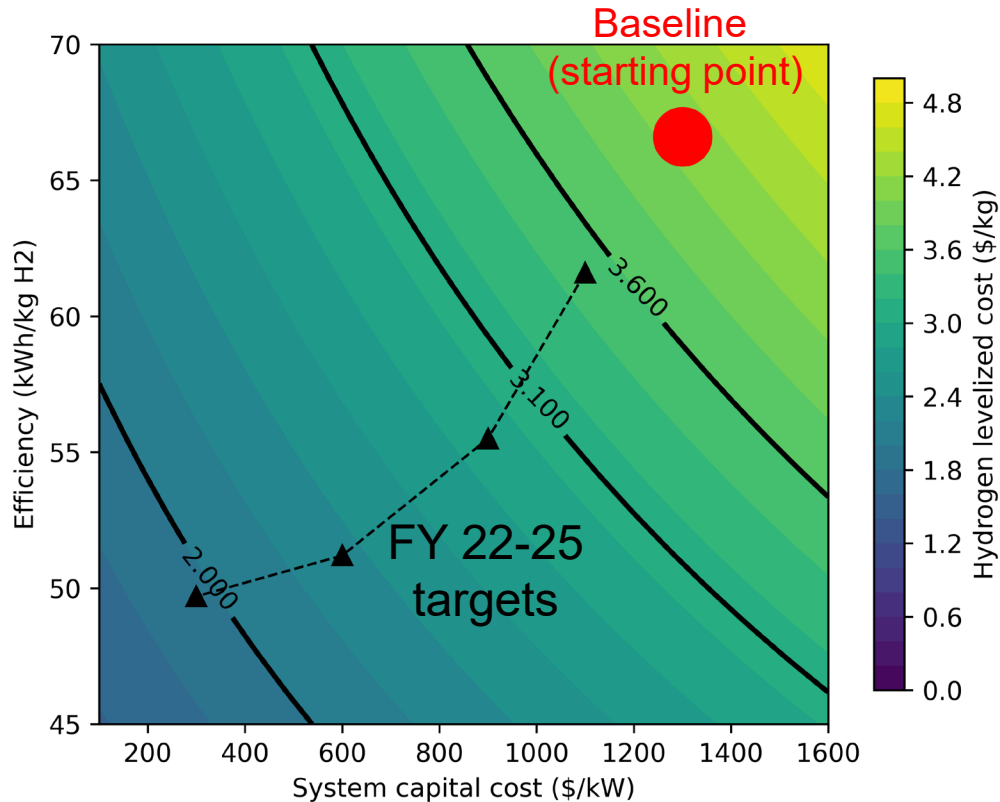
- Conduct analysis consistent with objectives Q2 milestone, meeting the following:

“Using the baseline PEM electrolyzer system model and the H2A Model, quantify at least three distinct pathways based on improvements to system cost, manufacturing, performance, and lifetime needed to achieve the H2 production cost targets of \$3.60/kg and \$3.10/kg by end of FY22 and FY23, respectively.”

- Approach: Use H2A to map pathways to cost targets as a function of key economic inputs
- Capacity factor – fixed
- Electricity price – fixed
- **Capital cost**
- **Efficiency**
- **Lifetime**

| Parameter | Baseline value | Units |
|----------------------------|----------------|--------------------------|
| Capacity factor | 90% | % |
| Electricity price | 0.03 | \$/kWh |
| Stack efficiency | 50.8 | kWh/kg H2 |
| System efficiency (LHV) | 66.6 (50% LHV) | kWh/kg H2 (% LHV) |
| Replacement lifetime | 40,000 | hours |
| Estimated replacement cost | 15% | % of direct capital cost |

Accomplishment: Mapping possible \$2/kg pathways

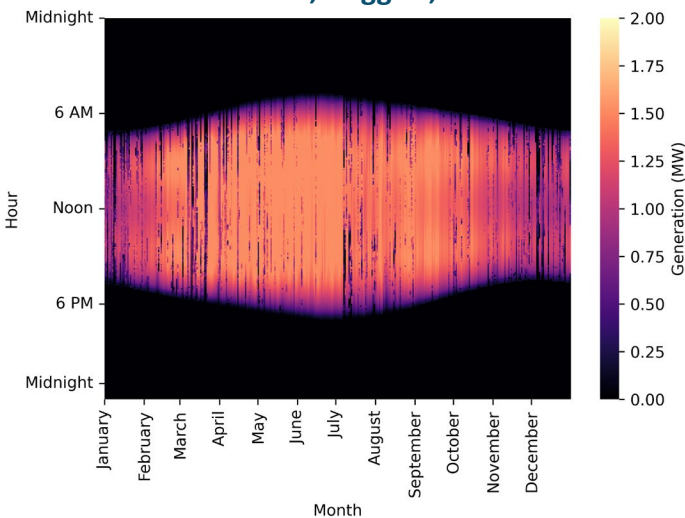


- Between efficiency and capital cost, several possible paths exist to meet FY22 and FY23 targets
- Pathways could focus on advancing one parameter or a combination of both

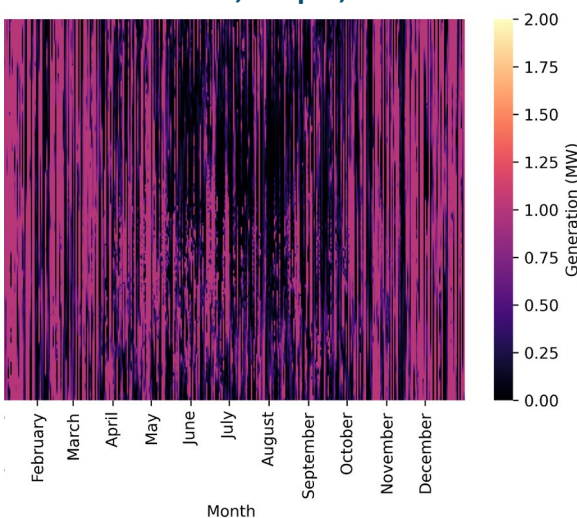
Approach: Optimal Operations Strategy for a Variety of Electricity Sources and Prices

- Task also involves analyzing LTE electrolyzers directly connected to wind and photovoltaic solar electricity generation
- NREL PySAM modeling platform is used to simulate outputs from PV/wind at any location or capacity
- Battery energy storage can also be added to the electrolyzer-renewable system

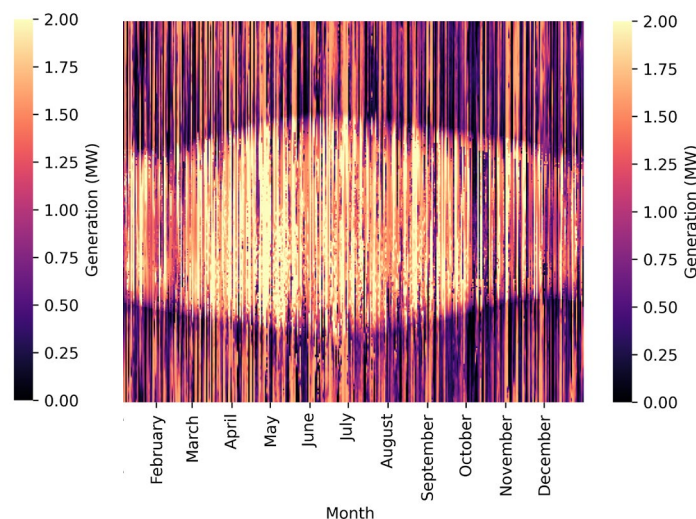
1 MW PV, Daggett, CA



1 MW wind, Casper, WY



1 MW wind + 1 MW PV, Amarillo, TX

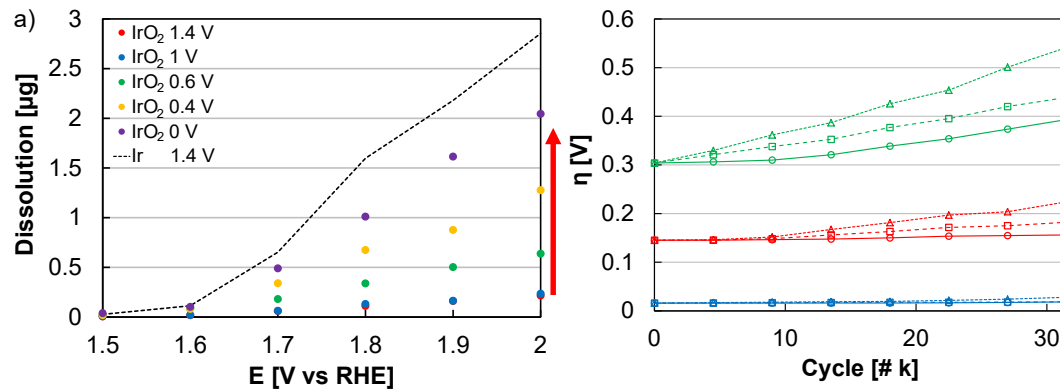


Low Temperature Electrolysis (LTE)

Task 1: Durability (more content in AMR Poster [P196a](#))

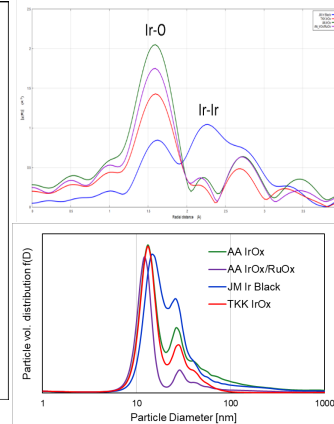
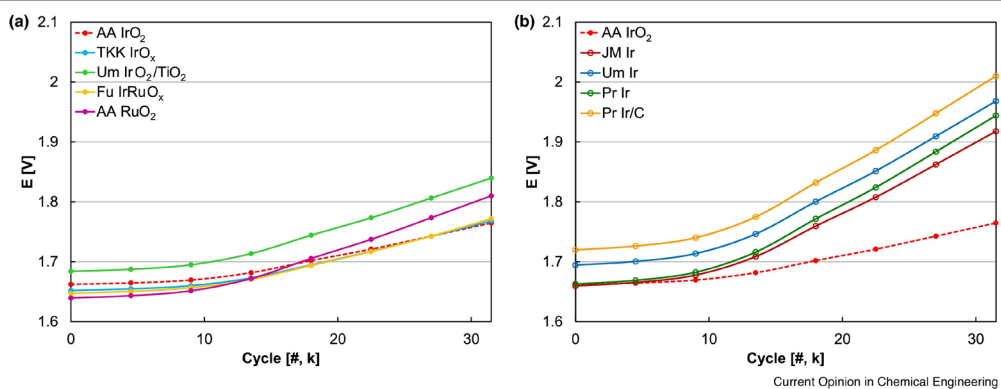
Accomplishment: Accelerated Stress Test - Start-Stop

- MEAs: Nafion 117 (to mitigate crossover concerns)
- 0.1 mg_M cm⁻² anode catalyst loading
- Start-stop: Voltage control 0 to 2 V triangle (vs. intermittent: 1.45 to 2 V triangle)
- Summary of findings:
 - ✓ Higher performance losses for AST cycle through catalyst redox (0 to 2 V): major changes in particle size, not final oxidation state
 - ✓ Performance losses correlate with fraction of Ir in metallic state
 - ✓ Loss mechanism: thinning/increased PTL site-access, clear agglomeration, increased interfacial tearing



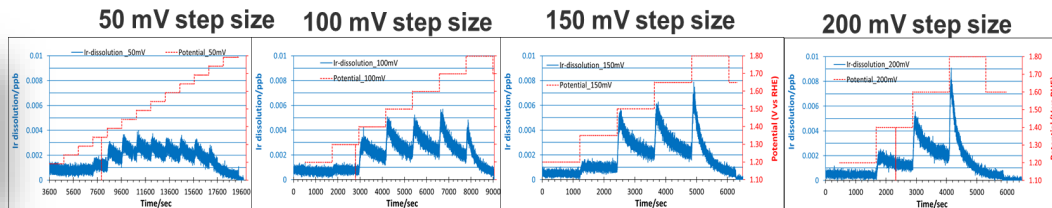
Soft shutdown

Loss of water, temperature



S.M. Alia, *Current Opinion in Chemical Engineering*, 2021, 33, 100703.

