



# Project Goal

- Conduct technoeconomic analysis to evaluate the cost to produce H<sub>2</sub> (\$/kg) through various technological production pathways (i.e., electrolysis, PEC, others) using
  - Design for Manufacture and Assembly (DFMA) capital cost estimation techniques,
  - heat & mass balances, and
  - H2 Analysis (H2A) discounted cash flow models.
- Estimate the cost of H<sub>2</sub> based on state-of-the-art technology at distributed and central production facilities (1.5-50 tons per day) and measure the cost impact of technological improvements in H<sub>2</sub> production technologies.
- Evaluate the cost drivers and recommend to DOE the technical areas needing improvement for each technology.

# Overview

## Timeline

- Project start date: 10/1/2021
- Project end date: 9/30/2024
- Percent complete: ~20% of project

## Budget

- Total Funding Spent
  - ~\$121K SA (though Mar 2022)
- Total DOE Project Value:
  - ~\$775k SA
- Cost Share Percentage: 0%  
(not required for analysis projects)

## Barriers

- Hydrogen (H<sub>2</sub>) Generation by Water Electrolysis
  - F: Capital Cost
  - G: System Efficiency and Electricity Cost
  - K: Manufacturing

## Partners

- National Renewable Energy Laboratory (NREL)
- Idaho National Laboratory (INL)



## Collaborators (unpaid)

- 4 Electrolyzer companies and research groups

# Relevance and Impact

- Investigates production and delivery pathways selected/suggested by DOE that are relevant, timely, and of value to HFCTO.
- Supports selection of portfolio priorities through evaluations of technical progress and hydrogen cost status.
- Provides complete pathway definition, performance, and economic analysis not elsewhere available.
- Provides analysis that is transparent, detailed, and made publicly available to the technical community.
- Results of analysis:
  - Identifies cost drivers
  - Assesses technology status
  - Provides information to DOE to help guide R&D direction

# Progress Toward DOE Targets or Milestones

On-track to achieve 2025 DOE electrolysis targets at an electricity price of \$0.03/kWh  
 Continued improvements needed to achieve \$1/kg H<sub>2</sub> by 2031

Levelized Cost of H2 Production (\$/kg)	SA H2A Current Status (\$0.03/kWh electricity)	SA H2A Future Status (\$0.03/kWh electricity)	DOE 2025 Target	DOE 2031/Ultimate Target
Distributed Water Electrolysis Cost (1.5 Tons/Day)	2.54 (2019 PEM Tech) (2019 Record) 3.76 (2025 AEM Pure Water Tech) 2.59 (2025 AEM KOH Tech)	1.92 (2035 PEM Tech) (2019 Record) 2.18 (2035 AEM Pure Water Tech) 2.02 (2035 AEM KOH Tech)	2.30	1
Central Water Electrolysis Cost (50 Tons/Day)	2.31 (2019 PEM Tech) (2019 Record) 2.36 (2019 SOEC Tech) (2020 Record) 2.41 (Alkaline – optimized oper. cond.)	1.86 (2035 PEM Tech) (2019 Record) 2.00 (2035 SOEC Tech) (2020 Record) 1.79 (Alkaline – optimized oper. cond.)	2.00	1
Solar Thermochemical (STCH) (100 Tons/day)	NA	2.54 (2022 Journal Article, NREL)	3.70	1

- All electrolysis H2A status values assume **\$0.03/kWh** in current and future
- Although the cost of H<sub>2</sub> for liquid alkaline water electrolysis is preliminarily estimated to be higher cost than PEM for the current case and lower cost than PEM for the future case, **the PEM 2019 case was not based on optimized operating conditions at BOL and EOL. Re-evaluation at optimized conditions for PEM are planned for future work.**

# Project Objective and Approach

## Project Objective

- Support HFCTO in their selection of portfolio priorities by evaluating technical progress of H<sub>2</sub> production pathways
- Assess the potential to meet H<sub>2</sub> production cost targets (H2 Shot: \$1/kg of H<sub>2</sub> by 2031)
- Determine the most optimal production pathway for specific end-uses
- Evaluate the uncertainty and show the potential for H<sub>2</sub> cost reduction for each pathway through single and multi-variable sensitivity analyses
- Perform rigorous review of system design and assumptions, confirm the validity of assumptions with experts external to the project, and document results in reports and presentations

## Approach

- Select H<sub>2</sub> production pathway to evaluate (in collaboration with DOE)
- Collect data from industry/researchers, assess data for consensus and trends
- Conduct system modeling (Design for Manufacture and Assembly (DFMA) bottom-up cost modeling, Mass/Energy Balance modeling w/ Aspen Hysys)
- Evaluate the cost of H<sub>2</sub> using the H2A tool
- Document in public reports: detailed, transparent statement of assumptions and cost results

Task	Description	Completed for 2022 Analysis?
1	Technologies Identification, Review, and Selection of Pathway	Milestone 1.1 submitted in October 2021
2	System Definition and Bill of Materials	Milestone 2.1 submitted in March 2022
3	Techno-economic Analysis	In Progress: Milestone 3.1 to be submitted in June 2022
4	Case Study Documentation and Project Reporting	Planned: Milestone 4.1 to be submitted in September 2022 (Go/No-Go decision metric)

# Selection of H<sub>2</sub> Production & Delivery Cases

- DOE selects cases that support the HFCTO development mission
  - Advanced Water Splitting
  - Biomass-based processes
  - Waste recovery to H<sub>2</sub> processes

- Cases selected based on:
  - Highest priority cases with direct application to HFCTO mission
  - Data availability
  - Ability to assist studies in providing relevant cost estimates
    - Beneficial for cases without cost estimates
    - Provide assistance for proper development of H2A cases

## Types of H<sub>2</sub> Production Cases:

- 1,500 kg H<sub>2</sub>/day distributed sites
- 50,000 kg H<sub>2</sub>/day central production sites (sometime larger systems)
- Typically two technology levels analyzed
  - Current or Near Term: current technology at high-manufacturing rate
  - Future: future technology at high-manufacturing rate

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## Cases Currently Under Development

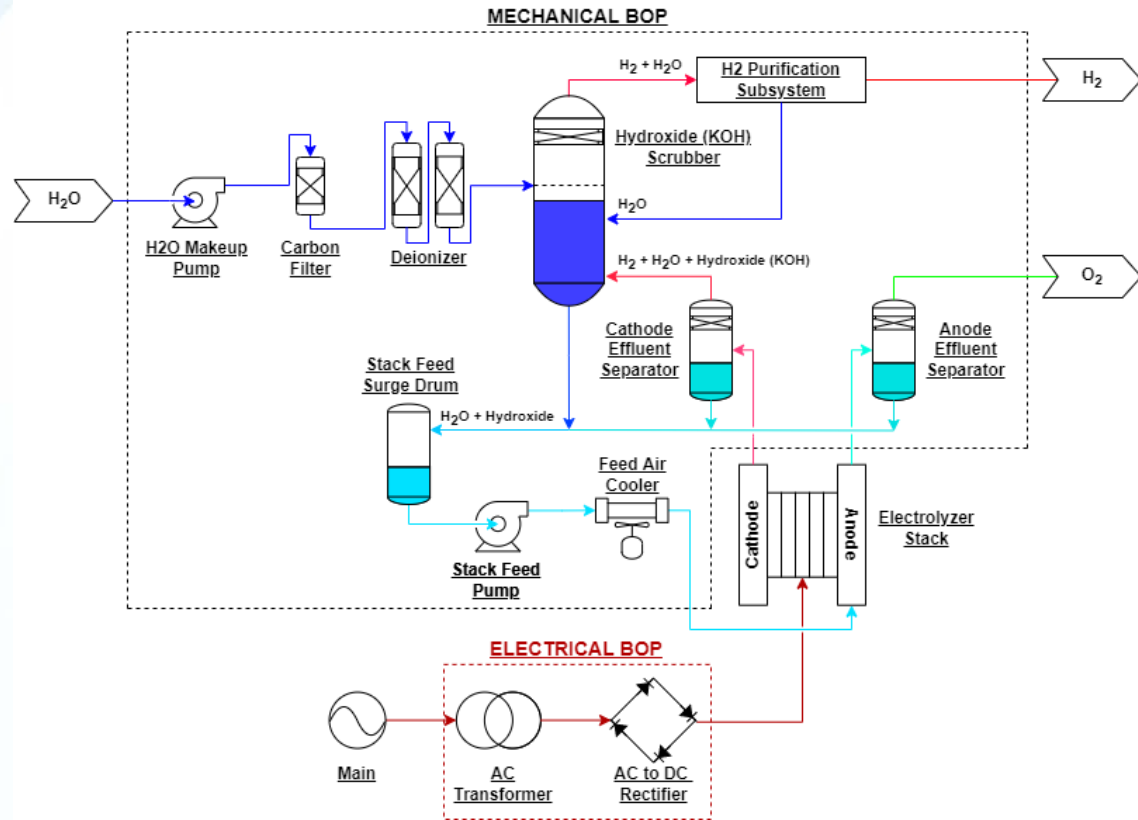
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- Liquid Alkaline Water Electrolysis (LAWE)
  - Current and Future Central Cases
- Anion Exchange Membrane (AEM) electrolysis
  - Near Term and Future Distributed Cases for pure water and KOH/water feeds

# Liquid Alkaline Water Electrolysis (LAWE) System Definition

## Process Flow Diagram

ALKALINE WATER ELECTROLYSIS (AWE) SYSTEM



Strategic Analysis 2022

## Table of System Design Parameters

		SA 2022 Cases	
	Unit	2021	2031
Plant Capacity (Rated)	MW_AC	118	97
Module Power	MW_AC	30	49
Rated Stack Input Power	MW_DC	3.9	3.8
Number of Stacks	#	7	12
# of Cells	#	450	450
Cell Area	cm <sup>2</sup>	10,000	10,000
<b>Rated Current Density</b>	<b>A/cm<sup>2</sup></b>	<b>0.7</b>	<b>1</b>
<b>Rated Cell Voltage (BoL)</b>	<b>V</b>	<b>2</b>	<b>1.7</b>
Degradation Rate	mV/1000 hrs	3.2	1.4
Pressure Cathode	bar	31	31
Temperature	degC	80	80
<b>Specific Energy Demand (Stack)</b>	<b>kWh_AC/kg</b>	<b>48.9</b>	<b>42.5</b>
<b>Specific Energy Demand (System)</b>	<b>kWh_AC/kg</b>	<b>53.7</b>	<b>44.9</b>
KOH Concentration	wt%	40	40
Output Pressure	bar	30	30
Gas Purity	%	99.99	99.99
Stack Lifetime	year	10	10