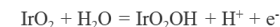
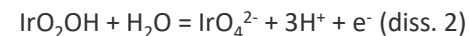
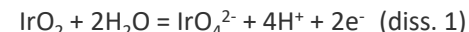
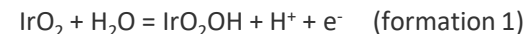
Accomplishment: Degradation studies of IrO₂ anode catalyst



Staircase Potential

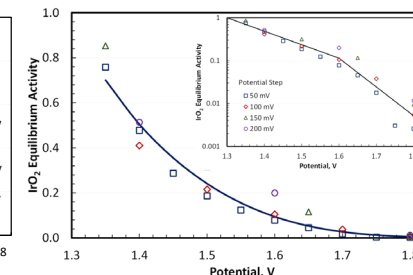
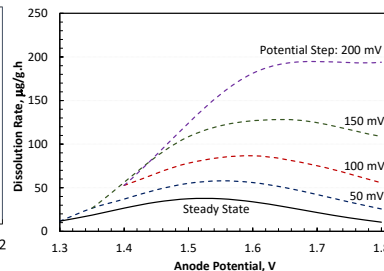
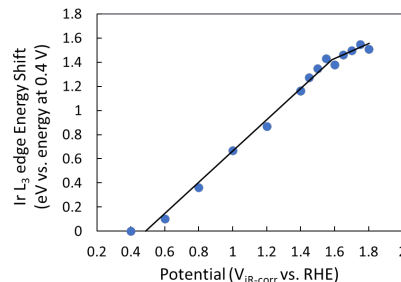
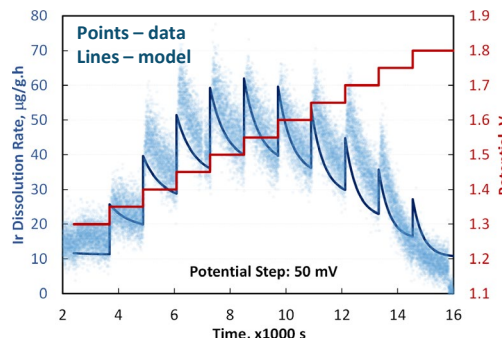
- Dissolution increases after stepping to UPL followed by passivation and decreases with continued hold at potential.
- Dissolution increases with increase in step size
- 200 mV > 150 mV > 100 mV > 50 mV

2-Species Dissolution Model



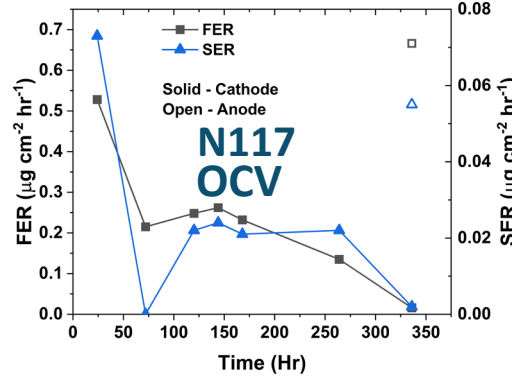
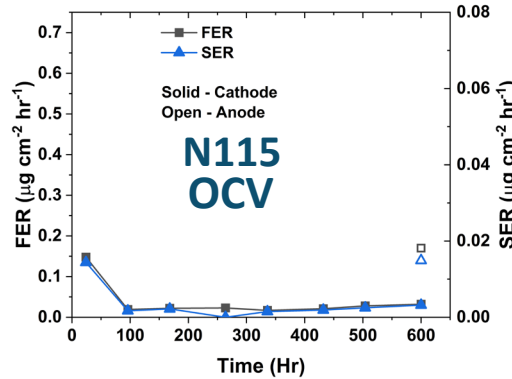
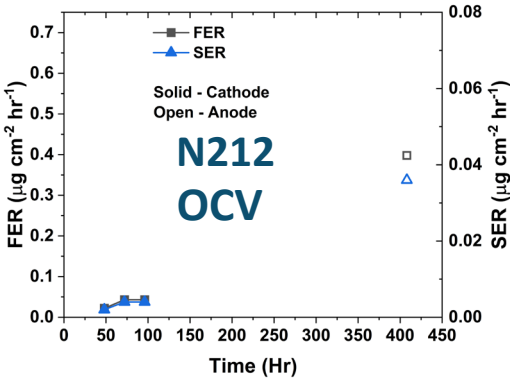
$$i_1^f = k_1^f(a_1)(a_1 - a_1^{eq})e^{\left(\frac{\alpha_1 F}{RT}(E - E_{0,1}^f)\right)} = k_1^f(a_1)(a_1 - a_1^{eq})$$

- On-line inductively-coupled plasma mass spectrometry Ir dissolution data modeled with dissolution of two species: IrO₂ and IrO₂OH, with the latter having slower dissolution kinetics
- IrO₂ activity decreases with increasing potential; Change in potential dependence at 1.6 V suggests an appearance of a different IrO_x species – corresponding with in situ X-ray data
- Under steady-state conditions, catalyst is least stable at potentials between 1.5 and 1.55 V; 37.5 mg/g·h peak dissolution rate



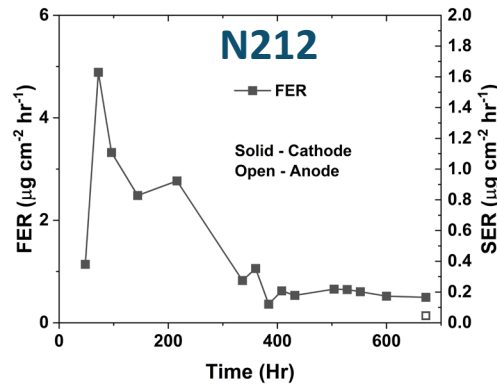
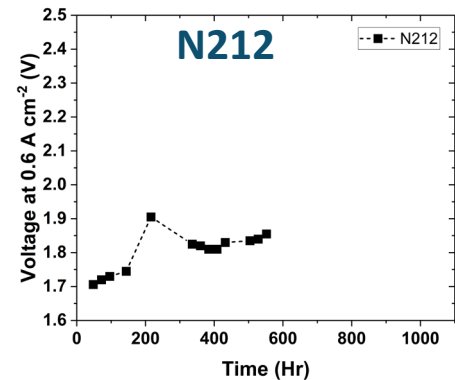
Accomplishment: Membrane Degradation

Cell Temperature: 80 ° C : Anode: 0.4 mg cm⁻² IrO₂ (Alfa Aesar) ; Cathode: 0.1 mg cm⁻² Pt/C (TEC10V20E).



Under simulated OCV conditions

- Anode FER > Cathode FER
- FER increases with initial fluoride inventory
- FER decreases with decreasing crossover



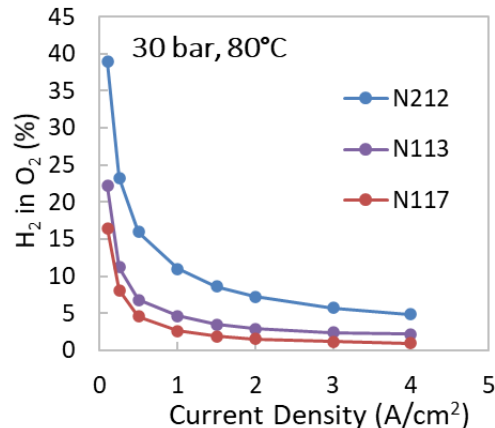
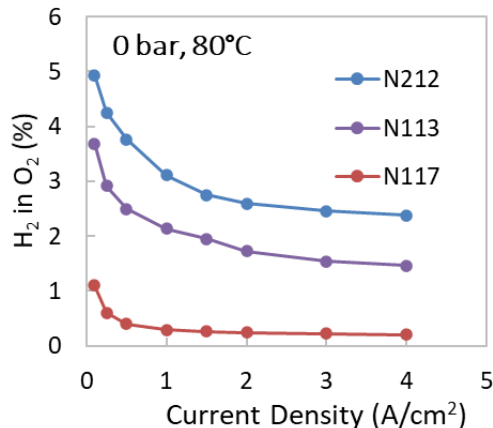
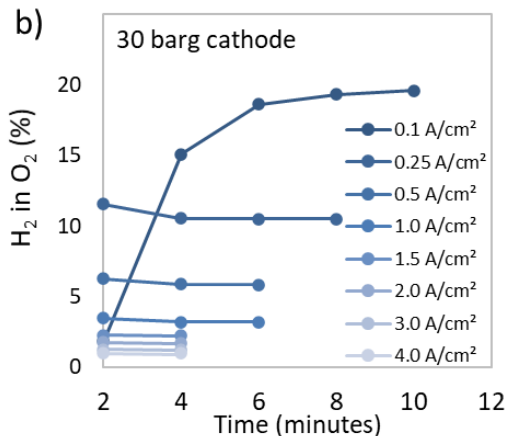
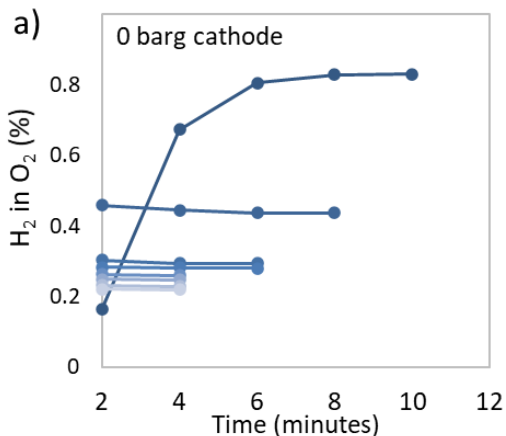
Under electrolyzer operating conditions

- Cathode FER > Anode FER
- Water generated at the cathode of the electrolyzer can be used as surrogate for membrane degradation rates (consistent with literature)
- FER is initially high but stable after 300 hours of testing
- Further testing under different conditions:
 - Membrane thickness
 - Hydrogen back pressure
 - Temperature
 - Current density

Low Temperature Electrolysis (LTE)

**Task 2: Performance and Benchmarking (more content in
AMR Poster [P196b](#))**

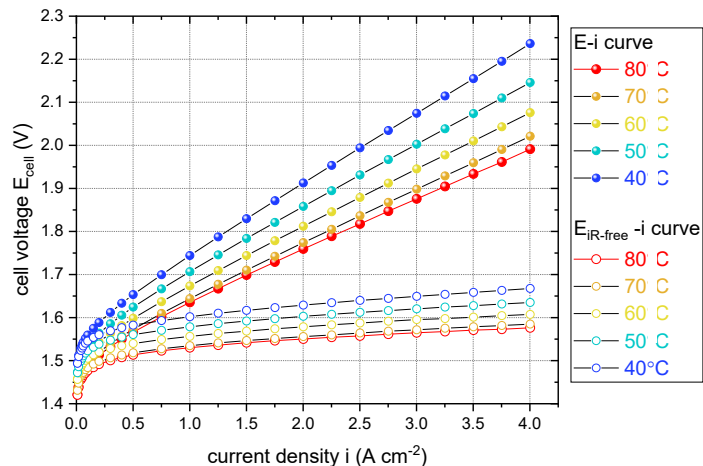
Accomplishments – Hydrogen Crossover (FY'22 Q1 QPM)



- Measured H₂Xover between atmospheric and 30 bar backpressure at various current densities up to 4 A/cm²
 - Leveraged diagnostic equipment developed within EMN HydroGEN project
 - Further advanced method to allow for rapid semi-automated operation
 - 2 min sampling time at most current densities
- ⇒ Gas mixture of anode exhaust stream can reach flammable H₂:O₂ ratios
- ⇒ Strong dependency of H₂:O₂ ratio on
- Current density
 - Operating pressure
 - Membrane thickness

Accomplishments – Benchmarking Voltage Breakdown Analysis

Data Set of 5 different temperatures



- Round Robin evaluation of Voltage Breakdown Analyses across four H2NEW Laboratories
- Deviating results originating from
 - Calculation of thermodynamic reversible cell voltage E_0
 - Determination of the “linear” Tafel region
- Harmonization of Voltage Breakdown Analysis through
 - Development of protocol
 - Harmonization of equations
 - Development of statistical tool that supports the selection of the “linear” Tafel region and enables the reproducible determination of the Tafel slope

Harmonized Thermodynamic Equation

$$E_{p,T}^0 = E^0 - \frac{S_{R,T^*}(T - T^*)}{2F} + \frac{2.303RT}{2F} \cdot \log \left[\frac{(a_{O_2})^{0.5} a_{H_2}}{a_{H_2O}} \right]$$

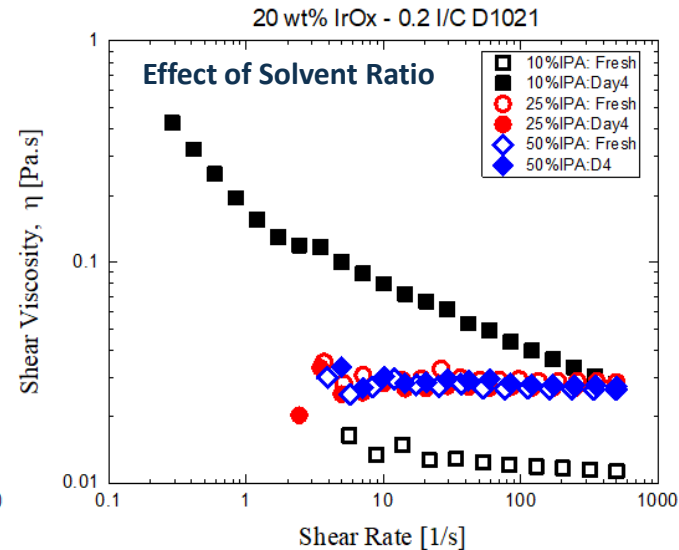
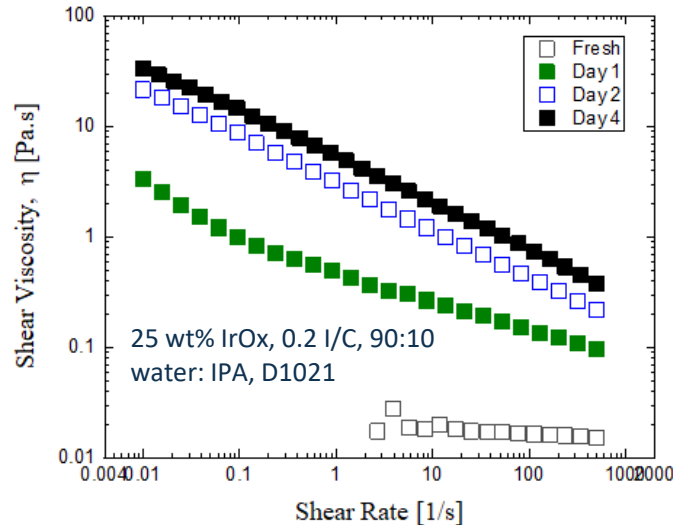
- ⇒ Successful benchmarking of Voltage Breakdown Analysis of low loading, high performance cell across at all participating four laboratories
- ⇒ Publication in preparation for dissemination to community

Low Temperature Electrolysis (LTE)

Task 3: Manufacturing, Scale-up and Integration (more content in AMR Poster [P196c](#))

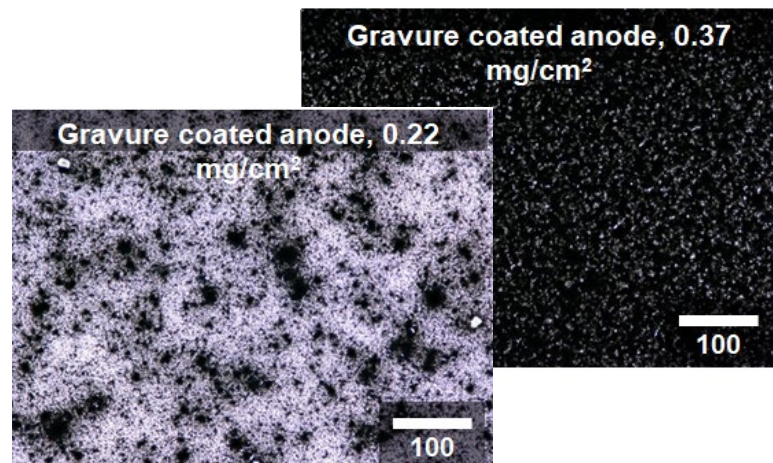
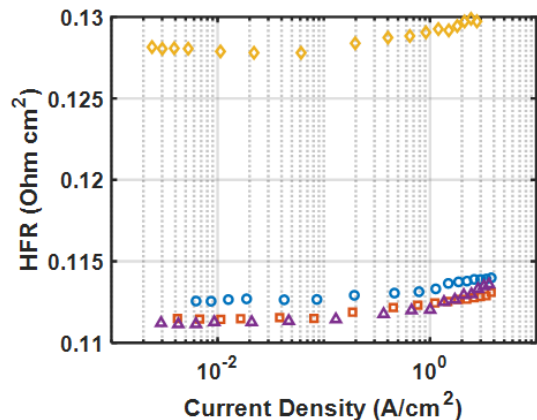
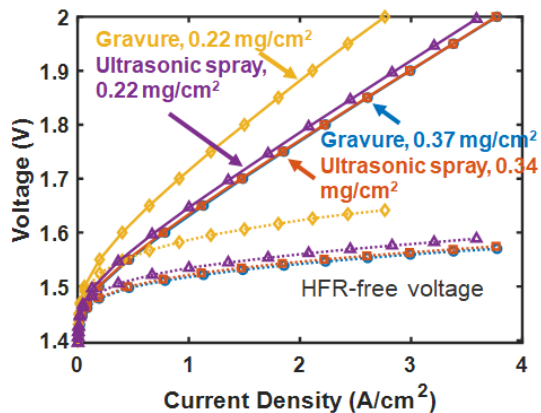
Accomplishment: Thickening of anode ink by aging

- Objective: Low viscosity of IrOx inks can be problematic for large-scale coating, and the catalyst particles are observed to crash out in ~a day, exacerbating coating issues. We want to further explore and understand parameter effects of the observation that anode ink “resting” can improve processability.
- Anode inks experiencing a multi-day dwell via simple rolling can increase in viscosity by several orders of magnitude
- However, this behavior is only observed for water-rich formulations
 - Observed for water mixtures with both IPA and nPA



Accomplishment: R2R Anode Loading Comparison

- Objective: evaluate R2R anode coating efficacy at different loadings
 - Study R2R gravure-coated anodes at 0.2 and 0.4 mg Ir/cm²
 - Compare performance of R2R coatings to spray-coated FuGeMEA at same loading



- Result:
 - Whereas R2R anode showed comparable performance to FuGeMEA at 0.4 loading, the R2R anode performance at 0.2 loading was much worse
 - Optical microscopy (transmission) showed poor micro-scale uniformity of the lower loaded R2R anode
 - Need additional studies to improve the uniformity of low-loaded anodes via R2R processes**

Low Temperature Electrolysis (LTE)

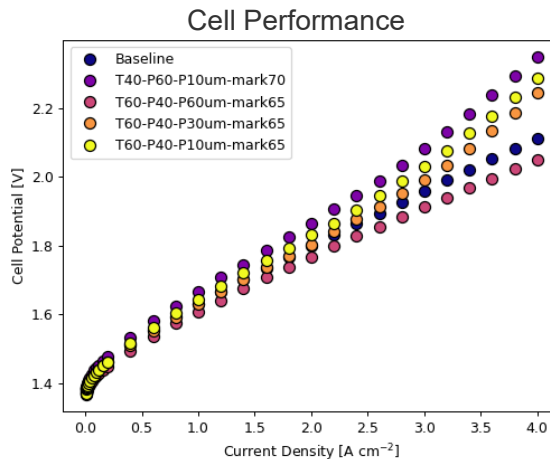
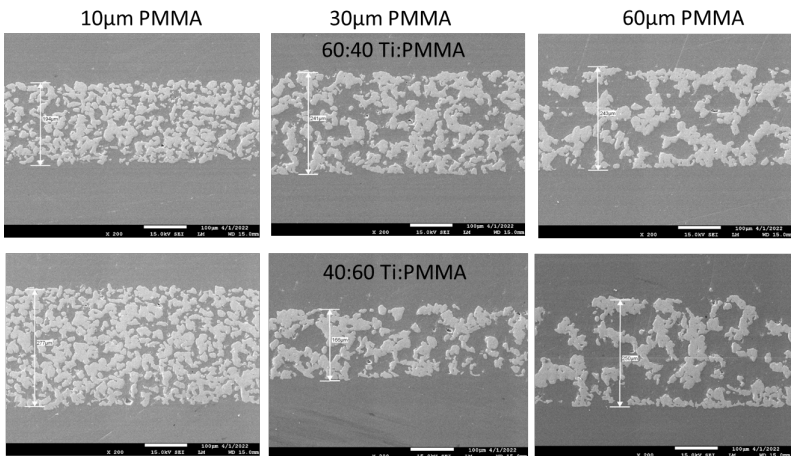
PTLs (Porous Transport Layers)

Cross-task effort (additional information found in Posters 196a, b and c)

Accomplishment: Ti PTL Pore Structure Optimization

- Produced and tested PTLs with range of poreformer size and loading
- Result: 60 v% Ti – 40v% 60 μ m poreformer outperforms baseline commercial PTL

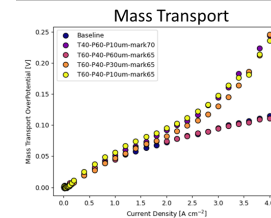
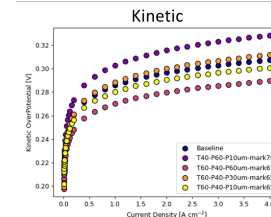
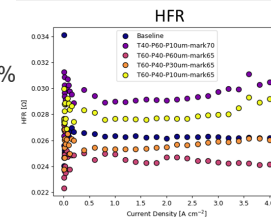
Tunable pore size and structure



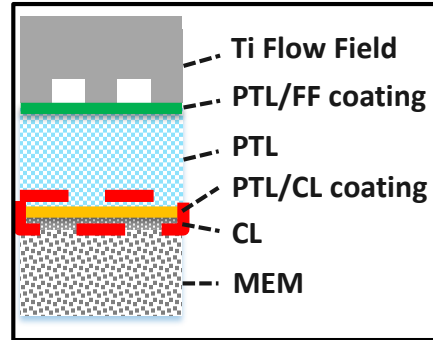
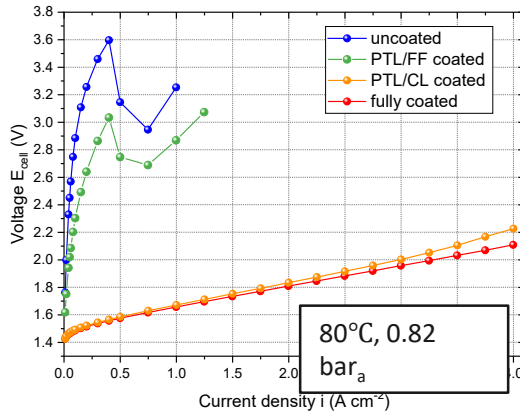
10 μ m PMMA 40v%
10 μ m PMMA
30 μ m PMMA
Mott
60 μ m PMMA

Conditions:

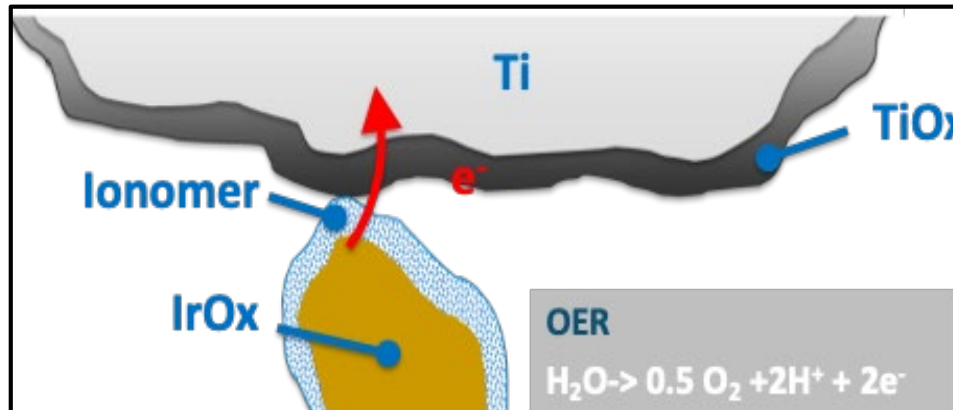
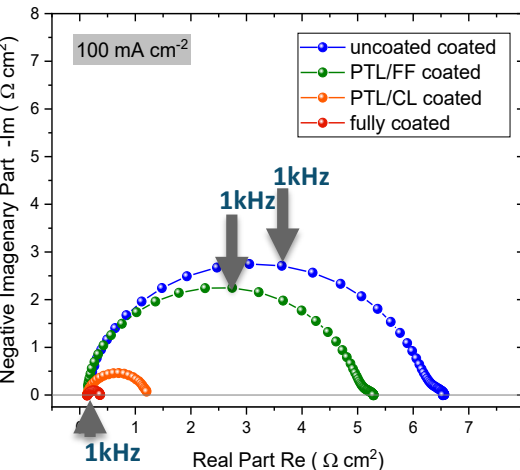
- 80°C
- Water flow in anode and fully humidified H₂ in cathode
- N117 membrane
- anode: Tkk IrO₂ (0.4 mg/cm²)
- cathode: Tkk10V50E Pt/C (0.1 mg/cm²)
- PTL thickness: 10 thou / gasket 10 thou
- GDL: Toray 120 / gasket 10 thou



Accomplishments – In-depth Study of PTL / Anode Interface



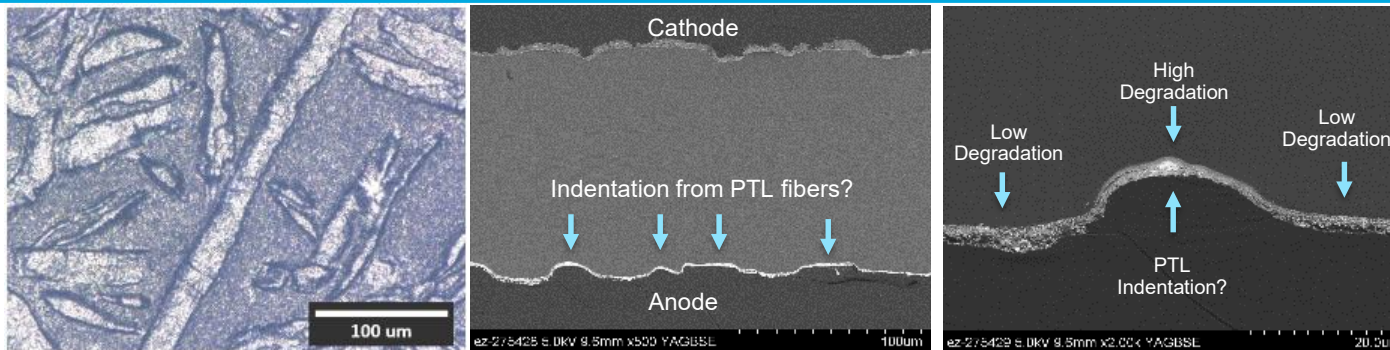
- Systematic study on the effect of PTL coatings for MEAs with low loading (FuGeMEA)
- PGM coating at the anode PTL/CL interface governs performance
- HFR data up to 300kHz indicate resistance is not the cause
- TiOx acts as a semiconductor and seems to be limiting the electron transfer