## Current Design

O<sub>2</sub> Gasket Seal: 1mm thick die cut EPDM sheets

O<sub>2</sub> Cell Frame: 2mm thick injection molded PEEK

O<sub>2</sub> Gasket Seal: 1.25mm thick die cut EPDM sheets

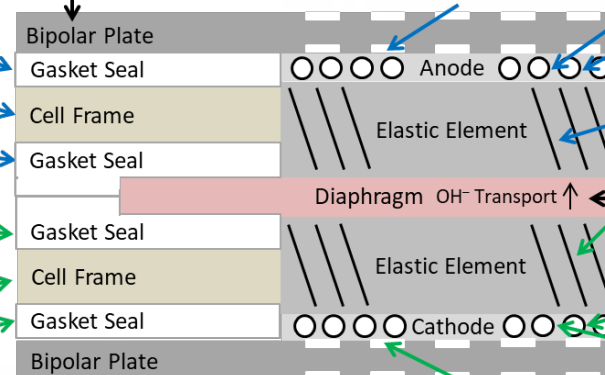
H<sub>2</sub> Gasket Seal: 1.25mm thick die cut EPDM sheets

H<sub>2</sub> Cell Frame: 2mm thick injection molded PEEK

H<sub>2</sub> Gasket Seal: 1mm thick die cut EPDM sheets

Ni BPP: 0.762mm thickness (0.2mm etched flow fields depth)

O<sub>2</sub> flow channel depth: 0.2mm



H<sub>2</sub> flow channel depth: 0.2mm (modeled as 0.2mm depth but likely to be deeper) 2<sup>nd</sup> BPP shown for channel depth. Only one BPP per cell.

### Water and O<sub>2</sub> Evolution

Perforated Ni plate anode substrate: 0.5mm thickness, 700 micron hole diameter

NiFe(OH)<sub>2</sub> catalyst: 7 mg/cm<sup>2</sup>

Ni wire elastic element: 3.5mm compressed thickness, 7mm uncompressed thickness, 0.15mm wire diameter

Zirfon (ZrO<sub>2</sub>/polysulfone) diaphragm: 500 micron thickness

Perforated Ni plate cathode substrate: 0.5mm thickness, 700 micron hole diameter

NiMo catalyst: 7 mg/cm<sup>2</sup>

### H<sub>2</sub> Evolution

## Changes between Current and Future Designs

- Finite Gap for Current and Zero Gap for Future
- Reduction of diaphragm thickness (500 to 220microns)
- Remove elastic elements and reduce number of components in the future design
- Anode/Cathode switch from perforated Ni plates to sintered porous Ni in future
- Bipolar Separator Plate switch from nickel plates to Ni-coated SS plates in future

## Future Design

O<sub>2</sub> Gasket Seal: 50µm thick die cut EPDM sheets

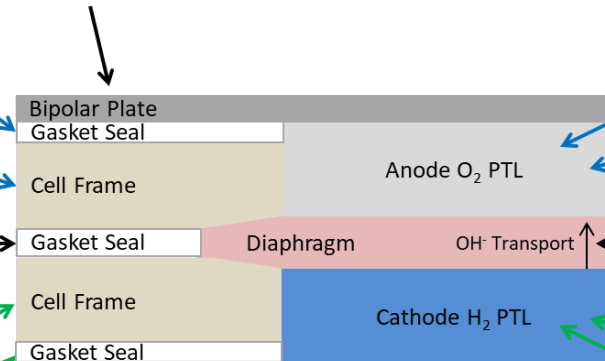
O<sub>2</sub> Cell Frame: 1.605mm thick injection molded PEEK

Gasket Seal: 110µm thick die cut EPDM sheet abutting the diaphragm

H<sub>2</sub> Cell Frame: 1.605mm thick injection molded PEEK

H<sub>2</sub> Gasket Seal: 50µm thick die cut EPDM sheets (alternatively, could insertion mold gaskets on both sides of BPP at high volume)

Ni-coated SS304L BPP: 0.0762mm thick SS304L with 10µm PVD Ni coating on both sides



### Water and O<sub>2</sub> Evolution

O<sub>2</sub> PTL anode: 1.6mm thick sintered porous Ni (95% porosity)

CCM: 220µm thick Zirfon (ZrO<sub>2</sub>/polysulfone) diaphragm membrane with decal-transfer

- anode (NiFe(OH)<sub>2</sub> catalyst: 7 mg/cm<sup>2</sup>)

- cathode (NiMo catalyst: 7 mg/cm<sup>2</sup>)

H<sub>2</sub> PTL cathode: 1.6mm thick sintered porous Ni (95% porosity)

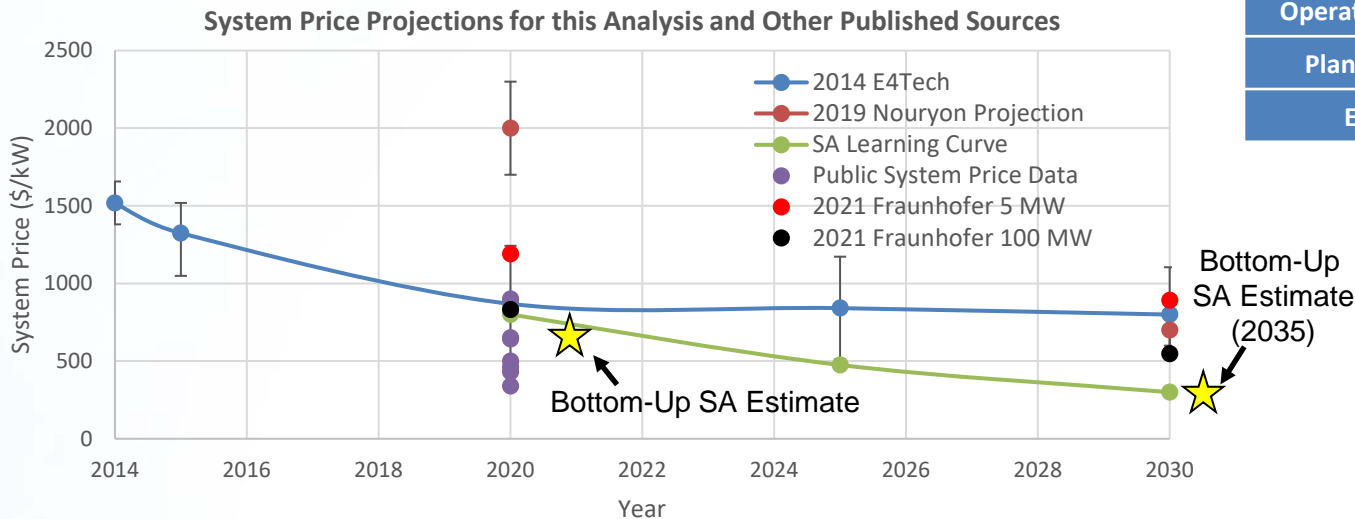
### H<sub>2</sub> Evolution

# Preliminary Liquid Alkaline Water Electrolysis H2A Modeling Assumptions/Results

## Current and Future Central LAWE H2A Cases

- SA DFMA-based System cost is ~\$610/kW (Current) and ~\$340/kW (Future).
  - This is in range of recent public system prices and towards low-end of public future projections.
  - Cost in \$/kW<sub>stack-input</sub> excl. installation
- H2A-based H2 cost projections are \$2.41/kgH<sub>2</sub> (Current) and \$1.79/kgH<sub>2</sub> (Future) based on:
  - \$0.03/kWh electricity)
  - Large Central (50 TPD) plants

	Unit	Current Central	Future Central
Avg. System Electrical Usage	kWh/kg H <sub>2</sub>	53.7	44.9
Avg. Stack Electrical Usage	kWh/kg H <sub>2</sub>	48.9	42.5
Thermal Energy Usage	kWh/kg H <sub>2</sub>	0.0	0.0
BoP Electrical Usage	kWh/kg H <sub>2</sub>	4.8	2.4
<b>Total System Price</b>			
Total System Price	\$/kW <sub>stack_input</sub>	\$610	\$344
Stack Price*	\$/kW <sub>stack_input</sub>	\$353	\$148
Total BoP Price*	\$/kW <sub>stack_input</sub>	\$257	\$195
Mechanical BoP	\$/kW <sub>stack_input</sub>	\$158	\$97
Electrical BoP	\$/kW <sub>stack_input</sub>	\$99	\$99
<b>Operating Capacity Factor</b>			
Operating Capacity Factor	(%)	97%	97%
<b>Plant Design Capacity</b>			
Plant Design Capacity	kg of H <sub>2</sub> /day	56,500	59,500
<b>Electricity Cost</b>			
Electricity Cost	(\$2016)/kWh	\$0.03	\$0.03



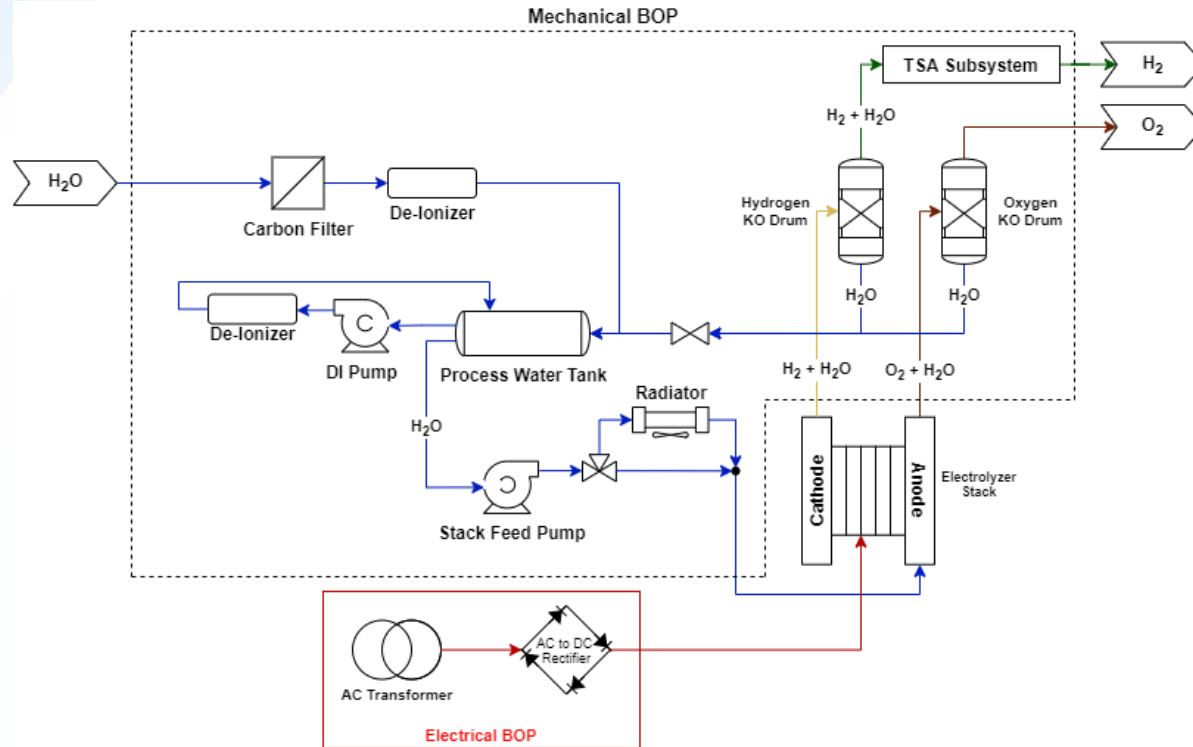
	Preliminary	Current Central	Future Central
<b>Cost Component</b>	<b>Cost Contribution (\$/kg)</b>	<b>Cost Contribution (\$/kg)</b>	<b>Cost Contribution (\$/kg)</b>
Capital Costs		\$0.44	\$0.22
Fixed O&M		\$0.31	\$0.19
Other Variable Costs (including utilities)		\$1.65	\$1.38
<b>Total</b>		<b>\$2.41</b>	<b>\$1.79</b>

\*Includes manufacturer markup

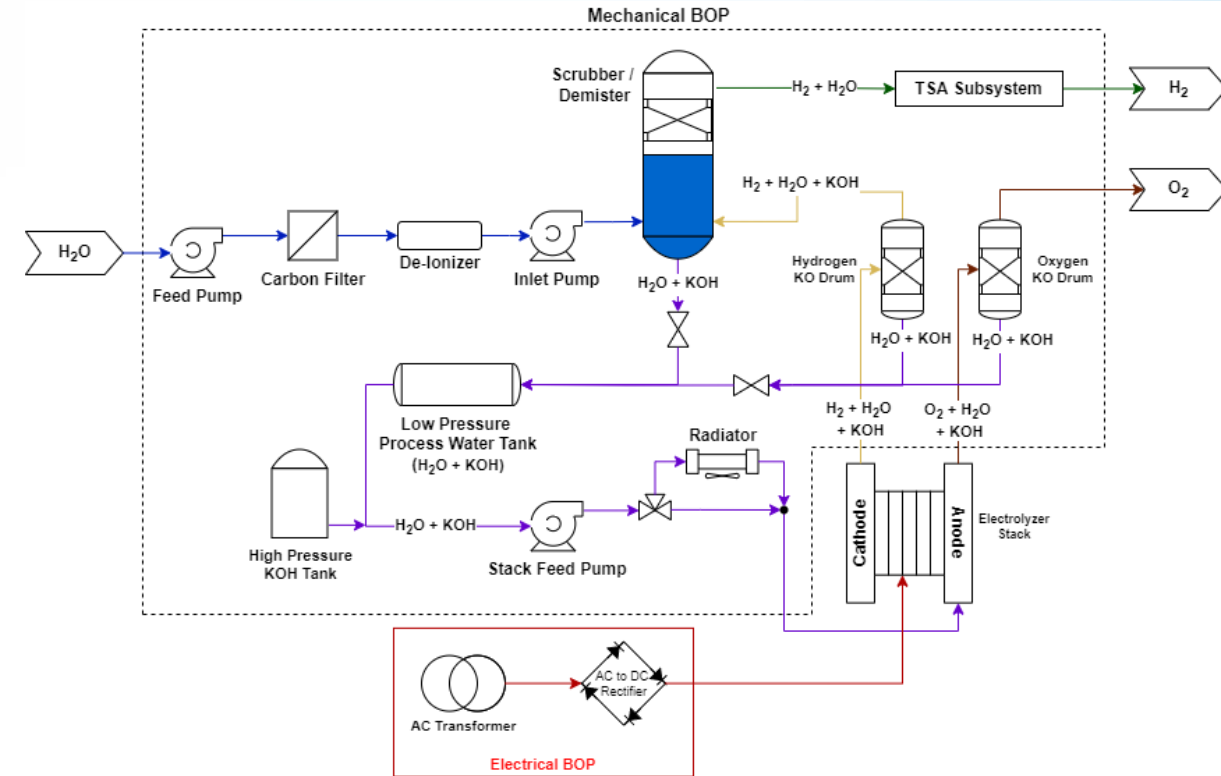
# Anion Exchange Membrane (AEM) Electrolysis Systems

## AEM Electrolysis Process Flow Diagrams

### Pure Water

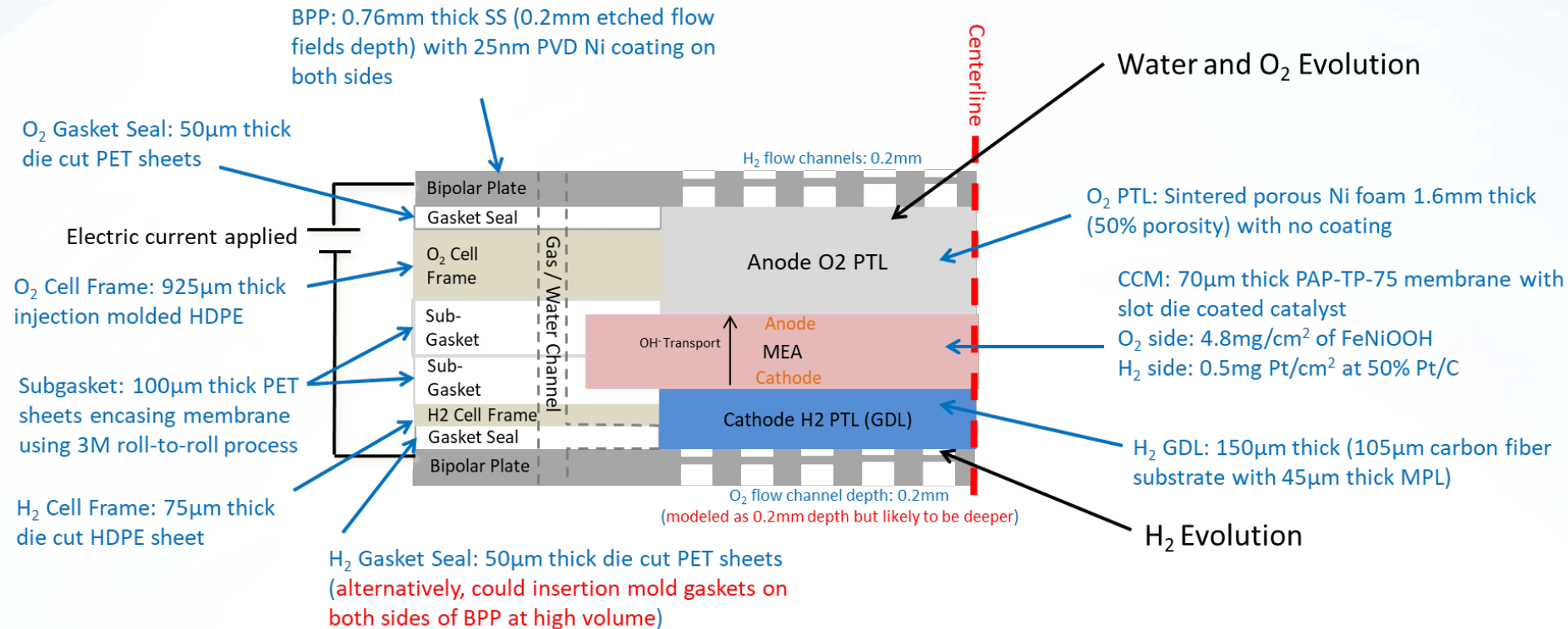


### 1M KOH



- Investigating both pure water and 1M KOH electrolyte system designs
- 1M KOH: Near-term demonstrated long life but complications of caustic KOH and low current density
- Pure Water: Active membrane research to eliminate KOH, improve durability, & deliver superior current density
- Both have potential for lower stack cost (than PEM or LA) due to inexpensive membrane, and low/no-PGM catalysts
- Both modeled at 20 bar operating pressure to eliminate need for additional mechanical compressor

# Preliminary AEM Stack DFMA Cost Analysis



## AEM Case Designs for Stacks

- Assumed no stack design changes between near term and future or between pure water and KOH
- Assumed improvement in stack performance between near term and current and pure water and KOH systems

## DFMA model capable of exploring wide design space

Stack DFMA Model Values	Unit	SA PEM Model (for Reference)	SA 2022 AEM Model
Range in Manufacturing Capacity	MW/year	10-10,000	300 (based on Enapter)-600
Range in Cell Area (Active/Total)	cm <sup>2</sup>	740-2981 / 1197-3900	740-1020 / 1196-1547
Stack Sizes	MW	1, 2, 2.5, 5	1, 1.5
System Sizes	MW	1, 4, 10, 100	3 (1.5 metric tons/day)

# Preliminary AEM Electrolysis System Definition

## AEM Key Technical and Cost Parameters (Distributed, 1,500kg H<sub>2</sub>/day)

Parameter	Unit	AEM Near Term 2025, Pure Water	AEM Near Term 2025, 1M KOH	AEM Future 2035, Pure Water	AEM Future 2035, 1M KOH
<b>Rated Operating Conditions</b>					
Rated Current Density	A/cm <sup>2</sup>	1	1.5	2	3
Rated Cell Voltage (BOL)	V	1.84	1.74	1.9	1.8
Stack Pressure	Bar	20	20	20	20
Stack Lifetime	years	1	3	7	10
<b>Optimized Operating Conditions</b>					
Current Density (BOL)	A/cm <sup>2</sup>	1.64	1.43	1.09	1.37
Cell Voltage (BOL)	V	2.1	1.72	1.68	1.60
EOL System Power	MW	4.8	3.5	2.95	2.80
<b>System Performance</b>					
Degradation Rate	mV/khrs	50	13	1.5	1
Stack Lifetime	years	1.1	3.4	7	10
Avg. Stack Electrical Usage	kWh/kg	63.7	48.8	44.3	42.0
Avg. System Electrical Usage	kWh/kg	66.1	50.8	45.3	42.9
<b>Capital Costs</b>					
Total System Price	\$/kW stack input	\$457	\$577	\$615	\$608
Stack Price	\$/kW stack input	\$87 (300MW/yr basis)	\$110 (300 MW/yr basis)	\$143 (600 MW/yr basis)	\$116 (600 MW/yr basis)
Total BoP Price	\$/kW stack input	\$371	\$467	\$472	\$493
Mechanical BoP	\$/kW stack input	\$272	\$368	\$373	\$394
Electrical BoP	\$/kW stack input	\$99	\$99	\$99	\$99

Stack costs are based on ground-up DFMA cost estimate. All dollar values are 2016\$.