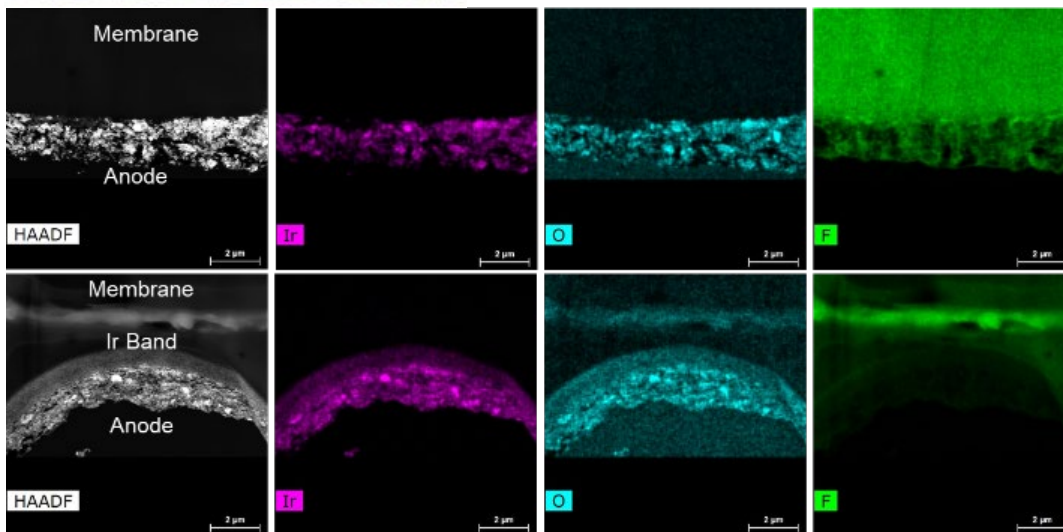
⇒ Pt interlayer creates Pt/TiOx junction

- Schottky diode metal/n-type
- Lower ohmic resistance

Accomplishments- Tracking Inhomogeneous Degradation with Electron Microscopy



- Optical images show indentations in anode electrode from PTL fibers
- Similar indentations visible in cross-sectional SEM images correlate with degree of degradation



Low Degradation Regions:

- Limited Ir dissolution
- Similar electrode thickness and density to beginning-of-test MEA

High Degradation Regions:

- Band of dissolved Ir in membrane
- Denser electrode structure
- Curved electrode suggests direct contact with PTL

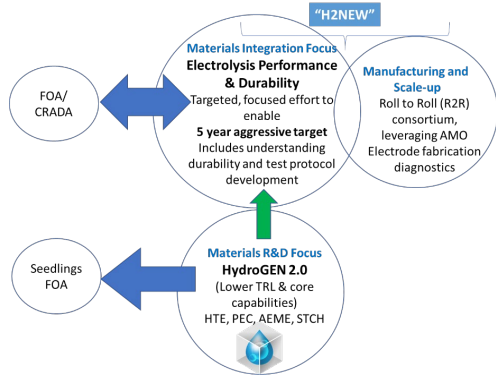
Responses to Previous Year Reviewers' Comments

- **Reviewer Comment:** The project is performed solely by national laboratories, so industry participation is largely missing.
- **Response:** We have done what we can to incorporate this through our Advisory Board and other engagement with industrial partners, but also expect this to change in the near future as additional formal engagement opportunities develop.

- **Reviewer Comment:** Alkaline electrolysis is not addressed in the project, yet there have been substantial technology advances in recent years, and it is likely to capture a large market share in 2030 and beyond.
- **Response:** The Bipartisan Infrastructure Law identifies this area as a target area, H2NEW co-sponsored H2-LAWE Workshop/Report in this area, and we anticipate it being included in the near future, we have already begun increasing capabilities to address.

- **Reviewer Comment:** There is no mention of other emerging technologies, such as HTE proton-conducting technology or LTE anion exchange membrane technology.
- **Response:** This was by design and is beyond the scope of H2NEW for now and in the realm of HydroGEN (2.0).

- **Reviewer Comment:** The project is focused on standard technologies and material sets. It would be great to see some investigation into promising alternatives that have the potential to leapfrog over the existing state of the art.
- **Response:** H2NEW co-sponsored H2-AMP Workshop/Report in this area to help identify research needs in the area of advanced materials for PEM electrolysis, and we anticipate novel material development and incorporation being included in the near future.



Leverage across other Consortium:
 HydroGEN 2.0 (HFTO)
 R2R (AMO)
 Million Mile Fuel Cell Truck (HFTO)
 ElectroCAT (HFTO)

LTE Strategic Advisory Board Members



Kathy Ayers
 VP R&D
 Nel Hydrogen

Cortney Mittelsteadt
 VP Electrolyzer
 Technology
 Plug Power

Andy Steinbach
 Specialist
 Materials
 Science
 3M

Jack Brouwer
 Professor
 U.C. Irvine

Mark Mathias
 Consultant
 retired (GM)

Numerous industrial, academia, and international interactions:
 (IEA, ASTWG, materials suppliers, informal collaborations)

Select group of advisors representing OEMs, tier 1 suppliers, analysis and manufacturing interests.

External Facing Website Established: h2new.energy.gov

Added 3 University participants (Carnegie Mellon, UC-Irvine, Colorado School of Mines), have widely coordinated with other research institutions and industry, expect academia and industry to engage further through FOAs and collaborative research.



Remaining Challenges/Proposed Future Work (LTE)

(Refer to P149a,b,c & d for more detail at Task Level)

Task 1 (Durability): Aging Studies; Mitigation Strategies; Ex-situ Characterization of MEAs/components/interfaces; Ex-situ Catalyst Durability; Ex-situ Membrane Durability; Accelerated Stress Test Development

Task 2 (Performance): Performance benchmarking, baselining, validation; Cell performance testing in support of electrode development; Ex-situ studies focused on performance factors; Cell level model development

Task 3 (Scale-up): MEA fabrication, Interface engineering (inks, electrodes, cell integration and interfaces); Components (porous transport layers, recombination layers)

Task 3c (Analysis): Performance, manufacturing, and system models; Durability factors; Energy system integration

Any proposed future work is subject to change based on funding levels

Summary: H2NEW LTE

Degradation and Accelerated Stress Test (ASTs)

- ✓ Probing of single stressor degradation pathways
- ✓ Rainforest stack for durability
- demo established for correlating with AST
- ✓ AST Working Group Established

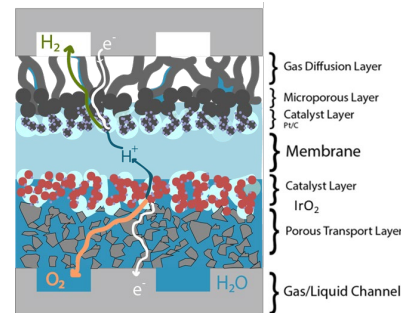
Task 1: Durability: Fundamental Understanding to ASTs to Mitigation

Operando cell and ex situ component studies

- ✓ Operando neutron diffraction cell commissioned
- ✓ Discretionary funding allocated to operando tomography
- ✓ Identical location microscopy and on-line ICMPs demonstrated for studying degradation

Mitigation Strategies

- ✓ To be informed by degradation and AST



Task 2: Performance: Structure-Property Relations to Optimized Performance

Baseline MEA

- ✓ Crosstalk team established common MEA, testing and prep
- ✓ Identified and sourced PTL

Cell Modeling

- ✓ Identified parameter space and experimental matrix that support the model needs of the consortium with regards to in-situ, ex-situ, and in-operando experiments

catalyst layers, porous transport layers

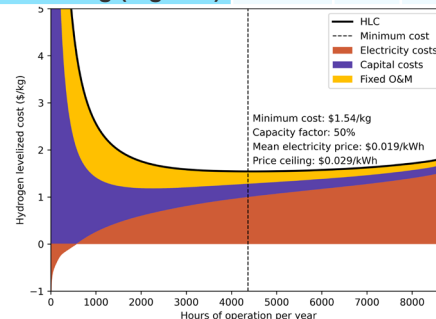
- ✓ Study of PTL structure property relations initiated
- ✓ Identified and sourced PTL
- ✓ Feasibility of PTL fabrication via 3-D printing and tape casting
- ✓ Demonstrated
- ✓ Concentrated ink baselining between 3 labs

Task 3: Component processing to Scale-up to Systems Analysis

TEA, Systems Analysis

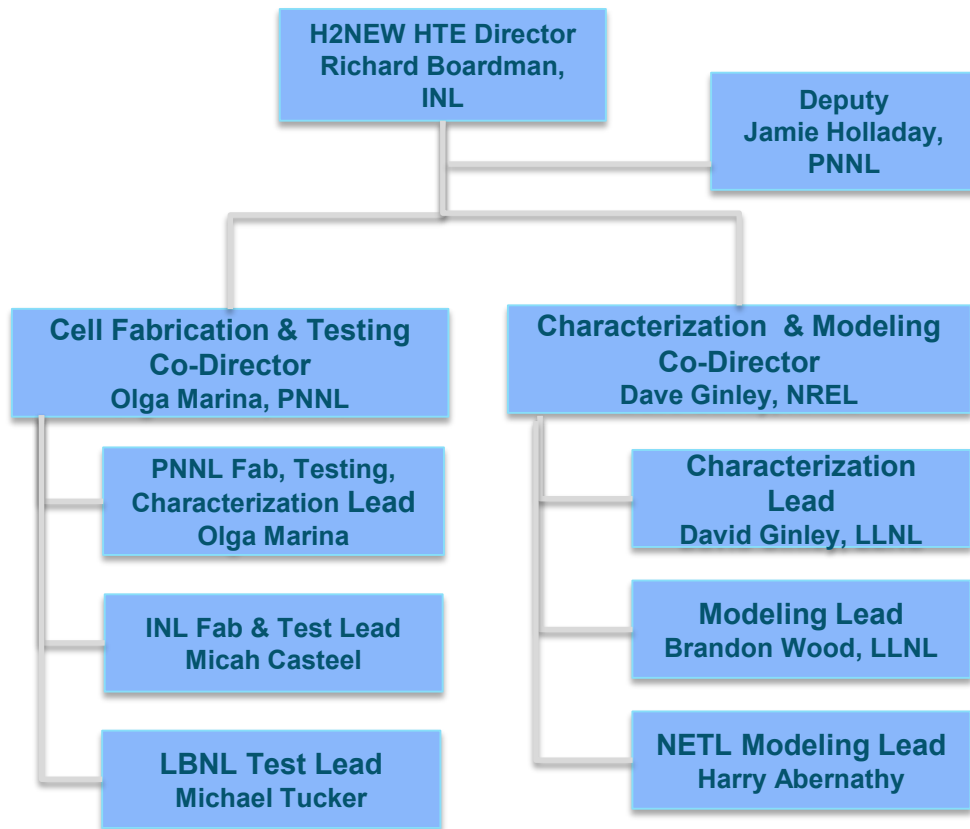
- ✓ Performance, manufacturing, and system analysis is underway and tools are being developed and implemented
- ✓ Identified Operating Strategy that Minimizes Cost for Several Electrolyzer Costs Identified and sourced PTL
- ✓ Initial Input to Stack Performance Targets
- ✓ Cell to system level performance model

| Stack Targets | Status | 2023 | 2025 |
|-----------------------------------|--------|------|------|
| Cell (A/cm ² @1.9V) | 2.0 | 2.5 | 3.0 |
| Efficiency (%) | 66 | 68 | 70 |
| Lifetime (khr) | 60 | 70 | 80 |
| Degradation (mV/khr) | 3.2 | 2.75 | 2.25 |
| Capital Cost (\$/kW) | 350 | 200 | 100 |
| PGM loading (mg/cm ²) | 4 | 1 | 0.5 |



H2NEW Activities: High Temperature Electrolysis (HTE)

Richard Boardman, H2NEW Deputy Director, and HTE Lead



TASKS

- Durability and Accelerated Stress Testing
 - a. Cell fabrication
 - b. AST methods development
 - c. Parametric matrix of test condition
- Task Integration and Review
 - a. Testing protocol
 - b. Inter-laboratory benchmarking
 - c. Stakeholder Advisory Board (SAB)
- Characterization
 - a. Microscopy and spectroscopy
 - b. Synchrotron measurements
 - c. In-operando measurements
- Multi-scale Degradation Modeling
 - a. Physics-based mechanisms and kinetics
 - b. Data-driven statistical analysis
 - c. Machine learning/Artificial Intelligence

❑ When strategically coupled with heat sources such as waste heat from nuclear or solar, **High Temperature [Steam] Electrolysis is a promising technology for achieving the Hydrogen Earth Shot goal of \$1 for 1 kg hydrogen in 1 decade.**

❑ Widespread commercialization of HTE is currently impeded by poor durability. Developing a fundamental understanding of all failure mechanisms is imperative and necessitates accelerated aging protocols that mimic the cell behavior of conventionally aged cells.