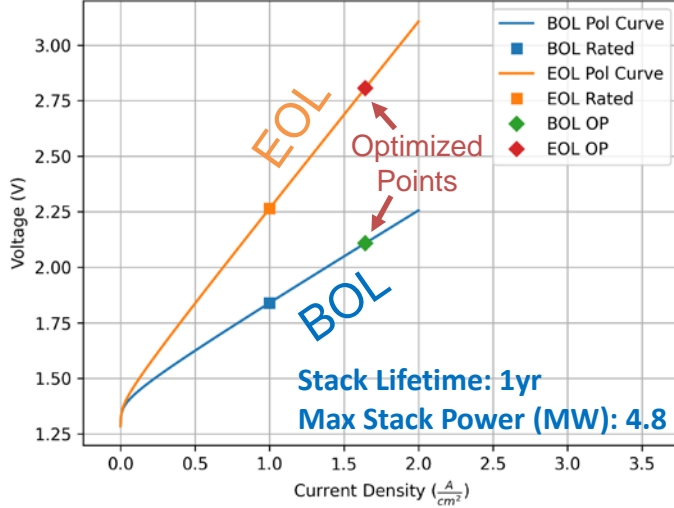
# Current & Future Polarization Curves Developed Considering Voltage Degradation

Pure Water

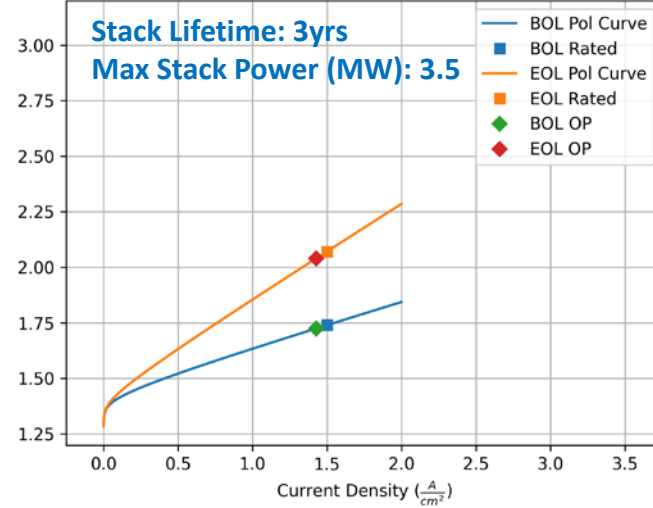
1M KOH

Near Term

Current Distributed AEM Water Case Pol Curves and Operating Points

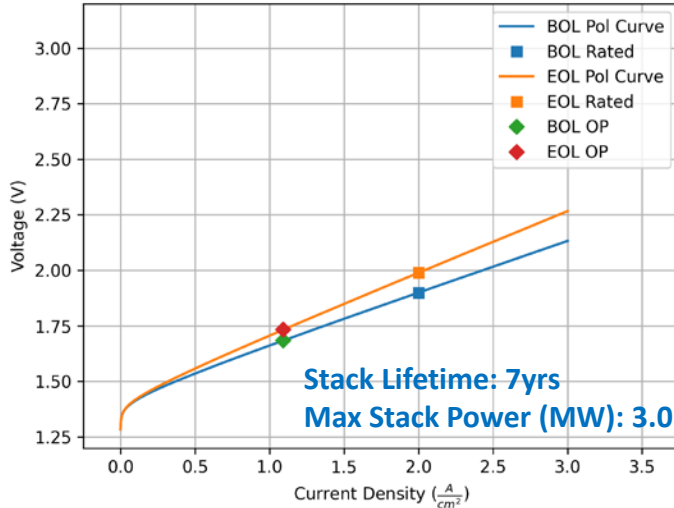


Current Distributed AEM Koh Case Pol Curves and Operating Points

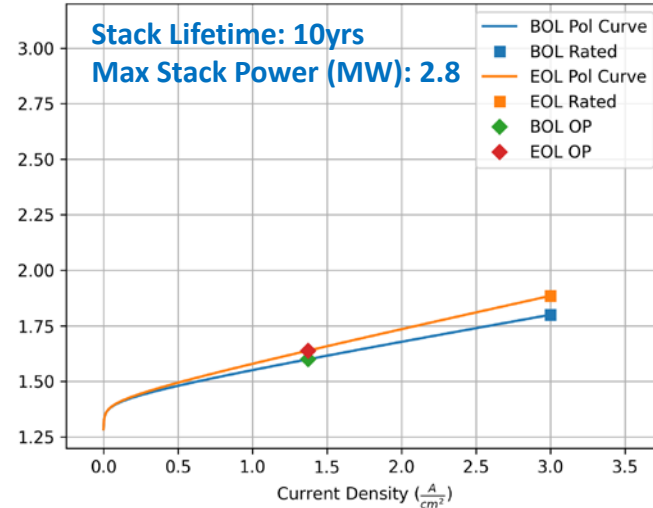


Future

Future Distributed AEM Water Case Pol Curves and Operating Points



Future Distributed AEM Koh Case Pol Curves and Operating Points



Pol. Curves used to determine Operating Point that leads to Lower H<sub>2</sub> Cost

Optimization Model Approach

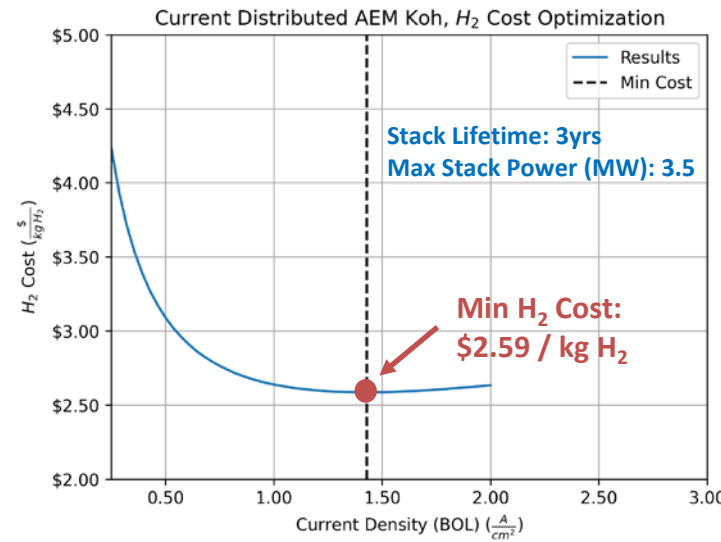
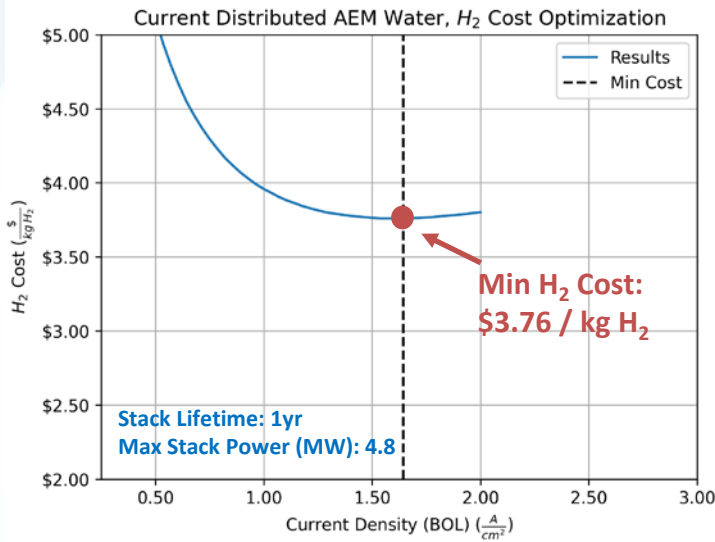
1. Specify BOL rated operating voltage, current density, and degradation (mv/1000hrs)
2. Determines EOL rated operating point
3. Estimates VI curve for BOL and EOL
4. Runs through each operating point along the curve to obtain cost of H<sub>2</sub>
5. Returns the operating point with the lowest H<sub>2</sub> cost

# Cost Optimization of Stack Operating Point

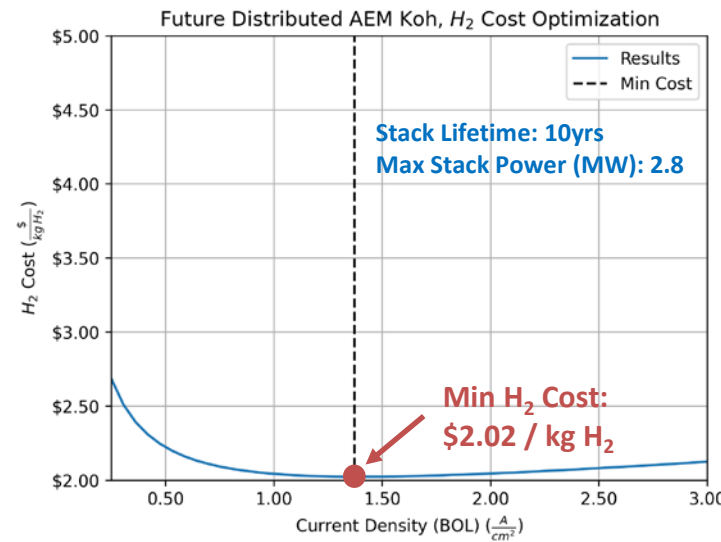
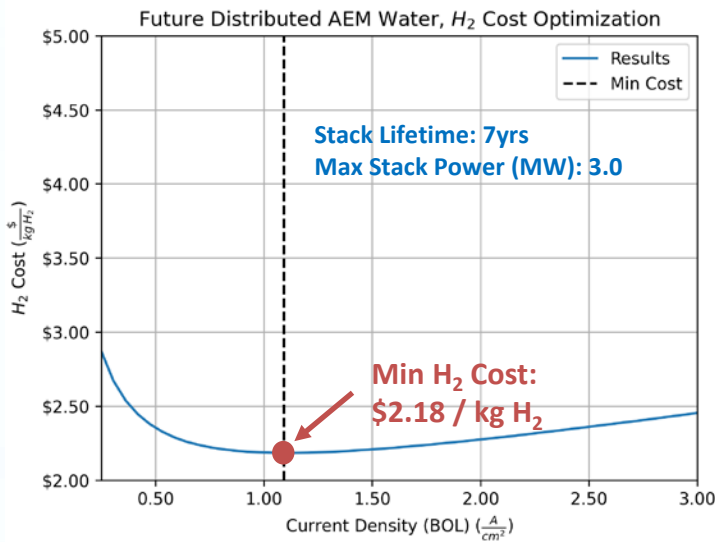
Pure Water

1M KOH

Near Term



Future



Pol. Curves used to determine Operating Point that leads to Lower H<sub>2</sub> Cost

## Optimization Model Approach

1. Specify BOL rated operating voltage, current density, and degradation (mv/1000hrs)
2. Determines EOL rated operating point
3. Estimates VI curve for BOL and EOL
4. Runs through each operating point along the curve to obtain cost of H<sub>2</sub>
5. Returns the operating point with the lowest H<sub>2</sub> cost

# Responses to Previous Year Reviewers' Comments

**This is the first year of this project and thus was not reviewed during the 2021 Annual Merit Review Meeting**



# Collaborators

Institution	Relationship	Activities and Contributions
<b>National Renewable Energy Laboratory (NREL)</b> <ul style="list-style-type: none"> <li>Genevieve Saur</li> </ul>	Subcontractor	<ul style="list-style-type: none"> <li>Participated in weekly project calls</li> <li>Assisted with H2A Production Model runs &amp; sensitivity analyses</li> <li>Drafted and reviewed reporting materials</li> <li>Managed and arranged H2A Working Group activities</li> </ul>
<b>Idaho National Laboratory (INL)</b> <ul style="list-style-type: none"> <li>Daniel Wendt</li> </ul>	Subcontractor	<ul style="list-style-type: none"> <li>Participated in select project calls</li> <li>Expert in Solid Oxide Electrolysis (which is not a current analysis focus in this year of the project)</li> </ul>
<b>Department of Energy (DOE)</b> <ul style="list-style-type: none"> <li>James Vickers (primary)</li> <li>Ned Stetson</li> <li>Dave Peterson</li> </ul>	Sponsor	<ul style="list-style-type: none"> <li>Participated in biweekly project calls</li> <li>Assisted with H2A Model and sensitivity parameters</li> <li>Reviewed reporting materials</li> </ul>
<b>Companies:</b> <ul style="list-style-type: none"> <li>Nel</li> <li>Versogen</li> </ul>	Reviewer	<ul style="list-style-type: none"> <li>Nel provided feedback on Liquid Alkaline Water Electrolysis design</li> <li>Versogen provided feedback on Anion Exchange Membrane design</li> </ul>

# Conclusions, Remaining Challenges and Barriers

## Liquid Alkaline Water Electrolysis Systems

- LAWE systems can have a wide range of cell/stack designs and the chosen single baseline design modeled may not fully represent any one system (i.e., finite gap and zero gap design approaches both exist in current systems)
- Design iterations have been somewhat stagnant for the last several decades with few present-day scientific research efforts addressing the durability of alternative materials and designs
- A DFMA analysis confirms the current low cost of LAWE systems (~\$600/kW) and shows a pathway to future cost of \$300/kW.
- The low cost of LAWE systems coupled with relatively high efficiency lend itself to a low near-term cost of hydrogen. While PEM, AEM, and SOE have the potential to beat LAWE on both capital cost and efficiency, LAWE remains a competitive option for near-term and long-term electrolyzer deployment

## Anion Exchange Membrane Electrolysis Systems

- AEM systems are promising for their potential for non-PGM catalysts, low membrane cost, and use of stainless components (i.e., Titanium or Nickel plates/plating are not required)
- Although pure water systems can have a simplified BOP system without a KOH scrubber, there are multiple advantages for operating with a supporting-electrolyte system (such as KOH) that include:
  - » improved durability over pure water systems
  - » improved current density over pure water systems
- TEA analysis shows that due to lower \$/cm<sup>2</sup> stack costs, AEM can be operated at lower cell voltages (than PEM) to achieve higher efficiency
- Durability and performance remain significant issues, even with KOH electrolyte added
  - Although the understanding of degradation mechanisms are slowly being uncovered, AEM systems in a way are playing “catch-up” to PEM systems (which has had more intensive R&D in the last 20 years)
  - If AEM durability & performance can rise to the level of PEM systems, AEM systems may be quite competitive on a \$/kgH<sub>2</sub> cost basis

# Proposed Future Work

- **Complete LAWE H2A Cases**
  - System Cost analysis
    - Finalize stack DFMA cost analysis
    - Re-evaluate BOP component costs (to ensure consistency with PEM and AEM cost assumptions)
    - Conduct sensitivity analysis
    - Vet cost results and sensitivity analysis with NREL and Nel collaborators
  - Publish H2A Results in a Case Study and DOE Record
- **Complete AEM H2A Cases**
  - System Cost analysis
    - Finalize BOP cost components
    - Conduct sensitivity analysis
    - Vet cost results and sensitivity analysis with NREL and Versogen
  - Publish H2A Results in Case Study DOE Record
- **\$1/kg H<sub>2</sub> Shot Scoping Study**
  - Investigate the ability of electrolysis to achieve the target by:
    - First estimating the cost with the lowest cost parameter values possible
    - Conducting a Monte Carlo multi-variable sensitivity analysis to determine successful parameter combinations

Any proposed future work is subject to change based on funding levels.

# Summary

- **Overview**
  - Conducted technoeconomic analyses for LAWE and AEM Electrolyzer hydrogen production technologies
- **Relevance**
  - Improve analysis models and increase understanding of areas demonstrating information deficiencies
  - Technoeconomic analysis for H<sub>2</sub> Production:
    - Defines a complete production and delivery pathway
    - Identifies key cost-drivers and helps focus research on topics that will lower cost
    - Generates transparent documentation available to the community with relevant data for improved collaboration
- **Approach**
  - Utilize various cost analysis methods for determining system cost: DFMA<sup>®</sup> and H2A
  - Collaborate with NREL, ANL, DOE, and tech experts to model SOA and future systems
  - Vet assumptions and results for correctness, completeness, and maximum transparency
- **Accomplishments**
  - (Planned) public distribution of H2A Model and Case Study Updates
  - Analysis of two H<sub>2</sub> electrolysis production systems (LAWE and AEM)

# Technical Backup and Additional Information Slides

# Technology Transfer Activities

Technology transfer does not apply to this analysis-type project

# Publications and Presentations

1. James, B., “Liquid Alkaline Electrolysis Techno-Economic Review,” Presentation at the US DOE Experts Meeting on Advanced Liquid Alkaline Water Electrolysis, January 2022.