DOE HTE Targets

| <i>HTE Electrolyzer Stack Goals by 2025</i> | |
|---|------------------------------|
| <i>Capital Cost</i> | \$100/kW |
| <i>Electrical Efficiency (LHV)</i> | 98% at 1.5 A/cm ² |
| <i>Lifetime</i> | 60,000 hr |

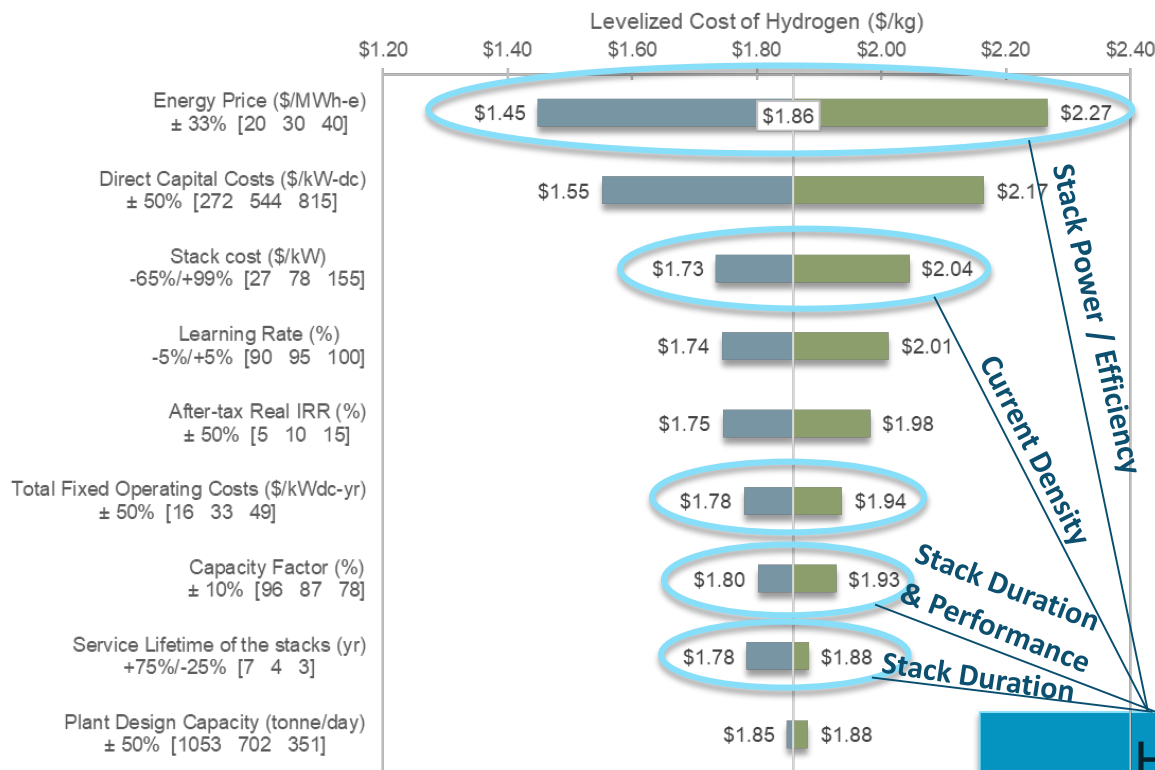
❑ **Successfully elucidating failure mechanisms (including validation of accelerated aging protocols) is contingent on robust characterization and modeling of HTE cells across multiple length scales from nanometer to millimeter.**

Quarterly Progress Milestones



| Milestone Description | | Summary of Criteria | |
|-----------------------|---|--|--|
| QP1 | Initial Round-Robin button cell testing under standard conditions for 500 hours after achieving steady-state | 20% Degradation rate standard deviation over at least 500 H operating time after steady-state has been achieved, in both intra-lab and interlab data sets. | |
| QP2 | Characterize cells prior to operation, during and following break-in in terms of performance degradation | Achieve steady state operations in < 500 hour; Measure performance, degradation and time-to-steady-state; Characterize virgin and harvested cells to evaluate break-in changes using SEM/EDS, TEM, XRD, XAS, and TXM | |
| QP3 | Complete button cell testing under standard operating conditions (1.3 V, 750°C), using aggressive steam concentrations (high and low) | Complete button cell testing at the thermoneutral voltage at 750°C with aggressive steam concentrations (high and low); evaluate aggressive steam as a stressor capable of accelerating degradation by > 20% | |
| QP3 | Complete parametric testing matrix of 2 to 3 kH long-duration experiments | Mean, steady-state degradation rate for 2 and 3 kH experiments within 3%/kH of the mean | |
| QP3 | Fabricate and test large planar cell degradation test and compare outcomes with button cells | Demonstrate similarities and differences of degradation mechanisms at relevant scales | |
| QP4 | Accelerated stress test protocol development and demonstration; aggressive steam as a stressor, comparing performance and degradation at 95%, 70%, 50% and 10%. | Demonstrate positive correlation between mean, steady-state degradation rates from one-, two- and three-segment, aggressive steam experiments, and steam value | |
| QP4 | Develop predictive model for one or more mechanisms of degradation | Compare model outcomes to test cell characterization results | |

Relevance/Potential Impact



□ <\$2/kg-H₂ is projected when:

- Nth-of-a-kind HTE project; 95% manufacturing learning rate
- 700 tonnes/day capacity (i.e. the capacity of a large nuclear power plant) selling power for \$30/MWh
- Stack cost of \$78/kW-dc; Assumes a current density of 1.5 A/cm²
- Stack service life of 4 years at 87% on-line capacity; Assumes a stack degradation rate of 0.86%/1000 operating hours
- Steam utilization of 80% per pass

**H2NEW Potential Impact:
Cost Reduction of \$0.5-0.75 /kg-H₂**

AMR Poster: Wendt et al. 2022, INL/RPT-22-66117

Known Mechanisms of SOEC Degradation

1. Cathode degradation

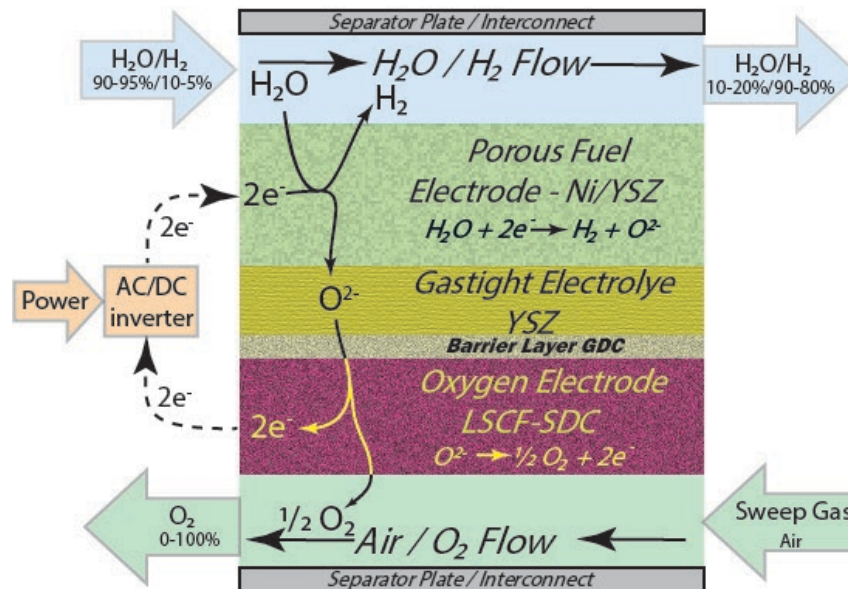
- Ni coarsening /sintering
- Ni oxidation
- Ni mobility

*Leads to reduced triple point dispersion and activity and delamination from electrolyte;
Results in decreased e^- conduction*

2. Dense electrolyte degradation

- YSZ composition changes
- Zr reduction/NiZr alloying

Results in decreased O^{2-} ion conduction



3. GDC barrier degradation

- Gd exsolution
- New Gd containing phases
- Coarsening
- Interfacial void formation

4. Anode degradation

- Interfacial void formation
- Sr reaction with Zr, SrZrO₃

Lead to increased pO_2 and delamination

High Temperature Electrolysis (HTE)

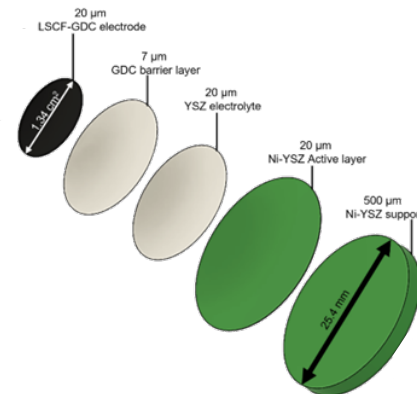
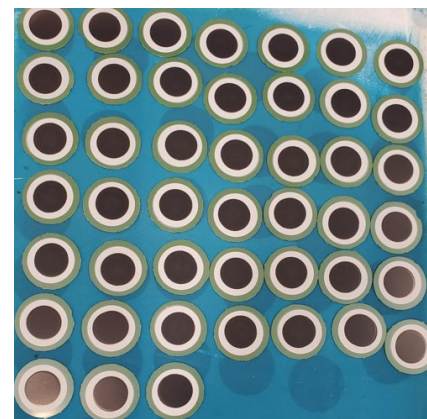
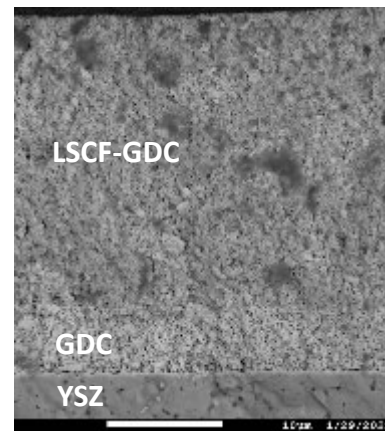
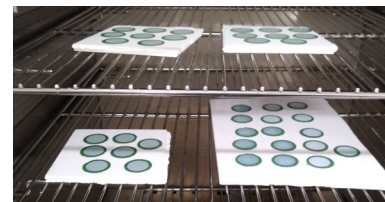
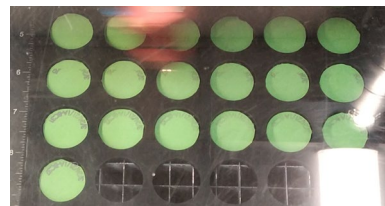
Task 5: Durability Testing and AST Development
(more content in AMR Poster [P196E, Olga Marina, et al.](#))

- Fabricate identical cells using well known state-of-the-art materials
- Define standard experimental conditions, including temperature, gas compositions, current/voltage, to obtaining baseline
- Initiative testing under standard operations to determine potential degradation phenomena
- Identify the number of test repeats to establish confidence
- Determine degradation mechanisms via extensive post-test characterization; compare findings to literature references for SOFC and SOEC
- Identify potential stressors for Accelerated Stress Testing (AST)
- Define AST matrix and conduct experiments using incrementally aggressive conditions
- Perform experimental validation and post-test characterization and compare results to those obtained under standard operating conditions
- Ensure active participation of HTE stakeholder advisory board that includes commercial cell developers and academia; seek feedback and guidance from experts in the field
- Develop AST Protocol

H2NEW Cell Fabrication (PNNL Leads)

Complete:

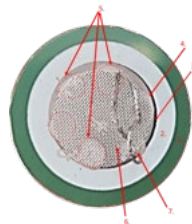
- Identified state-of-the-art cell designs and materials
 - Cells are Ni-YSZ electrode supported with the YSZ electrolyte, GDC barrier, and LSCF-GDC oxygen electrode
- Established a batch fabrication process of button cells to minimize the variance between the separate cells; developing QA/QC procedures
- Provided all labs with identical cells for testing and performance validation
- Optimized start-up, test and shut-down procedures between the labs for improved data comparison



Inter-Lab Standardization Testing Protocol

LSCF electrode paste used to mount platinum mesh current collectors on the oxygen electrode.

Steam/Hydrogen electrode uses nickel paste to mount nickel mesh.



Cell preparation standardization

- Current collectors & attachment
- Precious metal reduction

Cell testing standardization:

- Heat-up
- Reduction
- Compositions



Button-Cell
*2-4 cm²
active area*



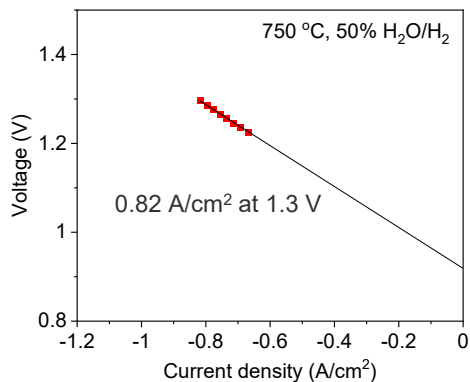
Enlarged Planar Cell
*10-25 cm²
active area*

Test Parameters

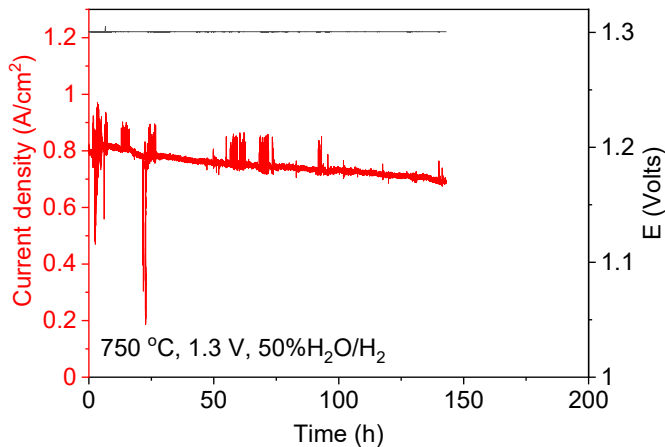
- Temperature
- Voltage & Current Density
- Steam/H₂ mixture flow and contact with cell

Accomplishment: INL example test results during cell break-in

Initial performance



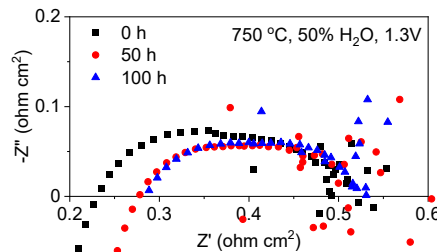
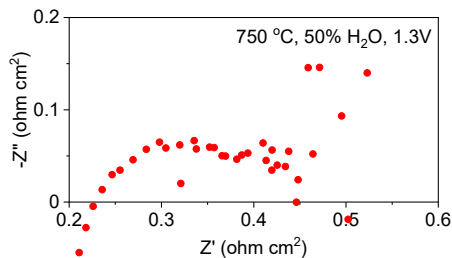
Long term test



PNNL Batch 3 cell #3

Current collectors:
 O₂ electrode: LSCF paste, Pt mesh, Pt wire
 H₂O electrode: Ni paste, Ni mesh, Pt wire

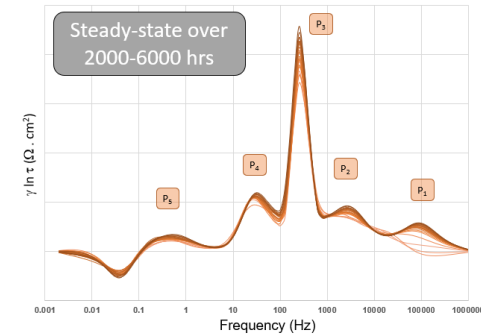
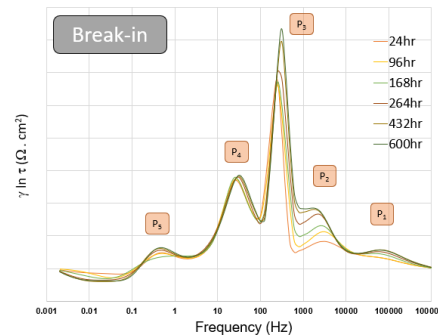
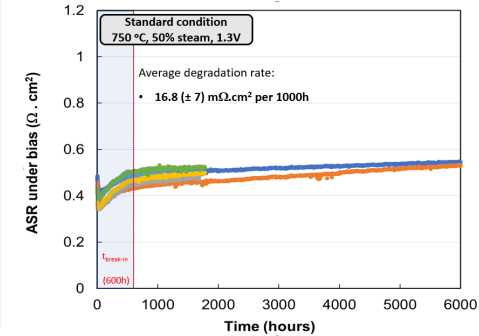
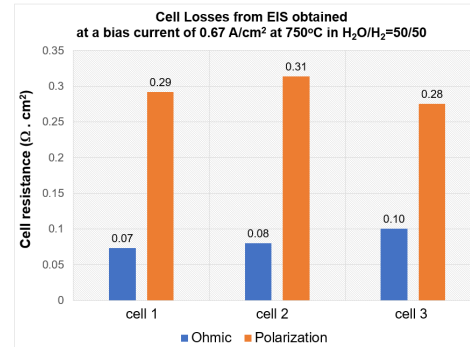
- Reduction for 22 h at 700 ° C
- OCV: 1.082 V (3% H₂O, 750 ° C), 0.92 V (50% H₂O, 750 ° C)



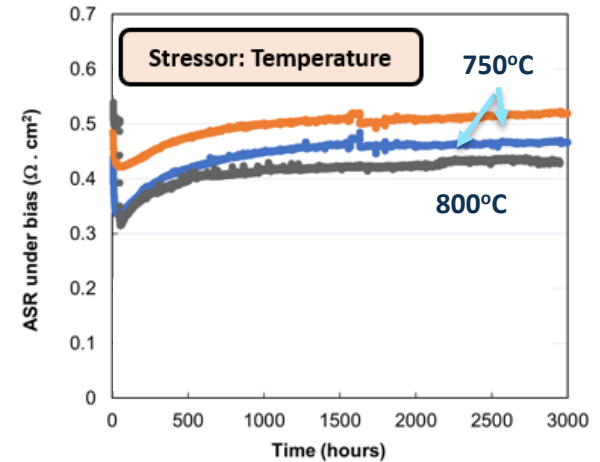
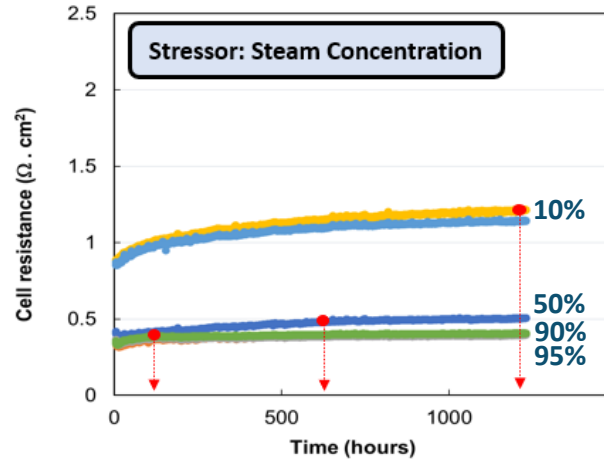
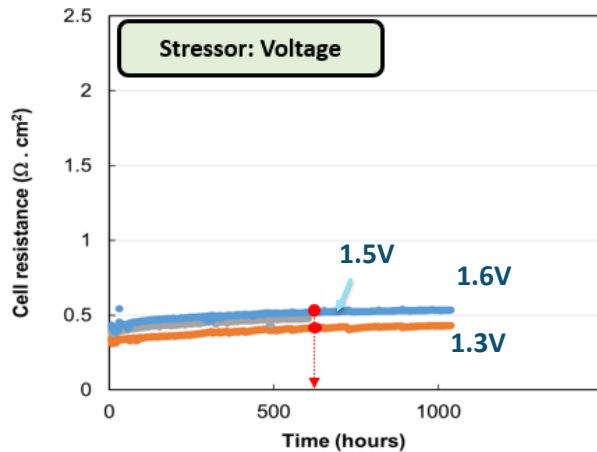
Accomplishment: PNNL Baseline Testing and Degradation Results

Complete:

- Obtained reasonable reproducibility cell performance data within the labs. Total cell losses are dominated by the electrodic resistance; ohmic losses are 1/3 of polarization losses
- Established baseline performance of multiple button cells under standard operating conditions (750°C, in 50% steam at 1.3 Volt) over 1000-6000 hours
- Two degradation periods are observed: fast degradation over the initial 600 hours (break-in) and a slow degradation during the following testing
- Average degradation rate is below 17 ohm·cm² per 1000 hours
- DRT peak-fit analysis attributes degradation during break-in period to two peaks (at 1kHz – 10kHz and 100 Hz – 1kHz) associated with diffusion phenomena in the oxygen electrode.



Accomplishment: PNNL Accelerated Stress Testing for Protocol Development



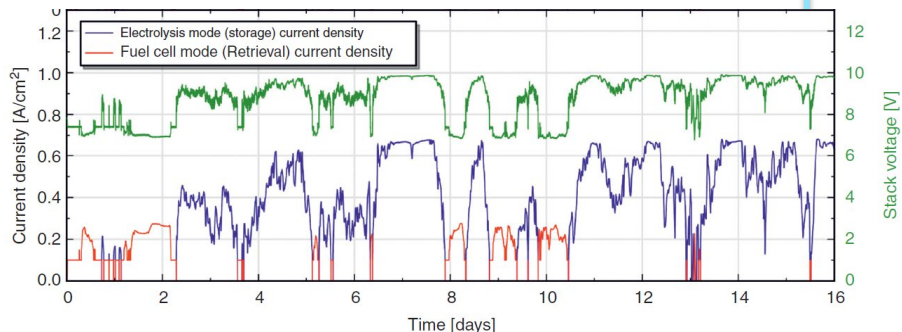
In Progress:

- To develop AST protocol as compared to standard operating conditions, the following stressors have been applied: higher temperature (800°C), higher voltage (1.5 and 1.6 V), higher steam (90 and 95%), and lower steam (10%)
- Higher voltage and higher temperature allows to shorten the break-in periods
- Higher steam concentration has no impact on break-in time and degradation rates

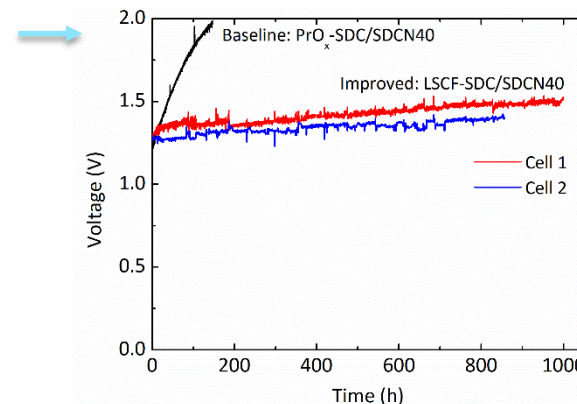
Accomplishment: LBNL Dynamic Cycling for Protocol Development

R&D SOEC testing typically at fixed conditions (V, I, T, P...)

BUT: anticipated operation is highly dynamic



C. Graves, J.V.T. Høgh, K. Agersted, X. Sun, M. Chen et al., July 2014, Department of Energy Conversion & Storage, Technical University of Denmark



Metal-supported SOEC, air/50% humidified hydrogen
F. Shen, R. Wang, M.C. Tucker, Journal of Power Sources, 474 (2020) 228618

Complete:

- Literature survey, discussion with advisory board complete
- Highest priority cycling variables: H₂/H₂O ratio, and voltage

Ongoing:

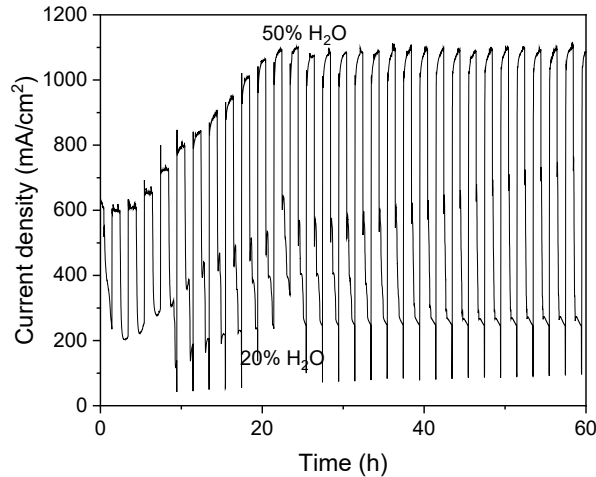
- Test standard cells under dynamic conditions
- Deliver cells to Task 7 for analysis
- Assess viability as an Accelerated Stress Test

| Dynamic Variable | Votes from industry |
|------------------|---------------------|
| Steam ratio | 5 |
| Power (I,V) | 4 |
| Temperature | 1 |
| Redox | 1 |
| Pressure | 0 |

Accomplishments: LBNL Dynamic Cycling Button Cell Test Results

Steam:Hydrogen Ratio Cycling

750°C, 1.3 V



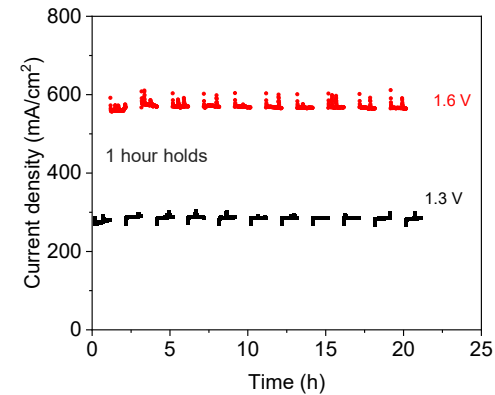
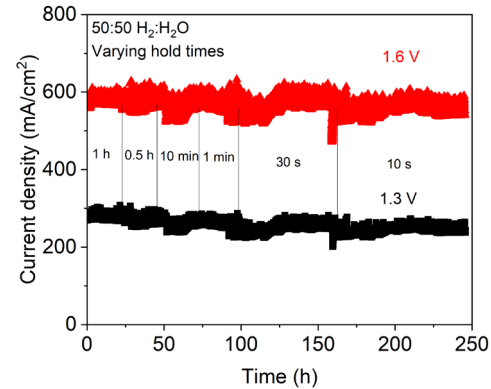
Rapid switching between steam content limits

Upcoming:

- Wider steam content window
- Long-term cycling at fixed steam content limits

Voltage Cycling

750°C, 50:50 H₂O:H₂



Stable cycling 1.3 to 1.6 V

- 1 h to 10 sec half-cycles

Upcoming:

- Wider voltage window
- Long-term cycling at fixed frequency

High Temperature Electrolysis (HTE)

Task 7: Advanced Characterization
(more content in AMR Poster P196e and p196f)

Overview: Classical and New Characterization methods for SOECs

Complexity of the cell components and interfaces in SOECs necessitate **multi-technique approach** to understand failure mechanisms over a range of aging conditions

Transmission X-Ray Microscopy (TXM)

30 nm Resolution

- Changes in pore structure
- Layer delamination
- Cation migration
- Identification of secondary phase formation