# Photoelectrochemical (PEC) Water Splitting

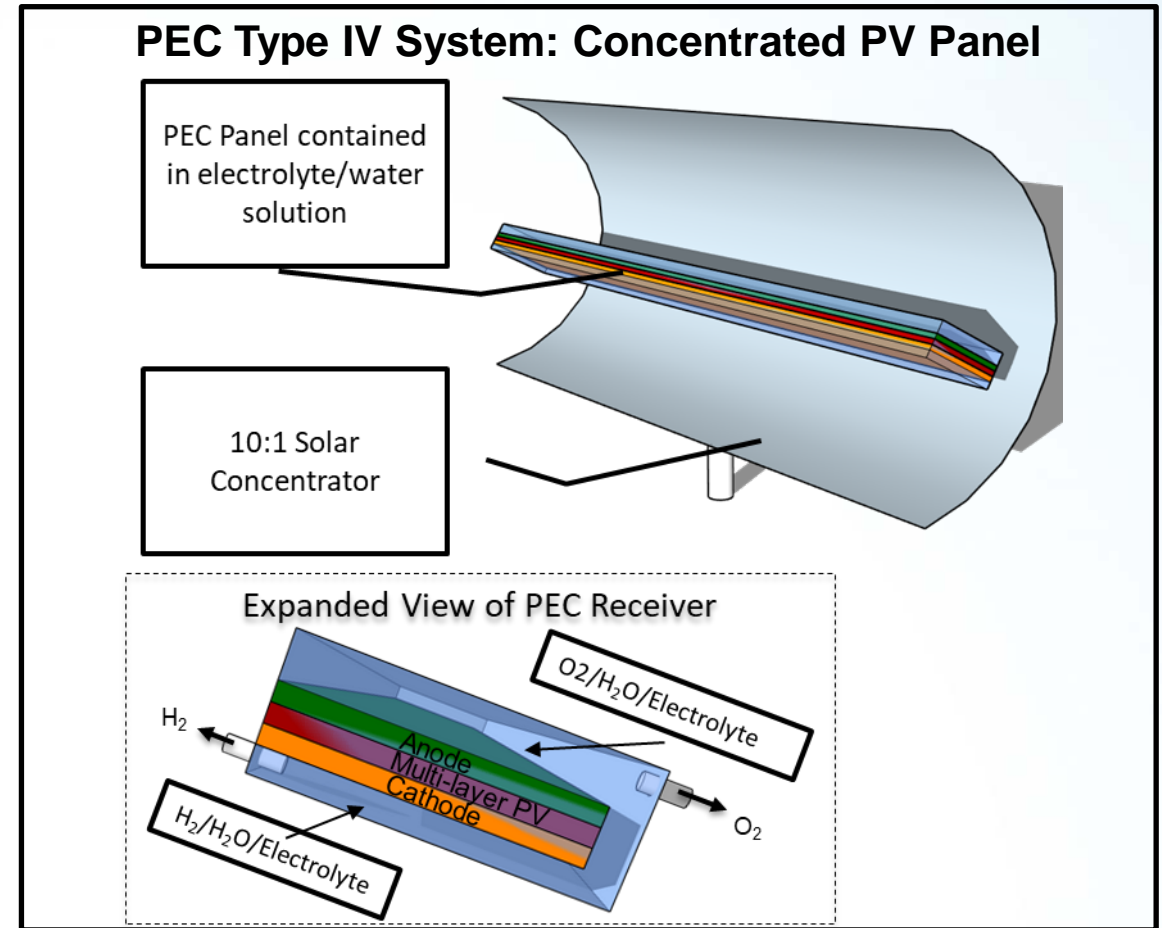
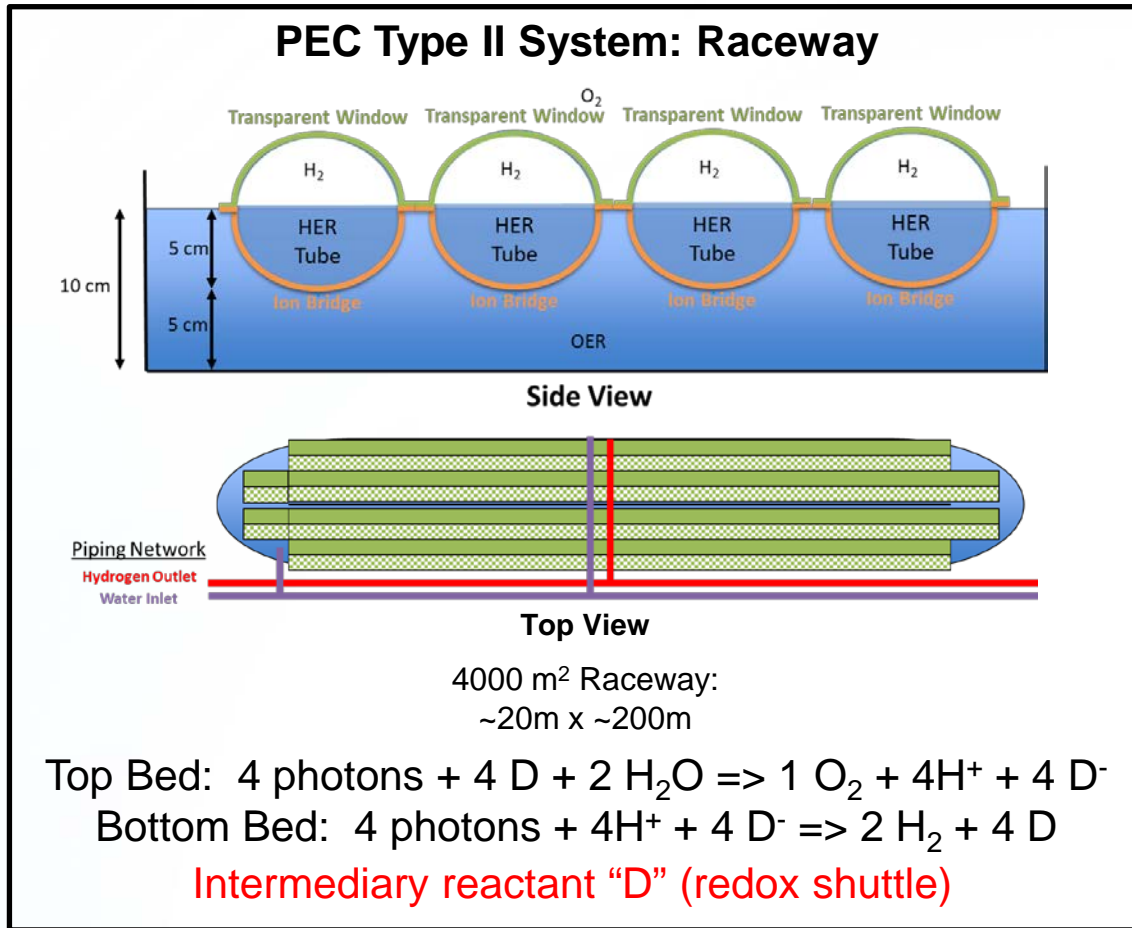
The following slides represent work conducted prior to the start of this project. Although this is a separate contract, the work conducted was on H<sub>2</sub> production pathway techno-economic analysis and these latest results have not yet been briefed to the public.

Collaboration with Shane Ardo (UC Irvine) and Rohini Bala Chandran (U of Michigan) on PEC catalysts and PEC type 2 raceway concepts



# Photoelectrochemical Water Splitting

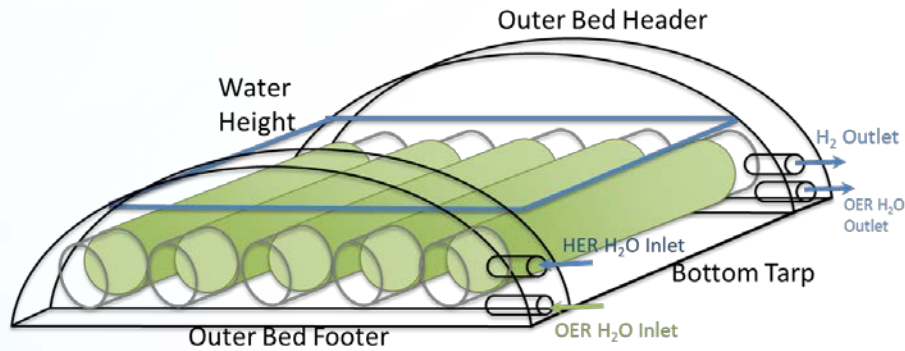
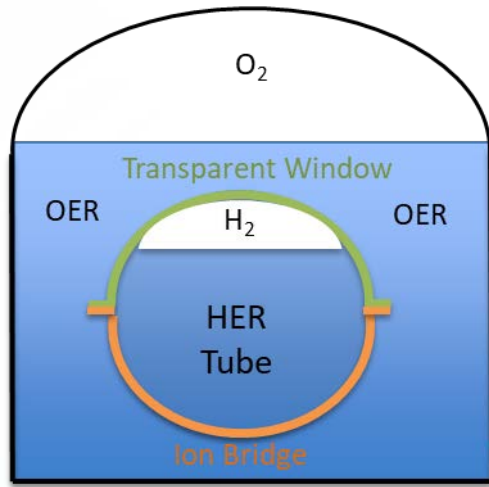
- Four Types of PEC Considered: Two selected for investigation by DOE
  - **Type II: Raceway System** - Nanoparticle catalysts contained in a separate aqueous systems
    - HER and OER reactions occur in separate aqueous systems connected via ion bridges
  - **Type IV: Concentrated PV Panel** – A PEC receiver contained in a water/electrolyte with concentrating solar panels