X-Ray Absorption Spectroscopy Imaging

1 μ m Resolution

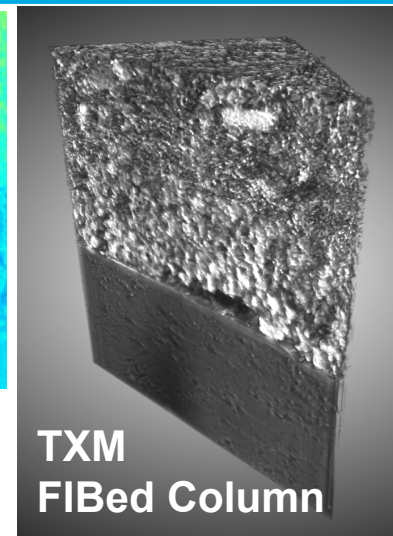
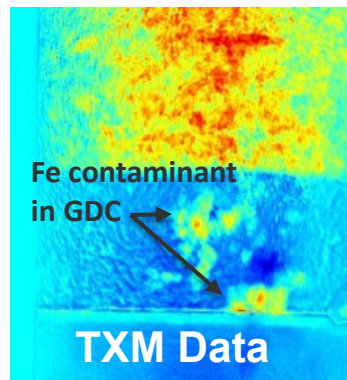
Parts per million elemental sensitivity

- Cation migration
- Identification of secondary phase formation

X-Ray Diffraction

Atomic sensitivity to crystal structures

- Secondary phase formation
 - Identification of contaminants
- In-operando experiments**



SEM/EDS

0.1 μ m Resolution

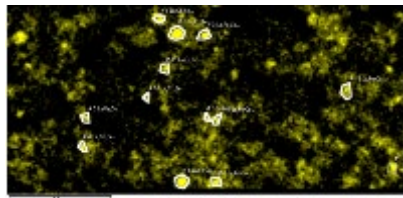
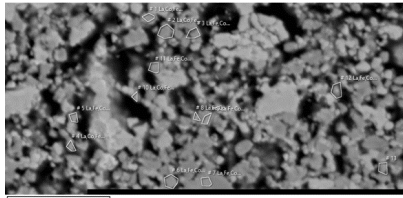
Parts per hundred elemental sensitivity

- Cation migration
- Large scale structural changes

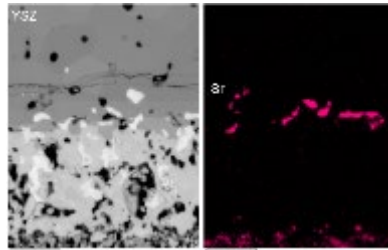
Robust characterization over multiple length-scales is critical

Changes in Oxygen Electrode after 3000 hrs tests at 1.3V

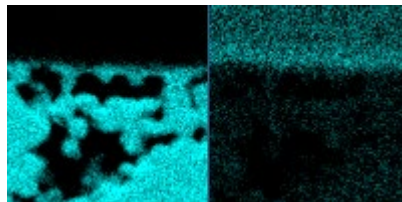
Co variance



Sr migration

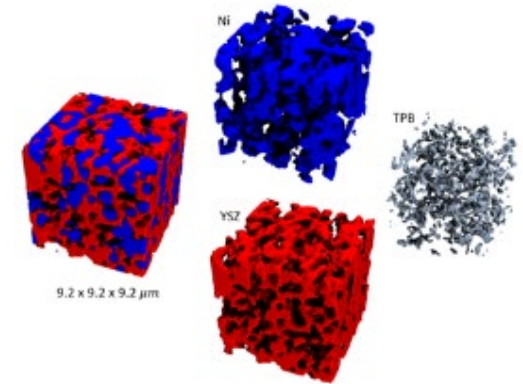
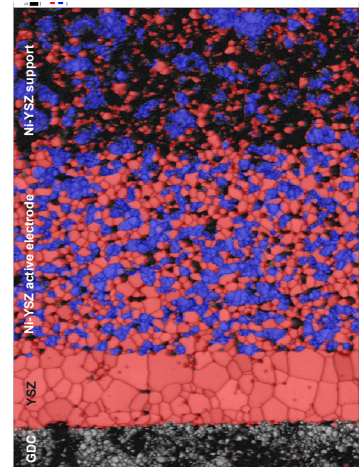


Gd enrichment at the interface



No Significant Changes in Hydrogen Electrode after 3000 hrs in 90% steam

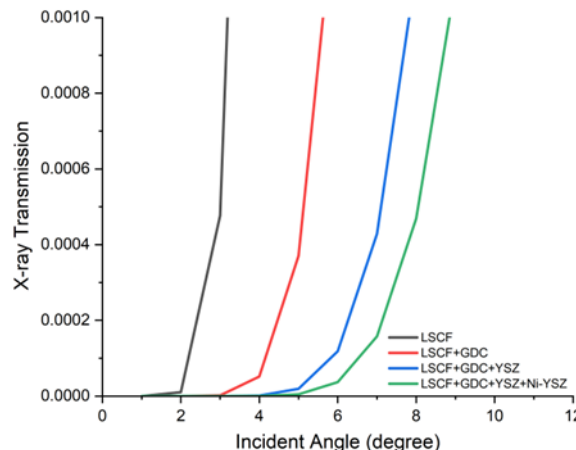
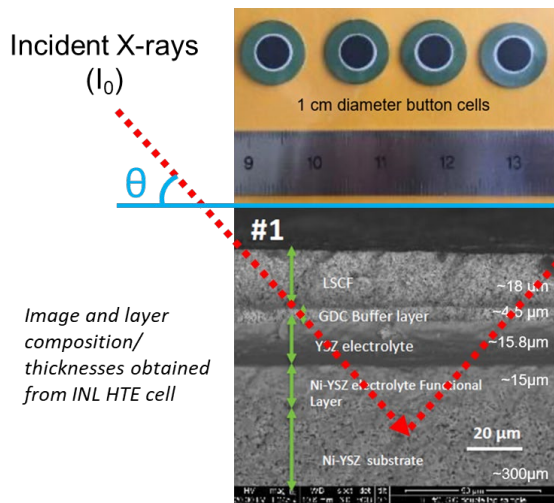
No Ni coarsening or migration



Triple-phase boundary determined using PNNL modeling tools

Accomplishment: Prediction of X-ray Attenuation through HTE Cells depth profiling as run button cells with synchrotron x-rays (by NREL)

First steps towards designing *in operando* HTE X-ray scattering measurements



The X-ray penetration depth into the HTE cell can be intentionally limited by the incident angle, θ (left) and energy (below)

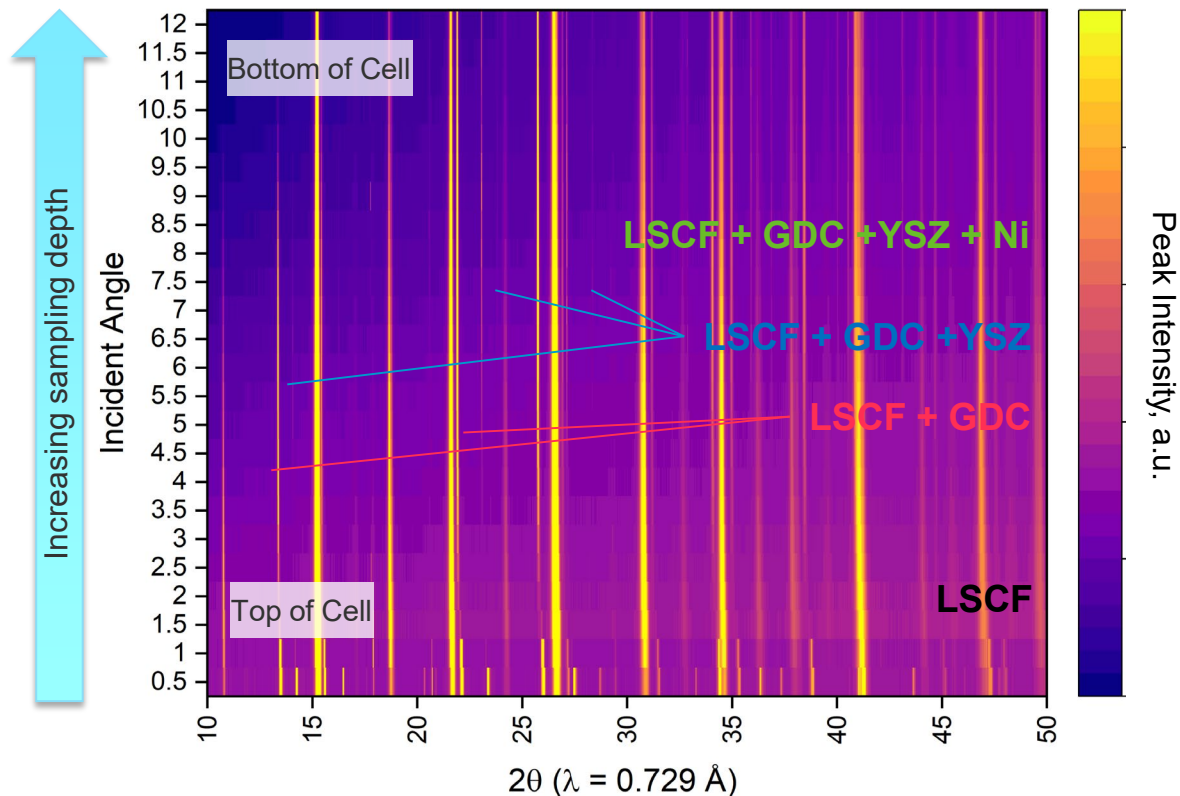
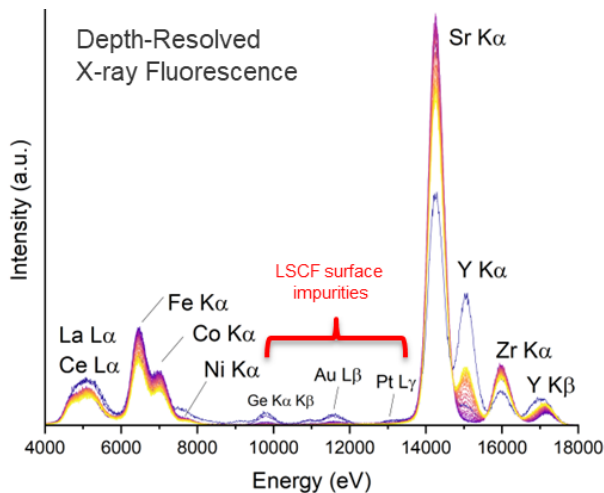
| Material Composition | Crystalline Density (g/cc) | Apparent Density (g/cc)* | Nominal Layer Thickness (μm) | (SSRL BL2-1) | (APS 11-BM) | (APS 11-ID-B) Si(311) | (APS 11-ID-B) Si(422) |
|---|----------------------------|--------------------------|---|------------------|------------------|-----------------------|-----------------------|
| | | | | I/I_0 @ 16 keV | I/I_0 @ 35 keV | I/I_0 @ 58.66 keV | I/I_0 @ 86.7 keV |
| $\text{La}_{0.6}\text{Sr}_{0.4}\text{Co}_{0.2}\text{Fe}_{0.8}\text{O}_3$ (LSCF) | 6.33 | 3.165 | 18 | 0.82 | 0.9638 | 0.975 | 0.99 |
| $\text{Ce}_{0.9}\text{Gd}_{0.1}\text{O}_{1.95}$ (GDC) | 6.98 | 6.98 | 4.5 | 0.85 | 0.98 | 0.974 | 0.99 |
| $\text{Zr}_{0.8}\text{Y}_{0.2}\text{O}_{1.9}$ (YSZ electrolyte) | 5.9 | 5.9 | 15.8 | 0.87 | 0.98 | 0.973 | 0.99 |
| Ni metal | 8.91 | - | - | 0.825 | 0.98 | 0.995 | 0.998 |
| Ni(50% vol)-YSZ(50% vol) (Ni-YSZ) | 7.405 | 3.7025 | 15 | 0.97 | 0.975 | 0.994 | 0.998 |
| | | | I/I_0 to bottom of cell | 0.485264 | 0.884443 | 0.913873 | 0.966422 |
| | | | I/I_0 total out of cell | 0.235481 | 0.782239 | 0.835164 | 0.933971 |

Note: these calculations assume an incident angle of 90°

*Density reduced by particle packing

Accomplishment: Depth Resolved XRD of Intact Button Cell by NREL

Validated X-ray attenuation predictions and **demonstrated** XRD can be used to resolve crystal structures and defects from individual layers of intact cells (XRD, right) with simultaneous compositional analysis (XRF, below)