# Progression of PEC Type 2 Designs

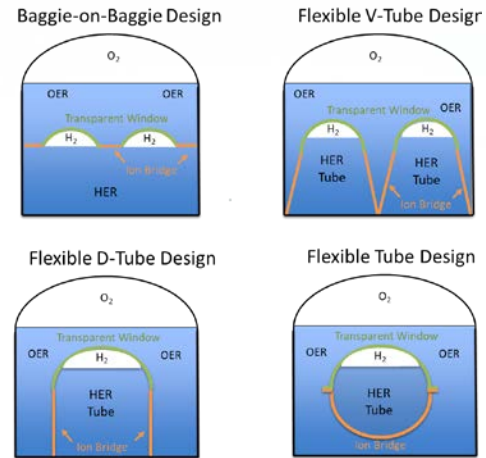
**Baggie within Baggie**

**A**

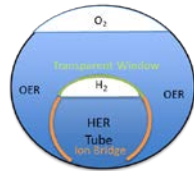


**Alternative Baggie Geometries**

**B**



Tube in Bottom Tube



Concentric Tube

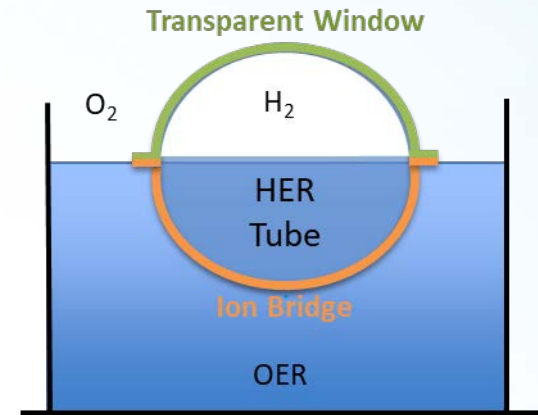


Double Tube in Tube

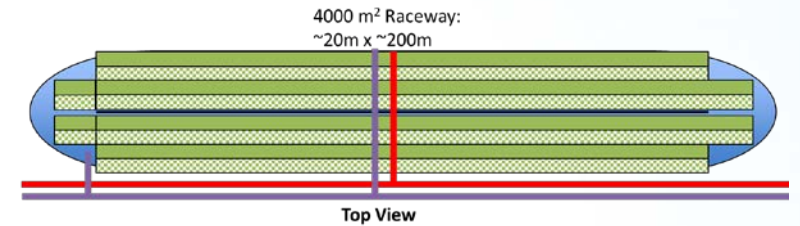


**Raceway**

**C**



**Side View**

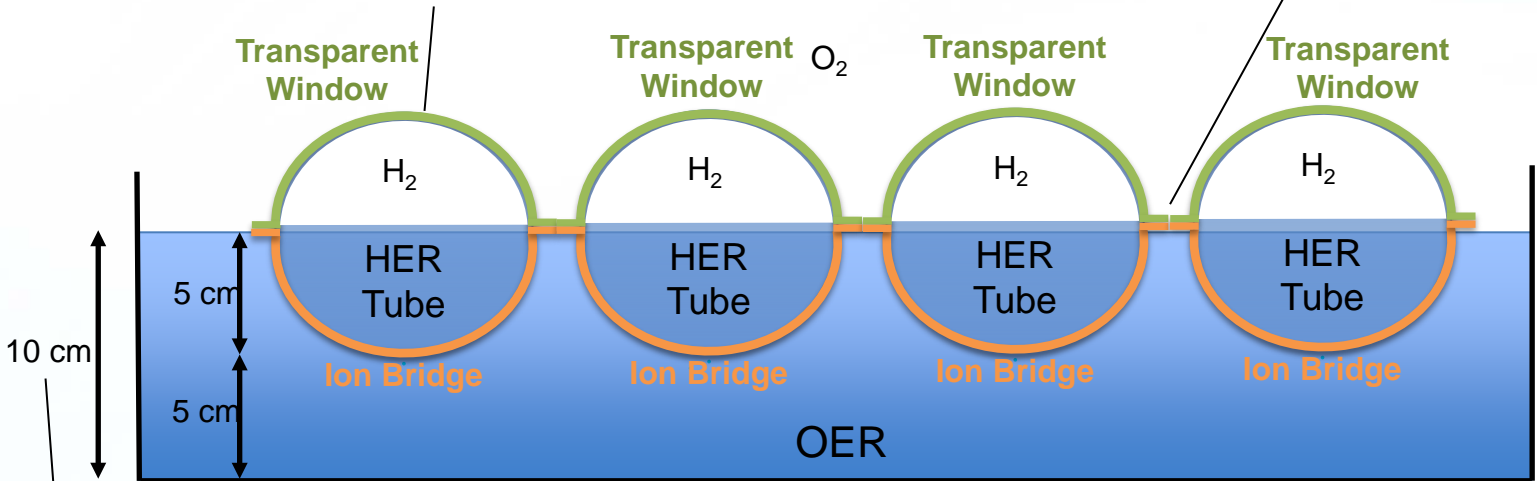


**Top View**

# PEC Type 2: Flexible Tube Design *at Scale*

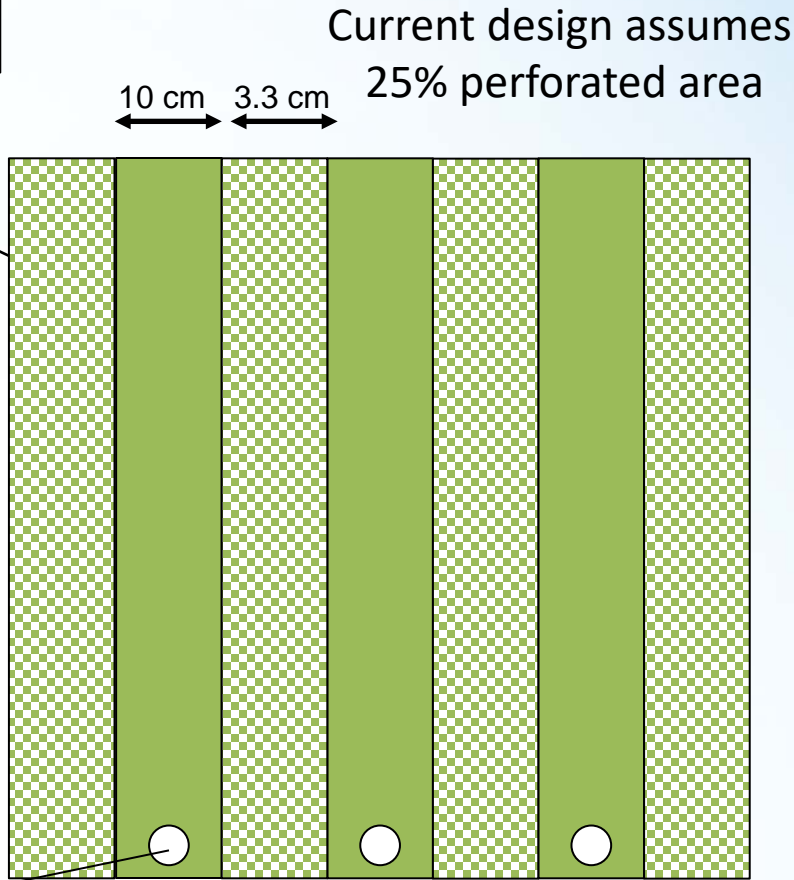
Sheet is perforated between tubes to allow oxygen to rise

Continuous window and ion bridge sheet is relatively easy to manufacture



Side View

Height of OER fluid could be 1-10 cm

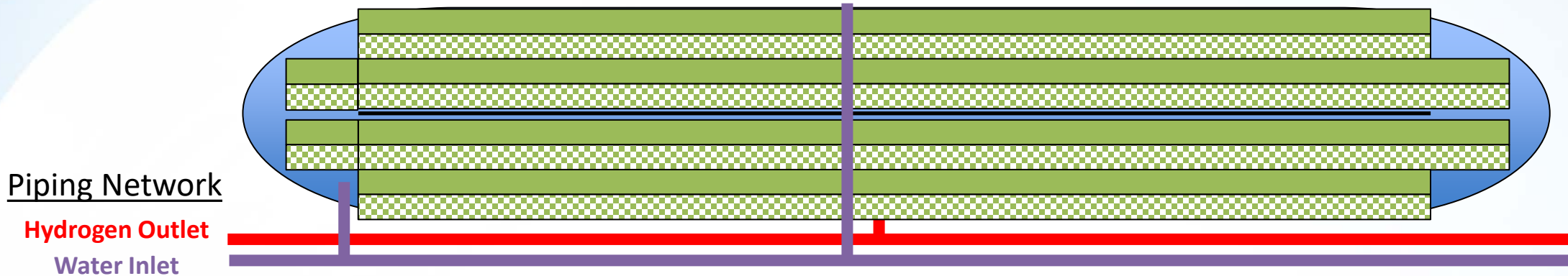


Top View

Port hole to collect H2 in central receiver

All dimensions are estimates only

# PEC Type 2: Proposed “Open” Design



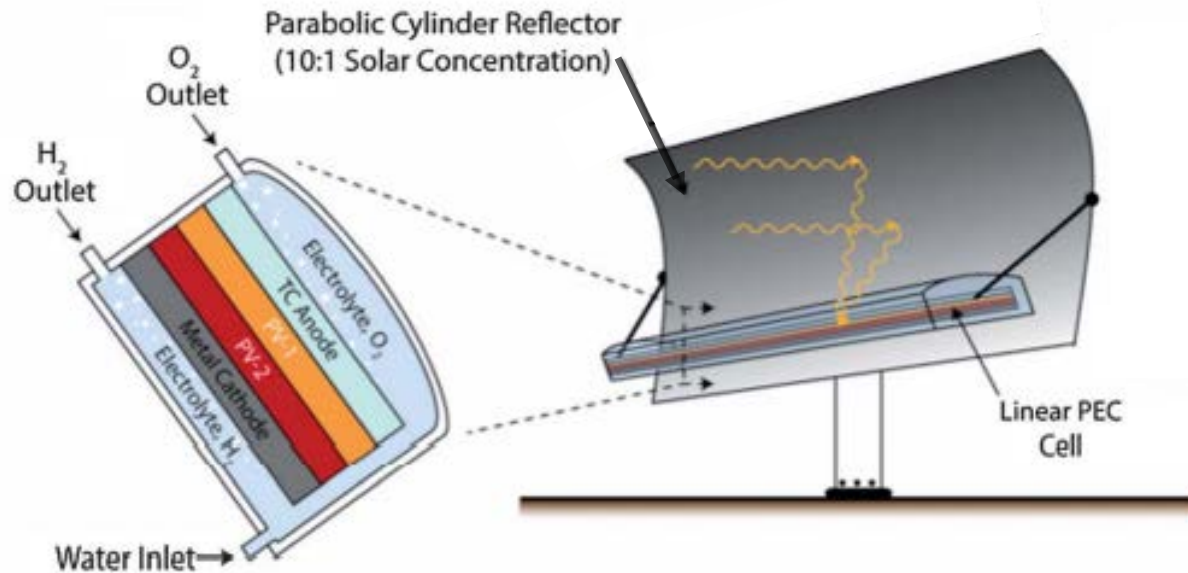
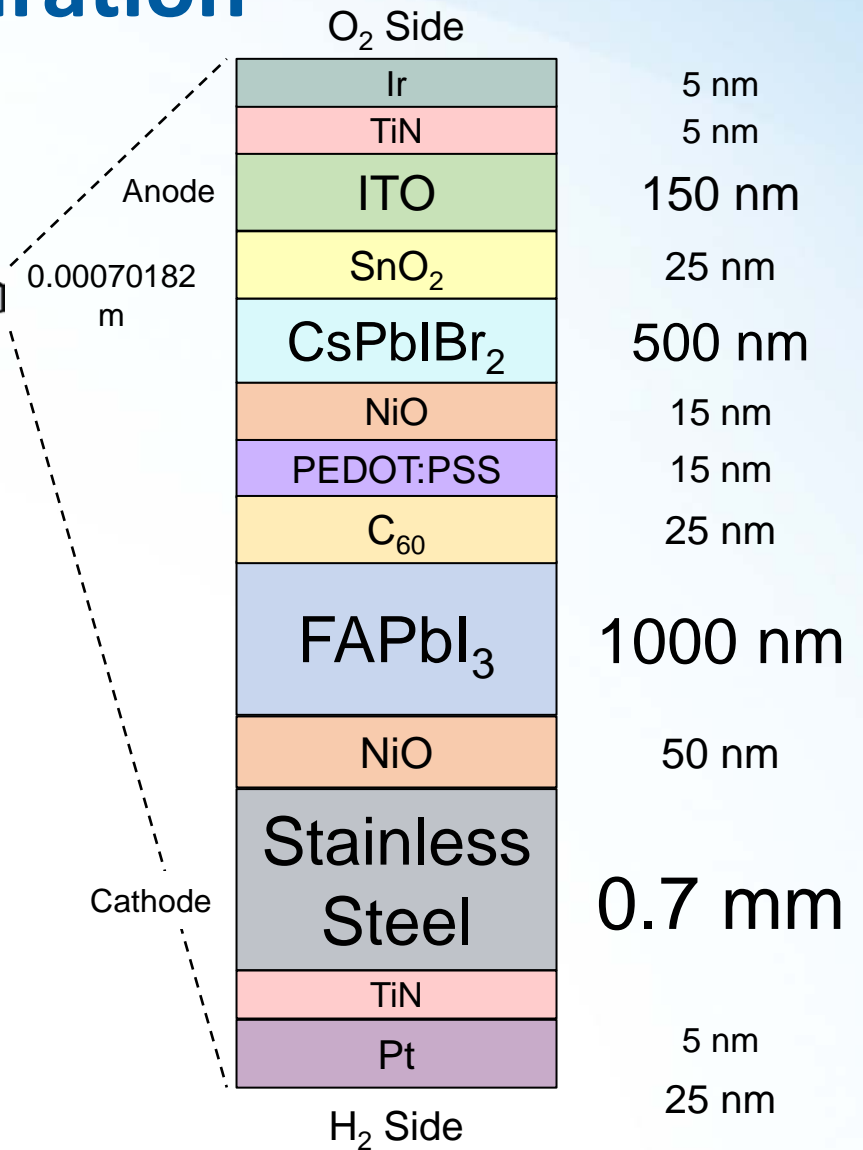
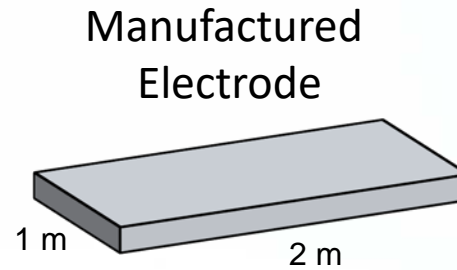
**Top View**  
4000 m<sup>2</sup> Raceway:  
~20m x ~200m

## Design Notes:

- Conventional algae raceway uses a motor and paddle. However, due to shallow pool, no motor and paddle are assumed for PEC raceway
- ~1400 horizontal cylinders/raceway (10 cm diameter per cylinder)
- In addition to water conversion to hydrogen, water will be carried by both the hydrogen and oxygen outlets
  - Will need replacement water for OER pond and HER cylinders
- Port installation assumed for only cylinder inlets and outlets. Raceway inlets only use perforations

# PEC Type 4: Panel Configuration

- Type IV Tracking Concentrator Array
  - Each concentrator array paired with a PEC panel encapsulating an electrode
  - Conducted DFMA model of panel manufacturing cost



Pinaud, Blaise A., et al. "Technical and economic feasibility of centralized facilities for solar hydrogen production via photocatalysis and photoelectrochemistry." *Energy & Environmental Science* 6.7 (2013): 1983-2002.