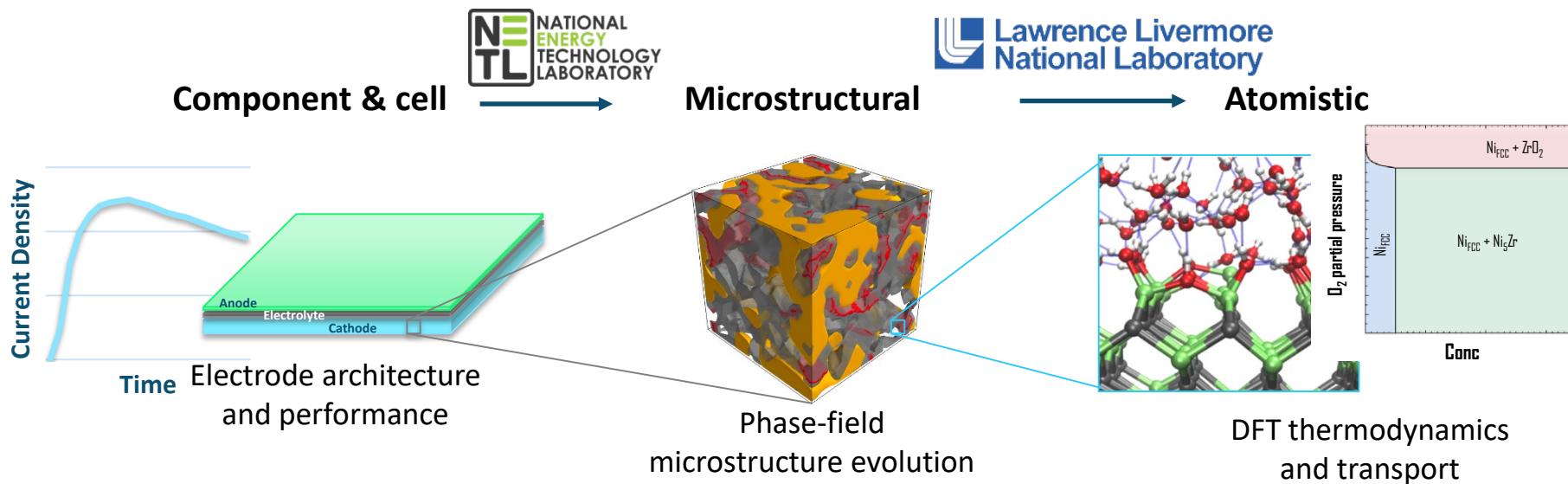


High Temperature Electrolysis (HTE)

Task 8: Multiscale Modeling
(more content in AMR Poster [P196g](#))

LLNL/NETL Multiscale Modeling of degradation modes

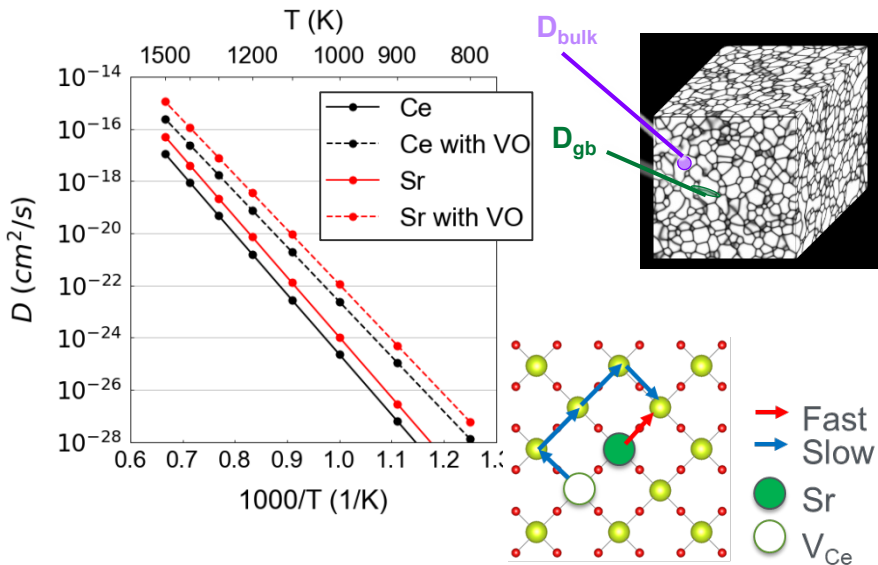
Physics-based models developed at LLNL and NETL are being linked across scales and accelerated using data science techniques to connect properties to performance loss



Linkage between processes at different scales is crucial to understanding performance degradation and designing effective AST protocols

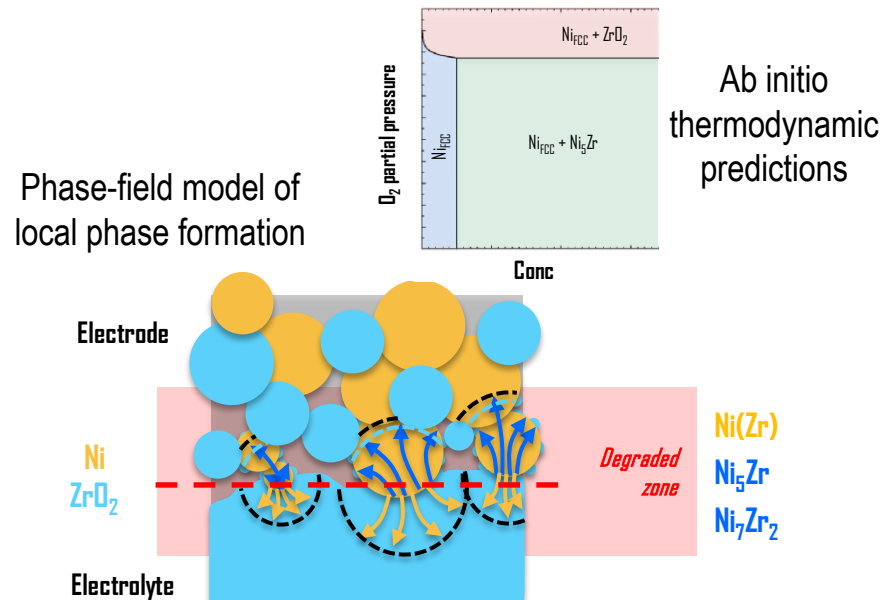
Models to Interrogate Proposed Degradation Mechanisms

Atomistic: Sr diffusion through interlayer



Early studies on Sr interdiffusion confirm that Sr cannot penetrate grains but could use grain boundaries as a migration mechanism

Mesoscale: N-iZr intermetallics at Ni-YSZ interface



We are exploring machine learning to accelerate simulations and implementing protocols to combine with NETL SOEC database to test performance under varied conditions

- A highly coordinated, interactive four-prong strategy – leveraging
- leading expertise, capability at the National Labs
- Stakeholder Advisory Board
 - **Scott Swartz**, Nexceris
 - **Greg Tao**, Chemtronergy
 - **John Pietras**, St. Gobain
 - **Peter Voorhees**, Northwestern
 - **Scott Barnett**, Northwestern
 - **Xiao-Dong Zhou**, University of Louisiana-Lafayette

| | |
|--|-----------------------|
| Task 5. Cell fabrication, operation, AST development | PNNL, INL, LBNL |
| Task 6. Task integration and protocol validation | INL & All Labs, SAB |
| Task 7. Advanced characterization | NREL, PNNL, ANL, SLAC |
| Task 8. Multiscale modeling | LLNL, NETL, PNNL |

Select group of advisors representing OEMs, Tier 1 suppliers, analysis and manufacturing interests

Synergy with DOE-Nuclear Energy

- *Light-Water Reactor Sustainability*
- *Integrated Energy Systems*

Leverages other Consortium

- HydroGEN 2.0 (HFTO)
- R2R (AMO)
- Million Mile Fuel Cell Truck (HFTO)
- ElectroCAT (HFTO)

Remaining Challenges and Barriers (HTE)

(Refer to P196e,f & g for more detail at Task Level)

Task 5: Durability Testing and AST Development

- i. SOEC materials and component degradation mechanisms are the result of multiple, coupled phenomena derived from operating conditions and a comprehensive and accurate understanding of the interplay of these phenomena remains to be established
- ii. More testing is needed to understand how to cause relative accelerated stress outcomes
- iii. Larger planar cell test stand capacity is needed to fully understand cell degradation phenomena

Task 6: Task Integration and Protocol Validation

- i. H2NEW is generating large volumes of data that require review, validation, and interpretation

Task 7: Advanced Characterization

- i. In-operando testing is needed to reduce the number of test samples and to expedite AST outcomes
- ii. Implementing HTE experiments operating at beam line user stations is not easy nor inexpensive

Task 8: Multiscale Modeling

- i. Defining and modeling the mechanisms requires the combined talents, experience and capabilities of each of the Lab Consortium
- ii. Degradation processes at the YSZ|GDC|LSCF-SDC interface are probably coupled in a non-linear fashion

(Refer to P196e,f & g for more detail at Task Level)

Task 5: Durability Testing and AST Development

- i. Continue development of AST protocols that accelerate degradation with fidelity to mechanisms functioning at standard operating conditions
- ii. Mature & deploy augmented testing capability sufficient to cover the parameter matrix and requisite

Task 6: Task Integration and Protocol Validation

- i. Ensure refined protocols are validated, widely available
- ii. Organize data volume and surmount archival,/curation challenges

Task 7: Advanced Characterization

- i. Refine and implement in situ / in operando analyses at APS, SLAC
- ii. Coordinate interface between testing, modeling and characterization

Task 8: Multiscale Modeling

- i. Ramp up modeling and simulation efforts with increased funding to LLBL, NETL and PNNL to project degradation phenomena exhibited by AST testing

Any proposed future work is subject to change based on funding levels

- The H2NEW HTE hypothesis is that a systematic, coordinated research program targeting the coupled degradation phenomena will yield refinements to composition, fabrication and operation that will enable HTE technology to overcome current durability barriers
- H2NEW HTE is making progress in developing accelerated stress testing protocol
 - Standard electrode-supported cell has been selected
 - PNNL is fabrication and supply the Consortium the standard cells
 - Inter-Laboratory cell testing is underway, and is closing on consistent test outcomes
 - Initial data from PNNL, and also from LBNL and INL have set a foundation for to establish AST protocol
- Cell characterization using standard microscopy is underway using the existing Consortium state-of-the-art X-ray and electron transmission microscopy.
- NREL has *validated* X-ray attenuation predictions and *demonstrated* XRD can be used to resolve crystal structures and defects from individual layers of intact cells (XRD, right) with simultaneous compositional analysis (XRF, below)
- The SAB has provided detailed feedback; a 3 hours seminar of Northwestern University R&D for the past decade
- The HTE Consortium will meet this summer to review progress before ramping up modeling and simulation

Technical Backup and Additional Information Slides

- Alex Badgett, Mark Ruth, Bryan Pivovar, “Economic considerations for hydrogen production with a focus on polymer electrolyte membrane electrolysis,” *Electrochemical Power Sources: Fundamentals, Systems, and Applications*, 2022, 327-364. <https://doi.org/10.1016/B978-0-12-819424-9.00005-7>
- Bryan S Pivovar, Mark F Ruth, Akihiro Nakano, Hirohide Furutani, Christopher Hebling, Tom Smolinka, “Getting Hydrogen to the Gigaton Scale,” *Electrochem. Soc. Interface*, 30, 85, 2021. DOI: 10.1149.2/2.F19214F
- Bryan S Pivovar, Mark F Ruth, Deborah J Myers, Huyen N Dinh, “Hydrogen: Targeting \$1/kg in 1 Decade,” *Electrochem. Soc. Interface*, 30, 6, 2021.
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- K Ayers, N Danilovic, K Harrison, H Xu, “PEM Electrolysis, a Forerunner for Clean Hydrogen,” *The Electrochemical Society Interface* 30 (4), 67, 2021.
- A Kusoglu, “Chalkboard 1-The Many Colors of Hydrogen,” *The Electrochemical Society Interface* 30 (4), 44, 2021.
- A Badgett, M Ruth, B James, B Pivovar, “Methods identifying cost reduction potential for water electrolysis systems,” *Current Opinion in Chemical Engineering* 33, 100714, 2021.
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- Zhenye Kang, Yingying Chen, Hao Wang, Shaun M Alia, Bryan S Pivovar, Guido Bender, “Discovering and Demonstrating a Novel High-Performing 2D-Patterned Electrode for Proton-Exchange Membrane Water Electrolysis Devices,” *ACS Appl. Mater. Interfaces* 2022, 14, 1, 2335–2342. <https://doi.org/10.1021/acsami.1c20525>
- Xiong Peng, Pongsarun Satjaritanun, Zachary Taie, Luke Wiles, Alex Keane, Christopher Capuano, Iryna V. Zenyuk, Nemanja Danilovic, “Insights into Interfacial and Bulk Transport Phenomena Affecting Proton Exchange Membrane Water Electrolyzer Performance at Ultra-Low Iridium Loadings,” *Adv. Sci.* 2021, 8, 2102950. <https://doi.org/10.1002/adv.202102950>
- M.R. Gerhardt, L.M. Pant, J.C.M. Bui, A. R. Crothers, V.M. Ehlinger, J.C. Fornaciari, J. Liu and A.Z. Weber, “Methods—Practices and Pitfalls in Voltage Breakdown Analysis of Electrochemical Energy-Conversion Systems”, *J. Electrochem. Soc.* 168, 074503, 2021. DOI:10.1149/1945-7111/abf061.
- Boardman, R.D., High Temperature Steam Electrolysis, *Encyclopedia of Nuclear Energy*, Volume 3 <https://doi.org/10.1016/B9780128197257.002026>
- Wendt, D.S., Knighton, L.T., Boardman, R.D., “High Temperature Steam Electrolysis Process Performance and Cost Estimates,” *IN L/R PT* 2266117
- Boardman, R.D., Bragg Sittion, S.M., Otgonbaatar, U., “Developing a low cost renewable supply of hydrogen with high temperature electrolysis,” *MRS Bulletin*, Vol. 47, March 2022, mrs.org/bulleting
- Bao, J., Naveen, K., Rechnagle, K., Wang, C., Koeppel, B., Marina, O., “Modeling Framework to Analyze Performance and Structural Reliability of Solid Oxide Electrolysis Cells,” *J. Electrochem. Soc.*, in press <https://doi.org/10.1149/1945-7111/ac6f87>
- Shen, F., Wang, R., Tucker, M.C., “Long term durability test and postmortem for metal support solid oxide electrolysis cells, *J. of Power Sources*, Vol, 474, 31 October, 2020, 228618