Construction and material selection is a modification of:  
[https://www.hydrogen.energy.gov/pdfs/review21/p191\\_yan\\_2021\\_o.pdf](https://www.hydrogen.energy.gov/pdfs/review21/p191_yan_2021_o.pdf)

# PEC Type 4: DFMA Panel Cost Results

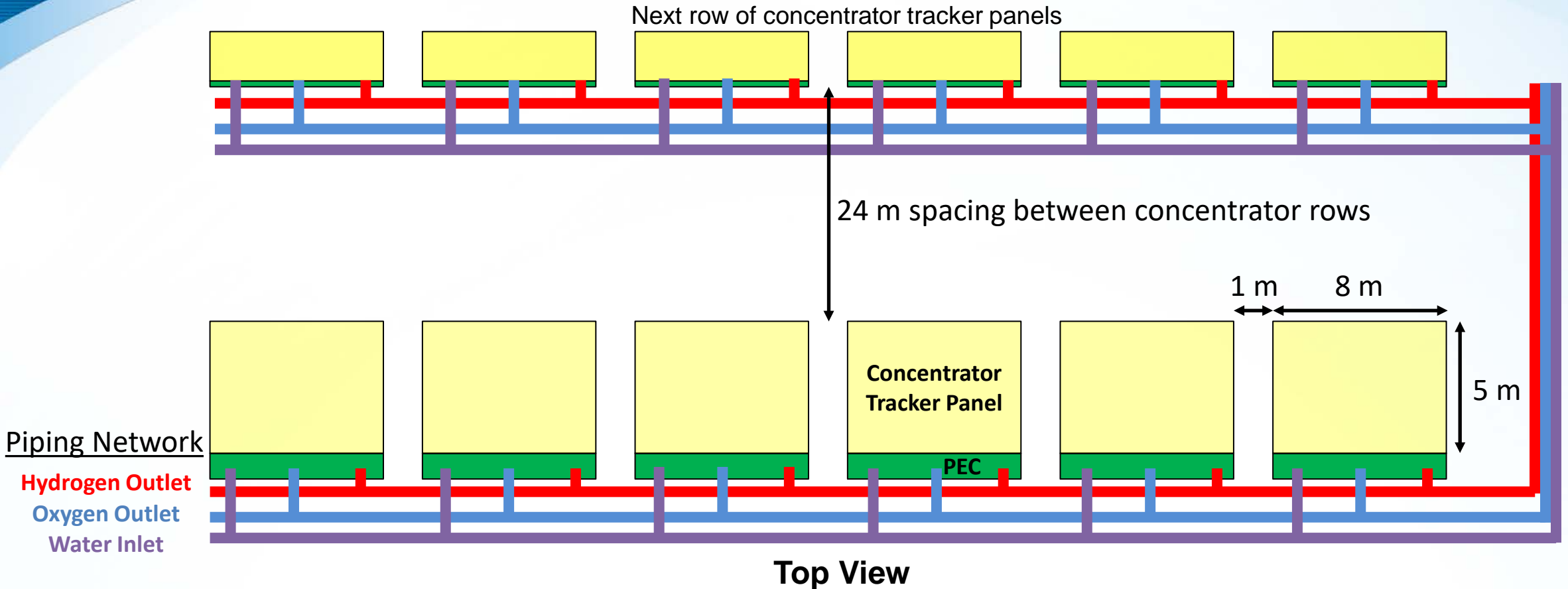
Annual Production Rate						
Area Per Year (m <sup>2</sup> /year)	1,000	10,000	50,000	100,000	1,000,000	10,000,000
Panels Per Year (panels/year)	500	5,000	25,000	50,000	500,000	5,000,000

PEC Panel Results						
PEC Panel Cost Per Area (\$/m <sup>2</sup> )	\$4,980	\$612	\$223	\$175	\$163	\$160
PEC Panel Material Cost Per Area (\$/m <sup>2</sup> )	\$118	\$118	\$118	\$118	\$118	\$118
PEC Panel Manufacturing Cost Per Area (\$/m <sup>2</sup> )	\$4,862	\$494	\$106	\$57	\$45	\$43
PEC Panel Cost (\$/panel)	\$9,960	\$1,223	\$447	\$350	\$325	\$321

## Design Notes:

- 10,000 m<sup>2</sup>/year manufacturing rate chosen for current cases (~\$615/m<sup>2</sup>)
- 10,000,000 m<sup>2</sup>/year manufacturing rate chosen for current cases (~\$160/m<sup>2</sup>)
- 50% manufacturing markup included
- Casing and plexiglass cover included as separate capital item

# PEC Type 4: Proposed Layout



## Design Notes:

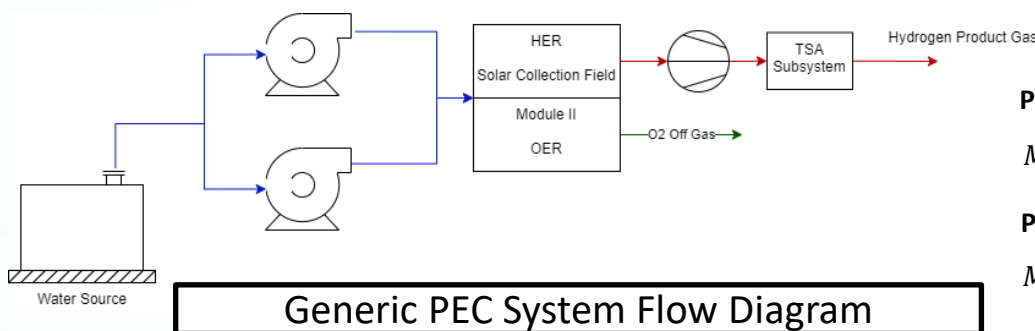
- Large spacing required between rows to avoid shadowing effect
- Spacing between panel rows derived in 2009 report and used here (24m spacing per 5m of concentrator height)
- Spacing between concentrator trackers required to allow movement of the panel to track the sun

# Preliminary Technical Specifications

- PEC Operation is water splitting with direct solar energy
  - Solar insolation rates are used to calculate the amount of active material needed
  - Separated Hydrogen Evolution Reaction (HER) and Oxygen Evolution Reaction (OER)
    - Separate Hydrogen and Oxygen beds in PEC Type II systems
    - Hydrogen and Oxygen are naturally separated by the shape and angle of the electrode in PEC Type IV System
- PEC typically has solar-to-hydrogen energy conversions below 20%
- A modular PEC design is envisioned in this analysis
  - Each module has a capacity of 1,000 kgH<sub>2</sub>/day
  - Multiple modules strung together to reach desired H<sub>2</sub> production
- Preliminary system specs shown in tables to right

PEC Type II System Technical Specifications		
	Units	Value
PEC Type	-	Type II
Average Insolation	kWh m <sup>-2</sup> day <sup>-1</sup>	5.77
STH Efficiency	%	8%
Average H <sub>2</sub> Mass Flow	kg day <sup>-1</sup>	1,000
Area Specific Mass Flow	Kg H <sub>2</sub> hr <sup>-1</sup> m <sup>-2</sup>	5.89E-04
Total Solar Area Required	m <sup>2</sup>	70,790
Raceway Length	m	200
Raceway Width	m	20
Raceway Height	m	0.01
Raceway Area	m <sup>2</sup>	3800
Floating Cylinder Width	M	0.01
Number of Floating Cylinders per Raceway	#	1425
Number of Raceways	#	19
Assumed Particle Density	kg m <sup>-2</sup>	0.00105
Particle Mass	kg	74

PEC Type IV System Technical Specifications		
	Units	Value
PEC Type	-	Type IV
Average Insolation	kWh m <sup>-2</sup> day <sup>-1</sup>	7.46
STH Efficiency	%	35%
Overall Solar Efficiency	%	33.3%
Collector Efficiency	%	95%
Average H <sub>2</sub> Mass Flow	kg day <sup>-1</sup>	1,000
Area Specific Mass Flow	kg H <sub>2</sub> hr <sup>-1</sup> m <sup>-2</sup>	3.33E-03
Total Area Collector Required	m <sup>2</sup>	13,200
Collector Length	m	8
Collector Width	m	5
Collector Area	m <sup>2</sup>	40
Number of Collectors	#	330
PV Area Required	m <sup>2</sup>	440



**PEC Type 2 Raceway Estimate:**

$$Module\ FTE = \left( \frac{0.0042\ FTE}{Particle\ lifetime\ (yrs)} + 0.018\ FTE \right) \frac{(Solar\ Land\ Required)}{4000m^2} + \frac{13\ FTE}{50\ TPD} (Module\ H_2\ Production)$$

**PEC Type 4 Estimate:**

$$Module\ FTE = \frac{0.016\ FTE}{1000\ m^2} (Total\ Land\ Required) + \frac{3\ FTE}{50\ TPD} (Total\ Facility\ H_2\ Production)$$



# PEC Sensitivity Analysis

## PEC Type II

- Case study suggests that the STH efficiency must be  $>6.5\%$  with a particle lifetime of  $>1$  years to achieve a target hydrogen price of  $\$2/\text{kg}$

## PEC Type IV

- Case study suggests that achieving a cost target of  $\$2/\text{kg H}_2$  will require an STH efficiency  $>25\%$  with a catalyst lifetime of  $>1$  years, and a concentration ratio of 30
- Concentration ratio of 30 incurs a system temperature of  $\sim 185^\circ\text{C}$ , requiring a system pressure of 300 psi

